

# Evaluation of oxygen isotopes and trace elements in planktonic foraminifera from the Mediterranean Sea as recorders of seawater oxygen isotopes and salinity

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**Abstract.** The Mediterranean Sea is characterized by a relatively strong west to east salinity gradient, which makes it an area suitable to test the effect of salinity on foraminiferal shell geochemistry. We collected living specimens of the planktonic foraminifer *Globigerinoides ruber albus* to analyse the relation between element/Ca ratios, stable oxygen isotopes of their shells and surface seawater salinity, isotopic composition and temperature. The oxygen isotopes of sea surface water correlate with salinity in the Mediterranean also during winter, when sampled for this study. Sea water oxygen and hydrogen isotopes are positively correlated in both the eastern and western Mediterranean Sea, though especially in the eastern part the relationship differs from values reported previously for that area. The slope between salinity and seawater oxygen isotopes is lower than previously published. Still, despite the rather modest slope, seawater and foraminiferal carbonate oxygen isotopes are correlated in our dataset although with large residuals and high residual variability. This scatter can be due to either biological variability in vital effects or environmental variability. Numerical models backtracking particles show ocean current driven mixing of particles of different origin might dampen sensitivity and could result in an offset caused by horizontal transport. Results show that Na/Ca is positively correlated to salinity and independent of temperature. Foraminiferal Mg/Ca increases with temperature, as expected, and in line with earlier calibrations, also in the high salinity environment. By using living foraminifera during winter, the previously established Mg/Ca-temperature calibration is extended to temperatures below 18 °C, which is a fundamental prerequisite of using single foraminifera for reconstructing past seasonality.

## 1 Introduction

Ocean circulation plays an important role in Earth's climate, by redistributing heat and also by impacting global biogeochemical cycles. Seawater temperature and salinity are key parameters for reconstructing ocean circulation, since together they determine seawater density and thereby large scale circulation patterns, including a substantial part of meridional overturning circulation. Reconstruction of past ocean environments largely relies on so-called proxy calibrations in which a

variable which can be measured in the geological record is related to a target environmental parameter. The incorporation of trace metals in foraminiferal shell carbonate, for example, is a popular tool to reconstruct past ocean parameters. More specifically, the incorporation of Mg (often expressed as the calcite's Mg/Ca) increases exponentially with seawater temperature, as first observed in culture studies (Nürnberg et al., 1996) and later confirmed by field calibrations (Anand et al., 2003).

In addition to temperature, salinity and inorganic carbon chemistry also affect Mg/Ca in some species of foraminifera (Allison et al., 2011; Dueñas-Bohórquez et al., 2011; Geerken et al., 2018; Gray et al., 2018; Hönisch et al., 2013; Kisakürek et al., 2008; Lea et al., 1999). For the best possible accuracy such effects need to be corrected for when using foraminiferal Mg/Ca for the reconstruction of temperature, which calls for independent proxies for these other environmental parameters.

Currently, salinity is often reconstructed through indirect relationships with other variables, such as the ratio of stable oxygen isotopes of sea water, which are recorded in planktonic foraminifera (Rohling, 2007), although direct approaches have also been suggested recently (Wit et al., 2013; Bertlich et al., 2018). Since seawater oxygen isotope ratio and salinity are both affected by evaporation and precipitation, the two often are linearly related (Rohling, 2007; Bahr et al., 2013), with their calibration depending on local conditions. If foraminifera precipitate their calcite in equilibrium with respect to sea water oxygen isotopes, their  $\delta^{18}\text{O}$  should reflect that of the seawater, and hence salinity. However, as seawater temperature affects stable oxygen isotope fractionation during calcification (McCrea, 1950; Urey et al., 1951) independent temperature reconstructions are needed to estimate seawater  $\delta^{18}\text{O}$  from  $\delta^{18}\text{O}_{\text{calcite}}$  (Rohling, 2007). Independent temperature reconstructions can be based for example on organic proxies such as  $\text{U}^{\text{K}}_{37}$  (Prah and Wakeham, 1987), TEX86 (Schouten et al., 2006) or the Mg/Ca of the foraminifera themselves (Mashiotto et al., 1999; Elderfield and Ganssen, 2000). Accuracy and precision of such reconstructions is debated because propagation of errors from combined inaccuracies of the analyses and the uncertainties in calibrations due to combining several proxies, is difficult to assess and seems too large for meaningful reconstructions of changes in salinity over time (Rohling, 2007). Because of the lack of a suitable alternative approach, the use of Mg/Ca to determine the temperature effect of foraminiferal  $\delta^{18}\text{O}$  continues to be applied in settings which are prone to large changes in salinity such as the Mediterranean Sea. This calls for an independent in-situ calibration in which all the involved parameters are measured and not determined by proxy-relationships.

Culture experiments using the benthic, symbiont-barren *Ammonia tepida* (Wit et al., 2013) and the planktonic *Globigerinoides ruber* (pink) (Allen et al., 2016), showed that Na incorporation in foraminiferal shell carbonate is positively correlated with sea water salinity. A field calibration confirmed this positive correlation for the planktonic foraminiferal species *G. ruber albus* (Mezger et al., 2016), as well as for *Trilobatus sacculifer* (previously called *Globigerinoides sacculifer*) in the Red Sea and the Atlantic Ocean (Mezger et al., 2016; Bertlich et al., 2018). Comparison of Na/Ca-salinity calibrations shows, however, that absolute Na/Ca values and also sensitivities to salinity vary between species (Mezger et al., 2016).

When using field calibrations to constrain accuracy and precision of potential reconstruction approaches, it is important to also consider the potential impact of lateral transport of foraminifera due to (ocean) currents. Foraminifera collected at a specific sampling location might actually have added the majority of their shell carbonate at a different location and hence under

different environmental conditions as they have been transported to the sampling location. This may add to the uncertainty in the variable to cross-correlate against or even introduce a bias in the resulting calibration. Recently this has been suggested for dinoflagellate cysts (Nooteboom et al., 2019) and planktonic foraminifera, collected from the water column (Ganssen and Kroon, 1991), from sediment (van Sebille et al., 2015) and also from sediment traps (Steinhardt et al., 2014), but can also be applied to specimens collected living from the sea surface.

Here we used a plankton pump and sea water samples collected from the Mediterranean Sea in January and February of 2016 to test viability of deconvolving salinity from combined temperature and sea water oxygen isotope reconstructions. We also investigate the potential of the newly developed salinity proxy Na/Ca in the Mediterranean Sea. Using samples collected in winter we also extend the calibration of Mg/Ca to sea water temperature for *G. ruber albus* towards its lower temperature tolerance limits (14°C; Bijma et al., 1990), which is essential for the application of this species for past seasonality reconstructions.

## 2 Materials and Methods

During two cruises (NESSC Cruises 64PE406 and 64PE407, RV Pelagia) between January 12th and February 25th in 2016, a total of 98 plankton samples were collected from the surface waters of the Mediterranean Sea along an east-west transect using a plankton pump system (Ottens, 1992). Surface water was continuously pumped on board from 5m water depth and lead through a plankton net with 100 µm mesh size. Replacing the cod-end every 6h (filtering 57m<sup>3</sup> of sea water on average, constantly monitored using a water gauge), accumulated samples were washed out of the net into a 90 µm sieve, rinsed thoroughly with deionized water to remove smaller particles as well as salts, and subsequently stored onboard at -80°C. At NIOZ all plankton samples were then freeze-dried, and dry oxidation by low temperature ashing (100°C) was used to combust the organic material while minimizing potential impacts on carbonate trace metal concentrations and δ<sup>18</sup>O (Fallet et al., 2009). After ashing, samples were rinsed again thoroughly with de-ionized water and ethanol to remove potential ash residues. A variety of samples containing specimens of *G. ruber albus* (Morard et al., 2019) was selected to cover a large range in salinities and temperatures. Specimens used for analyses were selected from the size fraction 150 - 250 µm, even though it has been reported that at this size fraction *G. ruber albus* and *Globigerinoides elongatus* cannot always be confidently distinguished due to similar morphology (Aurahs et al., 2011). Surface seawater samples for stable oxygen isotopes were collected every 60 minutes from the same pump, resulting in a set of 309 samples. A volume of 2 ml was stored without headspace at 4°C during the cruise to be analyzed at the home laboratory.

The elemental ratio of the final foraminiferal chamber (named the F-chamber) of individual shells were measured by laser ablation quadrupole inductively coupled plasma mass spectrometry (LA-Q-ICP-MS) using a circular spot with a diameter of 60-80 µm, depending on the size of the last chamber. The laser system (NWR193UC, New Wave Research) at Royal NIOZ was used in combination with a two-volume sample cell (TV2), which allows detecting variability in elemental ratios within the foraminiferal chamber wall due to a short wash-out time of 1.8s (van Dijk et al., 2017). Ablating only F-chambers minimizes sampling of older carbonate that might have formed under different environmental conditions due to lateral and

vertical transport. All specimens were ablated with an energy density of  $1 \pm 0.1$  J/cm<sup>2</sup> and a repetition rate of 6Hz in a helium environment. A 0.7L/m helium flow transported the resulting aerosol to an in-house-built smoothing device before entering the quadrupole ICP-MS (iCAP-Q, Thermo Fisher Scientific). Masses <sup>7</sup>Li, <sup>11</sup>B, <sup>23</sup>Na, <sup>24</sup>Mg, <sup>25</sup>Mg, <sup>27</sup>Al, <sup>43</sup>Ca, <sup>44</sup>Ca, <sup>57</sup>Fe, <sup>88</sup>Sr, <sup>137</sup>Ba and <sup>238</sup>U were monitored, <sup>44</sup>Ca served as an internal standard for quantification of the associated elements. The synthetic carbonate standard MACS-3 was used for calibration, in addition carbonate standards JcP-1, JcT-1, NFHS1 (NIOZ foraminifera house standard; Mezger et al., 2016) as well as glass standards SRM NIST610 and NIST612 were used for monitoring data quality. Accuracy of the analyses was 97%, while precision was 3.0% for Mg and 2.4% for Na measurements. Stable oxygen and carbon isotopes of foraminiferal calcite were measured on groups of whole specimens different from those used for LA-Q-ICP-MS, using an automated carbonate device (Thermo Kiel IV) which was connected to Thermo Finnigan MAT 253 Dual Inlet Isotope Ratio Mass Spectrometer (IRMS). The NBS 19 limestone was used as a calibration standard, the NFHS1 standard was used for drift detection and correction. The standard deviation and offset of the NBS19 and the NFHS-1 were always within 0.1‰ for  $\delta^{18}\text{O}$ .

Due to the large amount of material required (20 to 40  $\mu\text{g}$ ) and the small amount of specimens present in the samples, specimens from different samples sometimes needed to be combined. This resulted, for example, in combining 12 and 8  $\mu\text{g}$  of foraminifera from two adjacent transects and hence, the average temperature, salinity and  $\delta^{18}\text{O}$  seawater for these transects was calculated based on the relative contribution of the foraminiferal weight of the individual transects (i.e. 60 and 40 % respectively). Sea water oxygen and hydrogen stable isotopes were analysed with the Liquid Water Isotope Analyser (LWIA; Los Gatos Research Model 912-0008). This system measures the water samples using Off-Axis Integrated-Cavity Output Spectroscopy (OA-ICOS). The LWIA was connected with a GC PAL from CTC Analytics to inject 1  $\mu\text{l}$  per measurement. To achieve this, the GCPAL was equipped with a 1.2  $\mu\text{l}$  Hamilton syringe. In-House standards (S35, S45, NSW, LGR5 and double distilled water) were calibrated against VSMOW2-, VSLAP2- and GISP2- standard water obtained from IAEA in Vienna, using the same setup. The use of standard water VSMOW2, which has  $\delta^{18}\text{O}$  values identical to the older SMOW standard, allows for simple comparison with older data that was calibrated using SMOW, without additional corrections. Every sample and standard was measured 14 times sequentially, the first four runs were only used to flush the system while the last 10 measurements were used for the analysis. Additionally, between every sample or standard, the sample introduction line was rinsed with double distilled water. Data were processed with LGR LWIA Post Processor Software v. 3.0.0.88. Average standard deviation per sample was 0.14‰ for oxygen isotope measurements and 0.71‰ for hydrogen isotope measurements.

The likely provenance of the foraminifera sampled was computed by backtracking virtual particles in a high-resolution ocean model. For this, we used the Copernicus Marine Environmental Monitoring Service (CMEMS) Global Reanalysis model. The ocean surface currents, temperature and salinity are available at daily resolution and 1/12 degree horizontal resolution. In these fields, we backtracked particles using the OceanParcels v2.1.1 software (Lange and Sebille, 2017; Delandmeter and van Sebille, 2019). We released 10,000 particles equally spaced between the start and end locations of 25 of the transects (i.e. all for which there were sufficient foraminiferal specimens for isotope analysis), on the day these transects were sampled, and tracked the particles back for 30 days with a 4th order Runge-Kutta algorithm with a 1 hour time step, storing local temperature,

salinity and location for each particle every day. To avoid beaching of particles, we used an unbeaching Kernel similar to that in Delandmeter and van Sebille (2019). The full code of the simulations is available at  
135 [https://github.com/OceanParcels/MedForams\\_Daemmer/](https://github.com/OceanParcels/MedForams_Daemmer/).

### 3 Results

#### 3.1 Mediterranean Sea geochemistry

The sampled East-West transect spans a salinity gradient from 39.2 to 36.2 and an accompanying temperature gradient from 19°C (east) to 14°C. The 6 hour-intervals represented on average a distance of 57 kilometres (min 0 km, max 117 km). On  
140 average, this resulted in an internal variability of 0.14 salinity units and 0.33°C for each of the 98 transects.

Measured sea water  $\delta D$  values show a range from 2.83 to 9.46‰ VSMOW in the western Mediterranean Sea and from 5.98 to 11.15‰ VSMOW in the east. Values from the individual transects were used in combination with the  $\delta^{18}O_{\text{water}}$  to check for internal consistency (Fig. 1). The  $\delta^{18}O$  values of the seawater varies between 0.13 and 2.29‰ VSMOW in the west, and between 0.73 and 2.43‰ VSMOW in the east (Fig. 1). In our dataset,  $\delta^{18}O$  and  $\delta D$  of the water are positively correlated in  
145 both the western and eastern part of the Mediterranean Sea (Fig. 1). The sensitivities of the  $\delta D$  to  $\delta^{18}O$  correlations are, indistinguishable. The sea water oxygen isotopes are also linearly correlated with sea water salinity (Fig. 2) and do not show an offset between the eastern and western basins ( $p$ -value < 0.001;  $R^2 = 0.17$ ).

#### 3.2 Foraminiferal geochemistry

The foraminiferal oxygen isotope ratios ( $\delta^{18}O_{\text{foraminifer}}$ ) range from -0.41 to 0.68‰ and are significantly correlated to seawater oxygen isotope ratio (Fig. 3 a), albeit with much scatter ( $R^2 = 0.42$ ,  $p$ -value < 0.001). The  $\delta^{18}O_{\text{foraminifer}}$  are also positively  
150 correlated with sea surface salinity (Fig. 3 b) showing a similarly large amount of scatter ( $R^2 = 0.44$ ,  $p$ -value < 0.001).

#### 3.3 Na/Ca vs Salinity

Na/Ca values measured on individual F-chambers of *G. ruber albus* from the Mediterranean Sea range from 6.8 to 12.7 mmol/mol and are positively correlated with sea surface salinity (Fig. 4 a). The variability between individuals (1-2 mmol/mol)  
155 observed within transects is orders of magnitude higher than the analytical uncertainty (RSD of 5%) and is also higher than the uncertainty in the slope of the Na/Ca-salinity calibration (Fig. 4 a).

#### 3.4 Mg/Ca vs Temperature

Mg/Ca-values measured on individual F-chambers of *G. ruber albus* from the Mediterranean Sea range from 1.34 to 7.63 mmol/mol and are positively correlated with in-situ measured sea surface temperatures, although the temperature range  
160 sampled during winter time was rather narrow (Fig. 4 b).

### 3.5 Particle backtracking

Particle backtracking shows that foraminifera collected at each transect might actually have travelled long distances within the 30 days prior to sampling at the sample locations. The length of the modelled trajectories varies greatly from location to location, ranging between 200-500km. This resulted in a variability (SD) within one transect ranging from 0.11 to 1.0°C and 0.03 to 0.4 salinity units.

## 4 Discussion

### 4.1 Salinity, $\delta^{18}\text{O}$ and $\delta\text{D}$ of the sea water

A single uniform and stable trend in sea water stable isotopes with salinity is a prerequisite for reconstructing past salinities. This is important not only when using the stable oxygen isotopes measured on foraminiferal shell carbonates, but also for the interpretation of the hydrogen isotopic composition of alkenones, which are also used as proxies for paleo-salinity (Schouten et al., 2006; Vasiliev et al., 2013; Weiss et al., 2017).

The data presented here substantially increases the amount of data on the relation between salinity and water isotopes of the Mediterranean (Fig. 2). Although the new data clearly overlap with existing data, we also observe slight but statistically significant differences in the average salinity to  $\delta^{18}\text{O}$  relationship for the different data sets. The overall lower  $\delta^{18}\text{O}$  values of sea water measured here compared to the combined set of surface sea water isotopes from Stahl and Rinow (1973), Pierre et al. (1986), Gat et al. (1996), Pierre (1999) and Cox (2010) of approximately 0.3‰ (Fig. 2) may be explained by inter-decadal, seasonal and geographical variability between sample sets, or a combination of these factors. Importantly such offsets also give a first order indication of the limit to the accuracy and precision of reconstructions of past salinity using a combined temperature-stable isotope approach from the primary relationship used.

Although Gat et al. (1996) reported a markedly different  $\delta\text{D}/\delta^{18}\text{O}$  relationship for the Eastern Mediterranean Sea compared to that of the Western Mediterranean Sea, our results show no sign of such a longitudinal discontinuity for the same area (Fig. 2). This implies that the water isotopic composition of the entire Mediterranean Sea can primarily be described by a single mixing line between two end-members, with high versus lower salinity, respectively. The remarkable trend between  $\delta\text{D}/\delta^{18}\text{O}$  observed previously by Gat et al. (1996) was explained as a deuterium excess effect due to a combination of the composition of the lowermost air vapor and mixing with the enriched surface waters, most notable in winter months. The discrepancy in  $\delta\text{D}/\delta^{18}\text{O}$  relationship observed between our data and those of Gat et al. (1996) may be due to inter-decadal variability in the hydrological cycle or by differences in seasonal coverage. Potentially the observations of Gat et al. (1996) were hence either related to unusual conditions, spatially restricted features not covered by our sampling locations or the hydrological cycle in the eastern Mediterranean has recently changed considerably. Either way the observed offset between the western and the eastern basin is apparently not stable and should therefore probably not be considered when using Mediterranean stable isotope signatures for reconstructing paleo-salinities.

## 4.2 Na/Ca vs Salinity

The Na/Ca ratios measured on the carbonate shells of *G. ruber albus* from the Mediterranean Sea are significantly and linearly correlated to salinity (Fig. 4 a). This relationship is similar to the one reported previously for plankton pump-collected *G. ruber albus* from the Red Sea (Mezger et al., 2016). Mezger et al. (2016) suggested that there might be a combined effect of different environmental factors such as carbonate chemistry, salinity and temperature on the Na/Ca values in the field-collected specimens. In the Red Sea it is not possible to decouple these factors as they are strongly related. Since in contrast to the Red Sea where there is a strong negative correlation between salinity and temperature, the Mediterranean sea surface salinity and temperature are positively correlated to each other, comparing our data to that of Mezger et al. (2016) allows to decouple temperature from salinity (Fig. 5). This shows that the correlation between foraminiferal Na/Ca values and temperature observed in the Red Sea was not causal and more likely caused by salinity (Mezger et al. 2016). If temperature would have a significant effect on the Na/Ca values in *G. ruber albus*, we would expect different slopes and/or offsets for the Na/Ca to salinity calibrations for the Mediterranean Sea and Red Sea. This implies that temperature has no or only a minor impact on Na/Ca ratios in *G. ruber albus* shells, which is in line with similar findings showing a lack of temperature effect on the Na/Ca of *T. sacculifer* (Bertlich et al., 2018). The average standard deviation in Na/Ca values for a given salinity corresponds approximately to 2 salinity units, using the calibration given here (Fig. 4 a). This large variability is similar to the inter-chamber and inter-specimen variability in other El/Ca ratios, such as for example in Mg/Ca reported by Sadekov et al. (2008) and appears to be inherent to single-chamber El/Ca (de Nooijer et al., 2014b). It has been suggested that such variability between individuals and also between different chambers of the same individual, may be caused by differences in living depth (and hence environmental conditions (Mezger et al., 2018b)), lateral transport (van Sebille et al., 2015) or variability in element incorporation during biomineralization due to vital effects (Erez, 2003; de Nooijer et al., 2014a; Spero et al., 2015) or individual timing of chamber formation (Dämmer et al., 2019). Since specimens used here were collected from surface waters and add new chambers very frequently, vertical or lateral migration into waters with significantly different conditions as suggested by Mezger et al. (2018) and Van Sebille et al. (2015) appears to be an unlikely cause for heterogeneity between specimens in this case. The relatively large scatter in Na/Ca values observed for single chambers (Fig. 4 a) implies that accurate and precise reconstructions of salinity can only be based on combining a substantial number of specimens (Wit et al., 2013).

If salinity is reconstructed from the Na/Ca measurements using the calibration published by Mezger et al. (2016) and compared versus salinity measured in situ in the Mediterranean Sea, the reconstructed salinity follows the in situ measurements closely almost 1:1. The largest deviation from this 1:1 relationship occurs in the lower salinity range, at a salinity of 36.52 the reconstructed salinity estimates underestimate salinity by 0.71 salinity units. The average difference between in situ salinity measurements and salinity reconstructed based on one single-chamber measurement is an underestimation of salinity by 0.46 salinity units. This is still higher than the theoretical uncertainty associated when combining foraminiferal  $\delta^{18}\text{O}$  and

temperatures derived from Mg/Ca measured at exactly the same specimens (Rohling, 2007). An uncertainty (1SD) of 1 °C in  
225 the Mg/Ca-temperature calibration (which may be particularly optimistic at high seawater temperatures), would result in an  
uncertainty of ~0.37 units for the reconstructed difference between two salinities. This approach will lead to an improved  
salinity reconstruction when the (change in) past temperatures are determined more precisely, for example by reducing the  
error through increased sample size. The same applies for salinity reconstructions based on Na/Ca, for which not many  
calibrations are available and hence, leaves room for improvement.

230 While these reconstructions as well as the lack of a strong temperature effect are very encouraging results for the use of Na/Ca  
as a salinity proxy, the incorporation of Na into foraminiferal calcite does not appear to be homogenous across the entire shell.  
It has been shown that the majority of Na in *G. ruber albus* is located in the spines (Mezger et al., 2018a, 2018b), which are  
not well preserved in the fossil record.

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#### 4.3 *G. ruber albus* Mg/Ca values

The increase in Mg/Ca in *G. ruber albus* with temperature (Fig. 4 b) fits recent calibration efforts for Mg-incorporation and  
temperature (e.g. Gray et al., 2018). Since salinity and inorganic carbon chemistry also both affect Mg incorporation in this  
species (Kisakürek et al., 2008; Gray et al., 2018), and the Mediterranean exhibits large gradients in these parameters, it is  
240 necessary to correct measured Mg/Ca values for these parameters. After normalizing Mg/Ca values to a sea water salinity of  
35, using the calibration of Gray et al. (2018), the dependency of the Mg/Ca on temperature is similar to previously reported  
calibrations (e.g. Gray et al., 2018), although the Mg/Ca values at the lower most temperatures appear to be higher than  
expected (Fig. 6). This could potentially be caused by a combination of an underestimation of the salinity effect in these highly  
saline waters, since salinities observed here are well outside the calibration range used by Gray et al. (2018), and low  
245 temperatures, impacting the foraminiferal Mg/Ca comparatively little.

Adding our results to published Mg/Ca-temperature-calibrations for *G. ruber albus* (Anand et al., 2003; Babila et al., 2014;  
Fallet et al., 2010; Friedrich et al., 2012; Gray et al., 2018; Haarmann et al., 2011; Huang et al., 2008; Kisakürek et al., 2008;  
Mathien-Blard and Bassinot, 2009; McConnell and Thunell, 2005; Mohtadi et al., 2009) now extends the combined calibration  
to lower temperatures (i.e. < 18°C), maintaining a comparatively low temperature sensitivity in the colder part of the calibration  
250 (Fig. 6). This not only increases confidence in the application of Mg/Ca in this species as a paleotemperature reconstruction  
tool for colder temperatures, but also support application of individual foraminiferal Mg/Ca values for reconstructing  
seasonality (Wit et al., 2010). Although low densities were reported previously for *G. ruber albus* in the Mediterranean Sea  
during winter time, including being absent in large areas (Pujol and Grazzini, 1995; Bárcena et al., 2004) our finding implies  
that lowest values in Mg/Ca can be related to winter temperatures. *G. ruber albus* is not only present throughout the year as  
255 also shown by Rigual-Hernández et al. (2012) and Avnaim-Katav et al. (2020), but it also registers the in-situ temperature,



also during seasons which are close to its lower temperature limit. Admittedly the large scatter also observed at one single sampling time (i.e. season) makes the deconvolution of seasonality from analyzing single specimen Mg/Ca values challenging.

#### 4.4 $\delta^{18}\text{O}_{\text{foraminifer}}$

##### 260 4.4.1 Role of lateral transport on $\delta^{18}\text{O}_{\text{foraminifer}}$

Horizontal transport of planktonic foraminifera may increase exposure to variable environmental conditions, including different temperatures, salinities and seawater stable isotope compositions (van Sebille et al., 2015). Comparing the sampled transects with the calculated back tracking trajectories shows that especially close to the straits (Alboran Sea and Strait of Sicily) the area where the foraminifera might be derived from, potentially extends over considerable distances and therefore, 265 variability in environmental parameters. With the surface variability in temperature and salinity during the sampling period, the calculated variability in these parameters varied between 0.11 and 1.03°C per transect and 0.04 and 0.39 salinity units per transect (Fig. 7 b, c). This means that the majority of foraminifera experienced a variability of approximately 0.5°C and 0.15 salinity units.

When considering calibrations, this is not affecting the measured proxy variables as the as the difference may be unbiased, but 270 adds to the uncertainty of the environmental parameter to be reconstructed. Since foraminifera grow by periodically adding chambers and since the size of the added chambers increases exponentially in many species, the carbonate added closer to the sampling location makes up a larger proportion of the total shell mass than carbonate added at earlier life stages. Therefore, chambers formed early during a foraminifer's life, have less impact on average shell composition and hence the calibration and the back tracking trajectories (Fig. 7 a-c) thus indicate the largest possible range of conditions experienced by a single 275 foraminifer. This is relevant when considering whole-shell chemistry (i.e. oxygen isotopes; Fig. 4 a) and to a lesser extent also when considering the elemental composition of the final chamber (Fig. 4 b). The last chamber is affected by a much smaller range in environmental conditions, i.e. only the timespan during which the final chamber was built, not more than a few days prior to sampling.

Since  $\delta^{18}\text{O}$  of the calcite could not be measured on F-chambers only, like for element/Ca ratios, and several specimens were 280 needed for a single analysis, results reflect average composition of foraminiferal populations at the sampling areas. The averaging effectively cancels out differences due to inter- and intra-individual variability, but not offsets due to lateral transport. When transport directions are largely uniform, this result in biases and should not add to the scatter in the calcite's isotope composition. Hence this transport affects the calibration, but does not affect precision.

#### 285 4.4.2 Implications for proxies

Combining existing calibrations for foraminiferal Mg/Ca and temperature (Gray et al., 2018) and calibrations relating  $\delta^{18}\text{O}_{\text{foraminifera}}$  with temperature (Mulitza et al., 2003), the  $\delta^{18}\text{O}_{\text{seawater}}$  can be calculated. With our dataset we here assess the quality of such reconstructions by comparison to measured  $\delta^{18}\text{O}_{\text{seawater}}$  (Fig. 8). The Mg/Ca values used here were not corrected for salinity effects, since salinity is the target parameter that has to be reconstructed and is thus treated as unknown. Even  
290 though there is a carbonate ion effect on the Mg/Ca in *G. ruber albus* (Evans et al., 2016; Gray et al., 2018; Kisakürek et al., 2008), the measured values were not corrected for this, since this factor is also unknown in paleo-reconstructions.

Calculated and measured  $\delta^{18}\text{O}_{\text{seawater}}$  do not follow a 1:1 correspondence which could be caused by uncertainties in the different proxy calibrations, analytical uncertainties, heterogeneous element and isotope composition within and between specimens, variability in the location and timing of their calcification and the effect of salinity and pH on Mg/Ca. The lack of a strong  
295 correlation between calculated and measured  $\delta^{18}\text{O}_{\text{seawater}}$  in our dataset implies that calculating salinity from reconstructed  $\delta^{18}\text{O}_{\text{seawater}}$  values will not yield meaningful salinity reconstructions, since reconstructed values for  $\delta^{18}\text{O}_{\text{seawater}}$  are not well correlated to in situ measured  $\delta^{18}\text{O}_{\text{seawater}}$ . Calculating salinities from  $\delta^{18}\text{O}_{\text{seawater}}$  clearly adds much uncertainty due to spatial and temporal variability in the correlation of these two parameters (Conroy et al., 2017; LeGrande and Schmidt, 2006; McConnell et al., 2009).

300 It is important to note that the scatter in the foraminiferal chemistry can only to a small degree be explained by lateral transport (Fig. 7). This effect may be larger in areas where the environmental conditions vary more strongly over the distance travelled by the foraminifer, and/or in basins where there is simply more lateral transport over the foraminifer's lifetime. In our exercise, the calculated trajectories add only a minor component to the uncertainty in T (often within 0.75 °C; Fig. 7) and salinity (often within 0.25 salinity units).

305 In our dataset, the uncertainty in salinity estimates based on  $\delta^{18}\text{O}_{\text{seawater}}$  is much smaller when using in situ measured temperatures (Fig. 8). The sum of squares of the residuals (difference between reconstructed and measured values) is 9.04 when using temperatures derived from Mg/Ca and  $\delta^{18}\text{O}_{\text{foraminifera}}$ , but only 3.56 when using temperatures measured in situ, indicating a better reconstruction.

This shows that the uncertainty or offset in temperatures derived from Mg/Ca, even though the Mg/Ca-temperature relationship  
310 is studied relatively extensively for *G. ruber albus*, is most likely the most limiting step. Even though in our dataset temperatures reconstructed from Mg/Ca deviated less than 2°C from the measured temperature, these small offsets have a large effect on the reconstructed  $\delta^{18}\text{O}_{\text{seawater}}$ . It is therefore crucial to choose temperature proxies carefully, use a large enough number of specimens for analysis, be aware about potential effects of lateral particle transport as well as other environmental parameters, and to be conscious about how errors propagate in paleoclimate reconstructions.

315 Combining all foraminiferal shell chemistry results show that salinities based on  $\delta^{18}\text{O}$  and Mg/Ca may under some specific conditions allow calculating past salinity, but the uncertainties in  $\delta^{18}\text{O}_{\text{seawater}}$  are large even in a setting with a large salinity gradient such as the Mediterranean Sea. This is in line with predictions of uncertainty based on theoretical considerations

(Rohling, 2007). The most limiting step in these calculations is the reconstruction of past temperatures, which should be better than 2 degrees. The development, validation and improvement of other, more direct salinity proxies such as foraminiferal Na/Ca therefore remains crucial for more reliable paleo-salinity reconstructions.

## 5. Conclusion

Using plankton pump samples from the Mediterranean Sea, we showed that 1) the relationship of Mg/Ca in *G. ruber albus* and sea water temperature at lower temperatures follows an exponential relationship, therefore the proxy can now also be applied to lower temperature ranges (<18°C) than before, covering almost the entire temperature tolerance range of that species, though sensitivity of the calibration is comparatively low at low temperatures, 2) the combination of foraminiferal  $\delta^{18}\text{O}$  and Mg/Ca together with assumptions about  $\delta^{18}\text{O}_{\text{seawater}}$  values and  $\delta^{18}\text{O}_{\text{seawater}}$  – salinity relationships does not lead to useful reconstructions of seawater salinity 3) foraminiferal Na/Ca correlates well with sea surface salinity and is independent from temperature, making it a potentially valuable tool for salinity reconstructions.

**Data availability:** Upon publication, the data on which this manuscript is based will be available at the 4TU.Centre for Research Data (doi: 10.4121/uuid:25aa0dc8-2ce2-44e9-acf3-a669f4439932).

**Author contribution:** LKD, LdN and GJR designed the study and performed the sample collection. LKD and JGH prepared and processed the samples and the corresponding data. EvS performed the particle backtracking. All authors were involved in data interpretation. LKD drafted the manuscript with contributions from all authors.

**Competing interests:** The authors declare that they have no conflict of interest.

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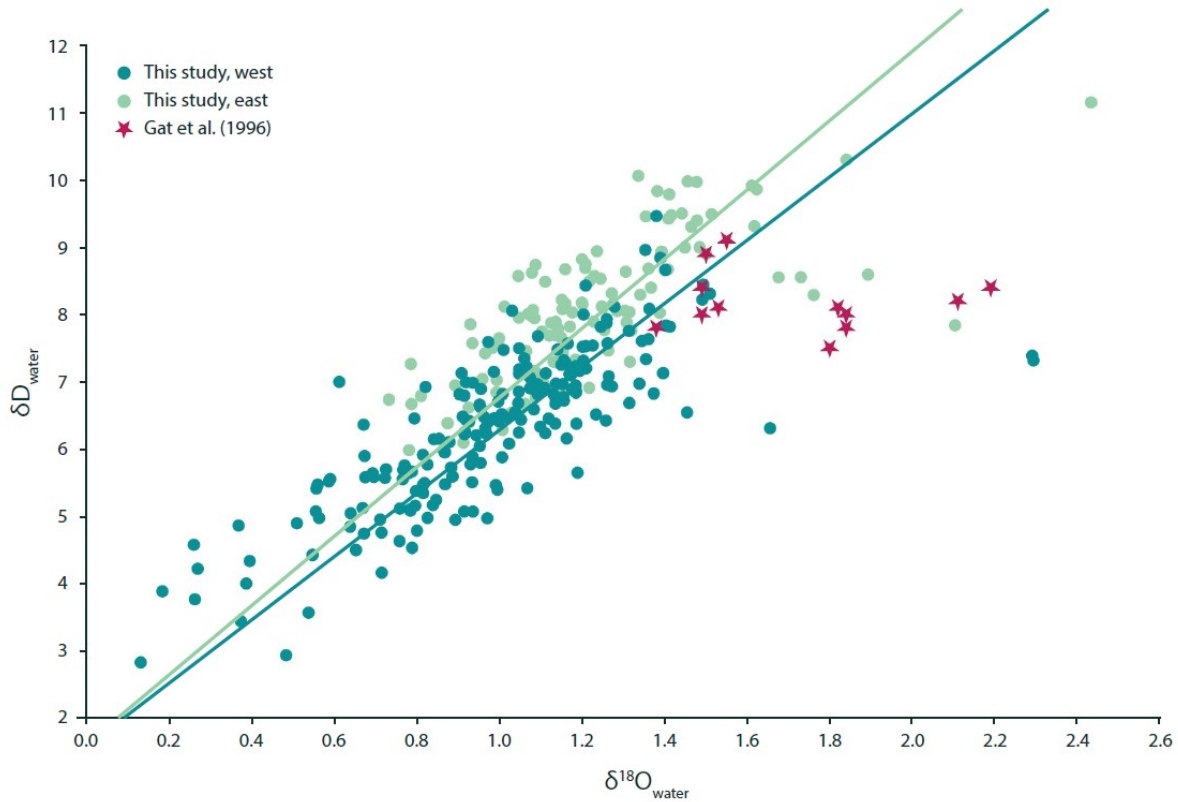
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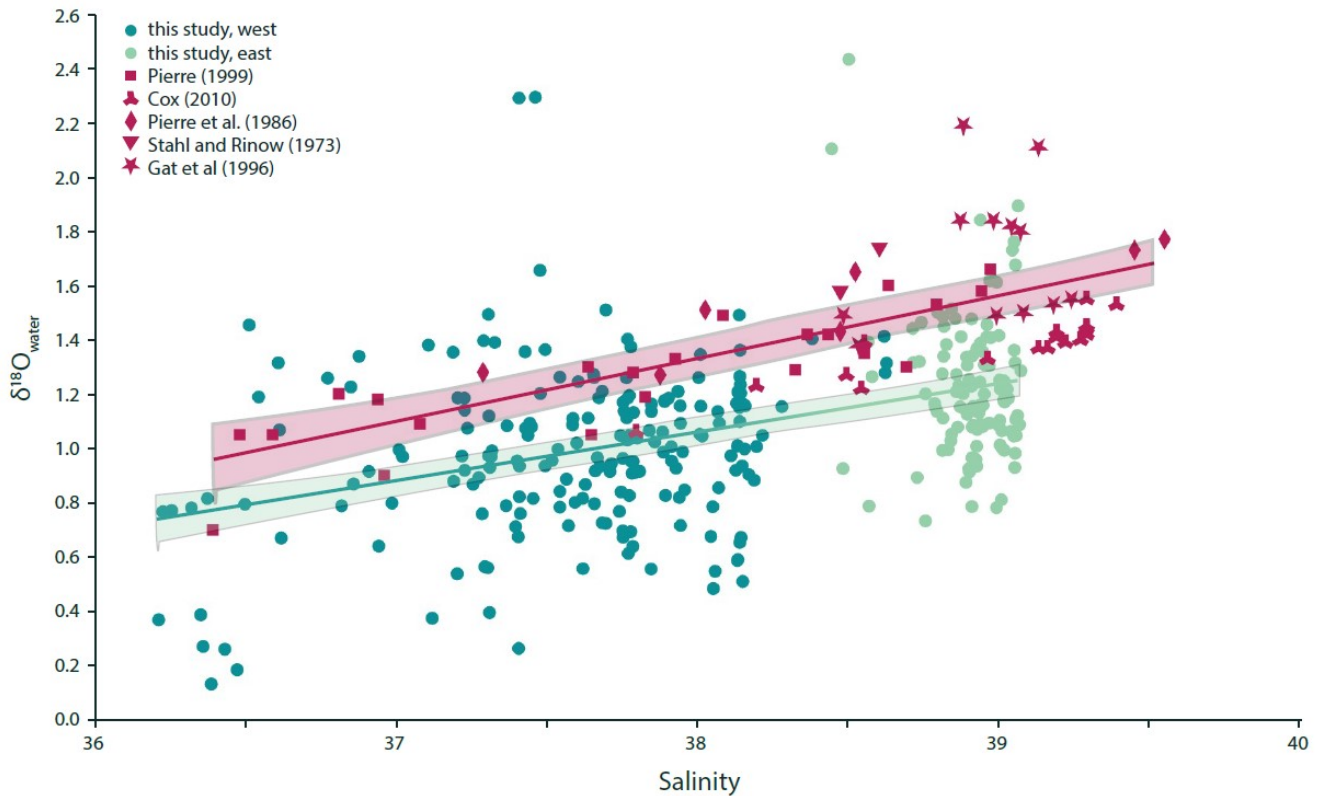
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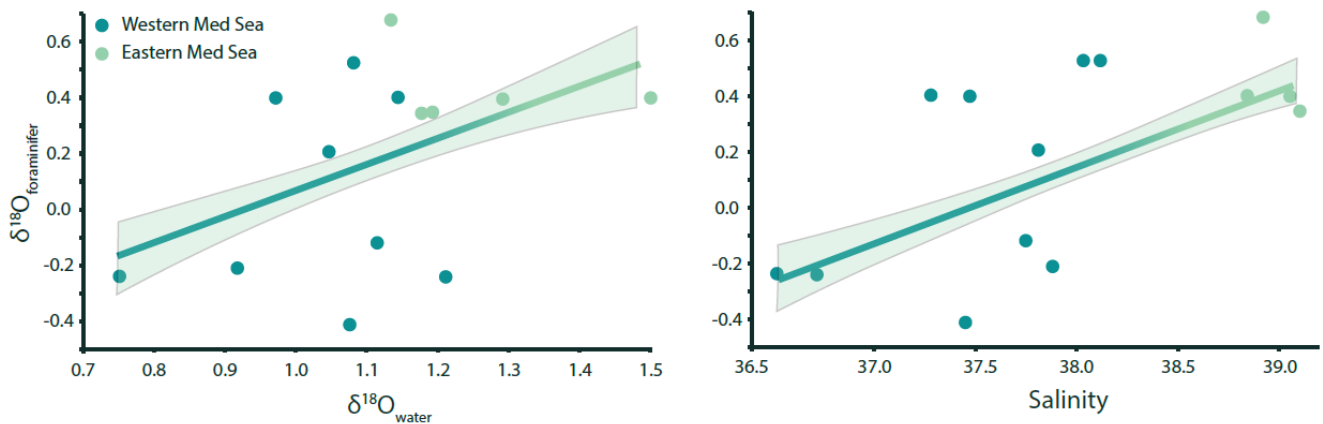
480 **Figure 1: The  $\delta D$  of the Mediterranean surface sea water is positively correlated with the local  $\delta^{18}O$ . The orthogonal regression of the western Mediterranean can be described as  $\delta D_{\text{water}} = 4.72 * \delta^{18}O_{\text{water}} + 1.67$  (dark green). The eastern Mediterranean is very similar to the western basin, the relationship between sea water  $\delta^{18}O$  and  $\delta D$  is  $\delta D_{\text{water}} = 5.19 * \delta^{18}O_{\text{water}} + 1.68$  (light green) here. Statistically they cannot be told apart. This was determined using a bootstrapping approach that generated 100 slopes and intercepts for both the eastern and the western dataset and subsequent t-testing using the mean and standard deviation of both groups of slopes and intercepts, which resulted in p-values > 0.05. In both areas the relationship is different from the observations made by Gat et al. (1996), whose dataset suggested no statistically significant relationship between  $\delta D$  and  $\delta^{18}O$  of the sea water (p-value > 0.05).**

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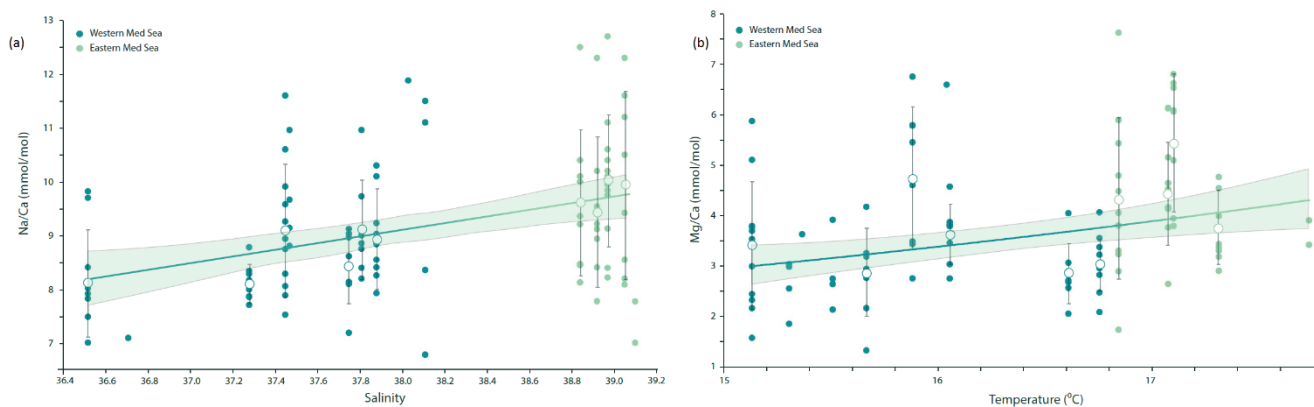


490 **Figure 2:** Surface sea water  $\delta^{18}\text{O}$  is positively correlated with sea surface salinity in the Mediterranean Sea, the relationship observed can be described as linear regression  $\delta^{18}\text{O}_{\text{water}}=0.17*S-5.39$  (p-value < 0.001, adjusted  $R^2=0.17$ ). Previously published data can be combined into one dataset with a similar relationship with a slightly steeper slope, that is offset towards relatively higher  $\delta^{18}\text{O}$  ( $\delta^{18}\text{O}_{\text{water}}=0.22*S-7.19$ ; p-value < 0.001, adjusted  $R^2=0.48$ ). The two regression lines are significantly different from each other (ANOVA p-value < 0.01).

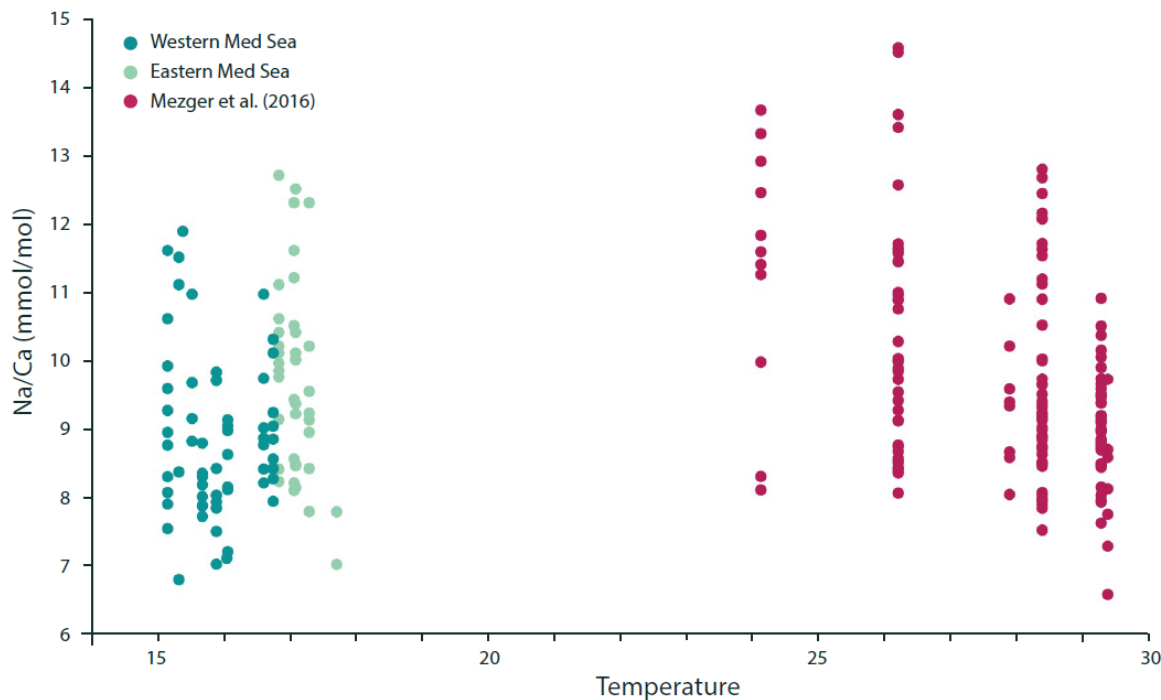


495 **Figure 3:** *G. ruber albus*  $\delta^{18}\text{O}$  measurements are positively correlated (p-value < 0.001) to both sea water  $\delta^{18}\text{O}$  (a) and salinity (b). The relationships can be described using the following equations:  $\delta^{18}\text{O}_{\text{foraminifera}}=0.28*S-10.59$  (adjusted  $R^2 = 0.42$ ) and  $\delta^{18}\text{O}_{\text{foraminifera}}$

$=0.95 * \delta^{18}\text{O}_{\text{water}} - 0.89$  (adjusted  $R^2 = 0.24$ ). The relationship between  $\delta^{18}\text{O}_{\text{water}}$  and salinity in this subset of samples is linear and comparable to that of the entire dataset ( $\delta^{18}\text{O}_{\text{water}} = 0.13 * S - 3.91$ ; p-value < 0.05,  $R^2 = 0.37$ ).

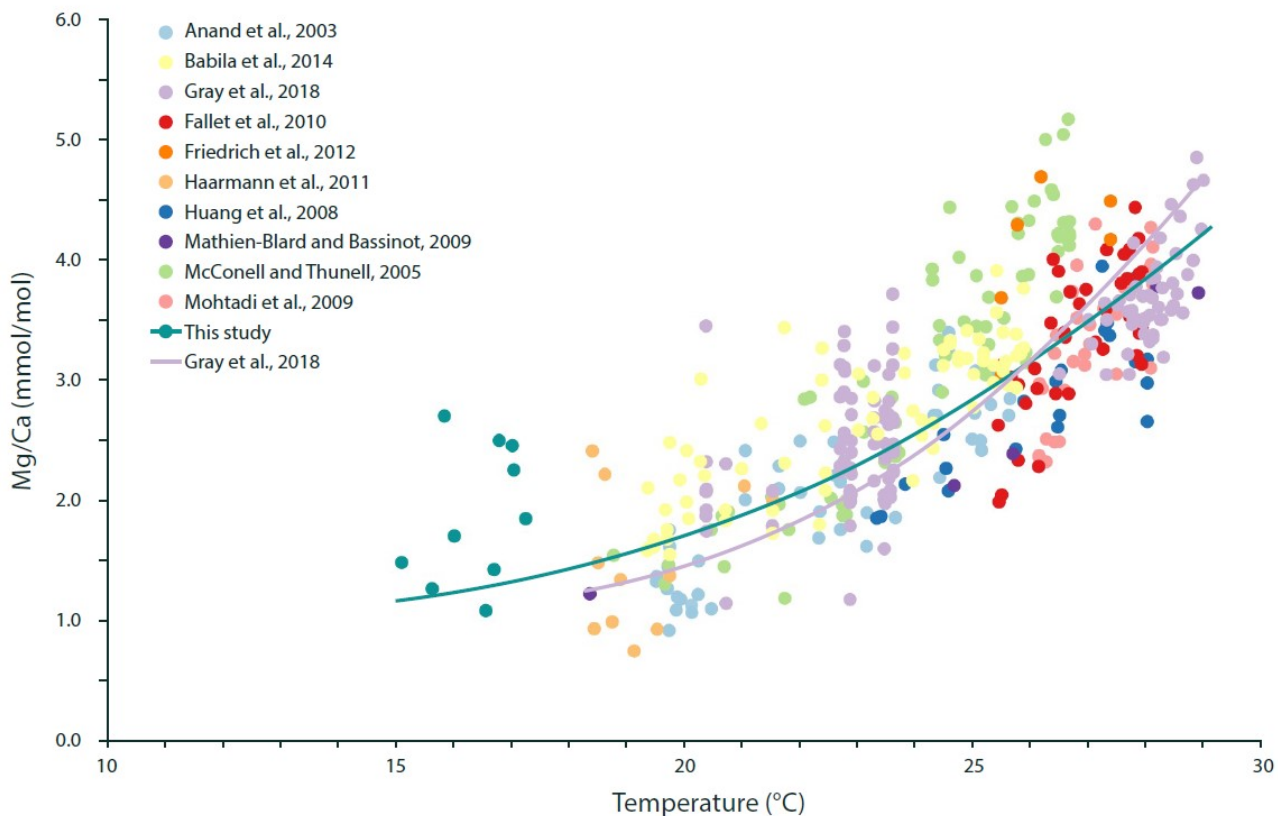


500 **Figure 4: (a) Na/Ca measured in *G. ruber albus* F-chambers collected as living specimens from the eastern and western**  
**Mediterranean Sea correlates well with local salinity (p-value < 0.001,  $\text{Na/Ca} = 0.60 * S - 13.84$ ), even though a large natural spread of**  
**elemental composition around the mean values per station exists ( $R^2 = 0.13$ ). For salinities with more than 5 individual Na/Ca**  
**measurements, hollow circles with whiskers indicate average values and standard deviations. (b) Mg/Ca in F-chambers of *G. ruber***  
***albus* specimens collected from the water column of the Mediterranean Sea is positively correlated with sea surface temperature and**  
 505 **can be described with the exponential relationship  $\text{Mg/Ca} = 0.37 * \exp(0.14 * T)$ . For temperatures with more than 5 individual Mg/Ca**  
**measurements, hollow circles with whiskers indicate average values and standard deviations. Regression lines were calculated using**  
**all individual data points.**

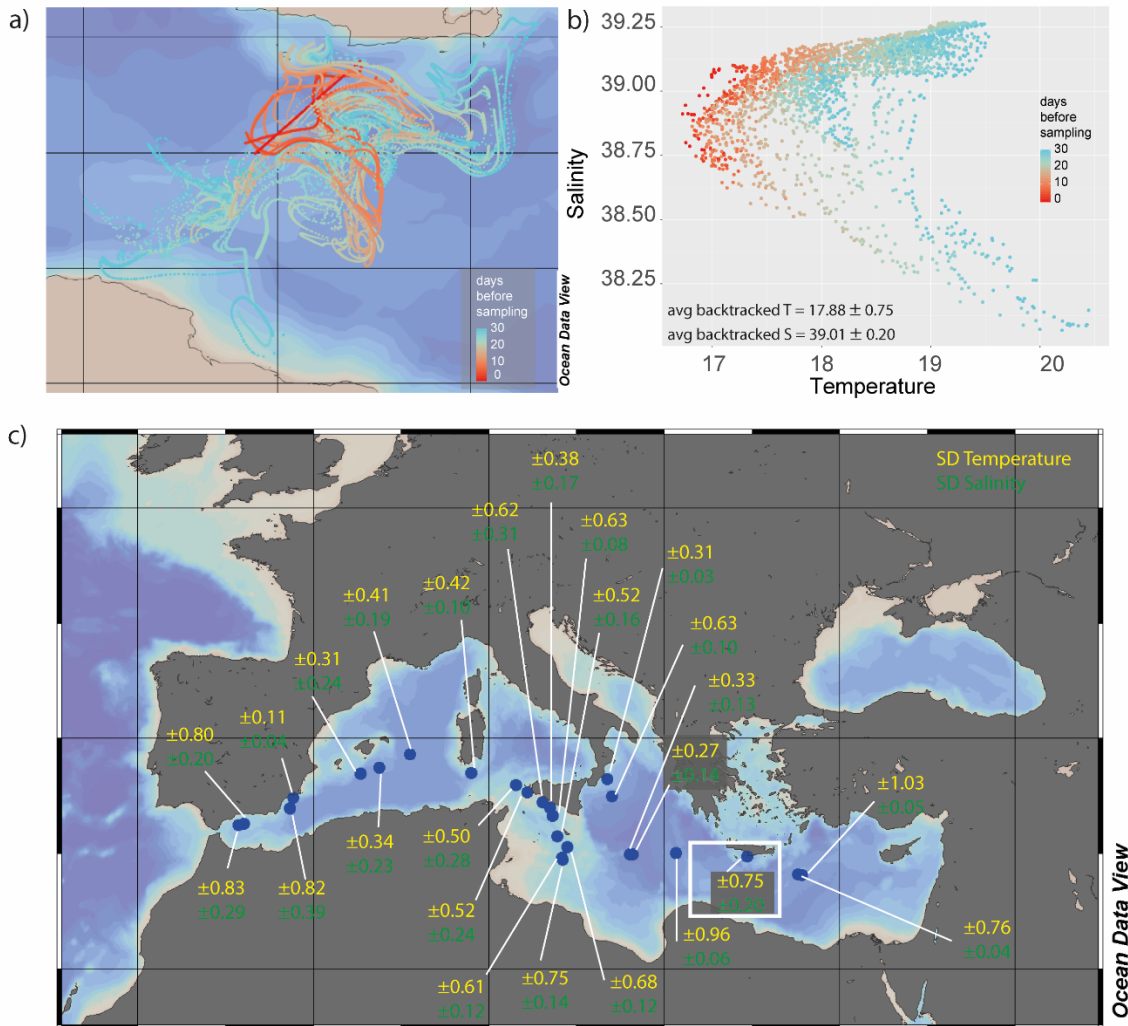


510 **Figure 5:** The ratio of Na/Ca in *G. ruber albus* appears to be independent from sea water temperature. While Mezger et al. (2016) showed a negative relationship between temperature and foraminiferal Na/Ca in specimens collected from the Red Sea, the addition of new data from the Mediterranean Sea shows clearly that the previously hypothesized negative impact of temperature on Na/Ca is likely an artefact of the negative relationship of temperature and salinity in the Red Sea and that temperature has no significant impact on Na/Ca.

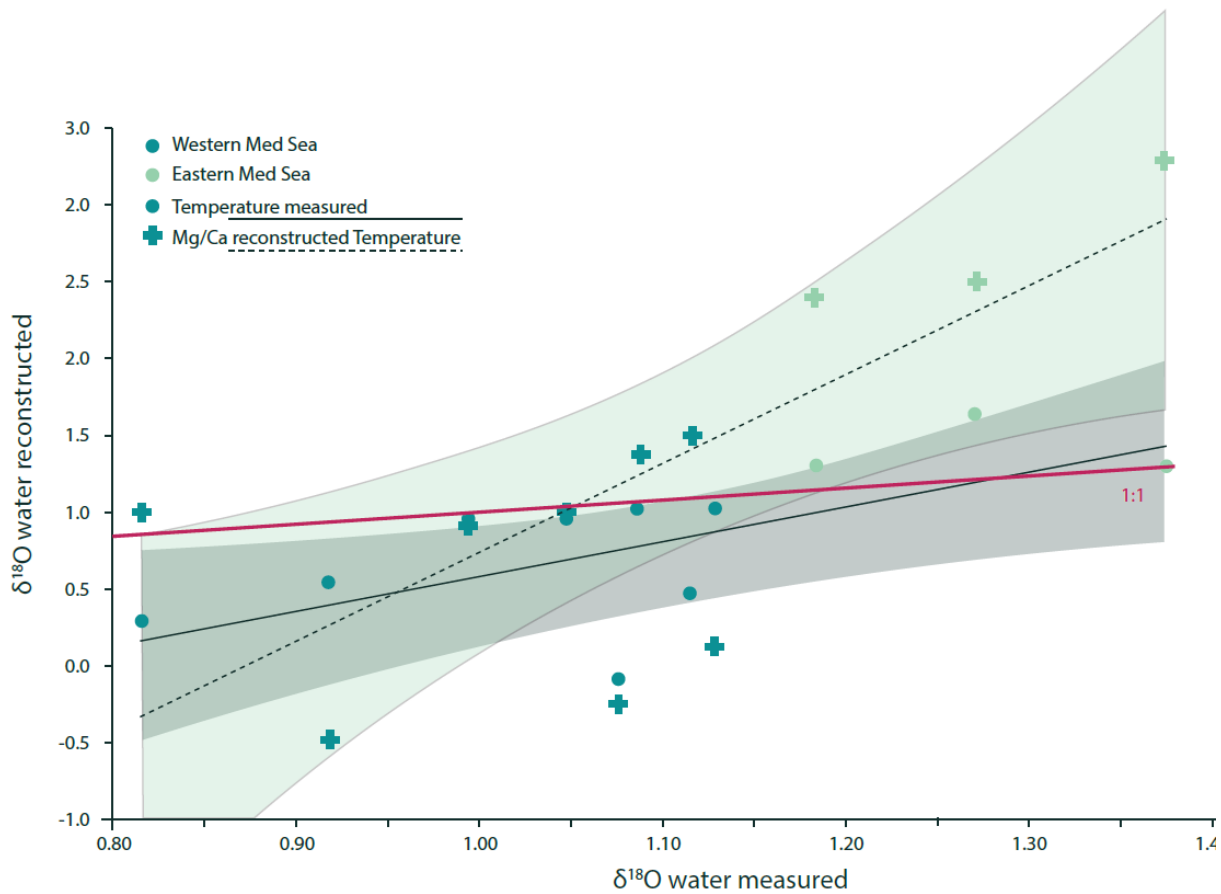
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520 **Figure 6:** The relationship between Mg/Ca in *G. ruber albus* and temperature during calcification can be described using the following exponential equation:  $Mg/Ca=0.278*\exp(0.093*T)$  for a temperature range from 15.1 to 29.1°C.



525 **Figure 7:** (a) Example of back-tracked pathways for a single transect (the one marked by a white rectangle in panel c). The colour indicates the time before sampling up to 30 days. (b) Analysing the different environmental conditions at the different locations of these potential paths show that foraminifera sampled very likely experienced a large range in temperatures as well as salinities. (c) The variability in potentially experienced environmental conditions varies considerably from location to location, as indicated by notation of 1 standard deviation for both parameters for each sampling location. Maps in (a) and (c) were generated using Ocean Data View version 4.



530 **Figure 8: The relationship between  $\delta^{18}\text{O}_{\text{seawater}}$  measured in the Mediterranean Sea and  $\delta^{18}\text{O}_{\text{seawater}}$  calculated from foraminiferal**  
**geochemistry (*G. ruber albus*). The relationship shown with dashed lines and cross shaped markers represents values calculated**  
**using foraminiferal  $\delta^{18}\text{O}$  as well as Mg/Ca as an additional temperature proxy to decouple the effect of temperature and salinity on**  
 **$\delta^{18}\text{O}$ . The relationship shown with the continuous lines and circular markers shows the same samples, but instead of using**  
535 **temperature values derived from foraminiferal Mg/Ca ratios, in situ measurements for temperature were used, the relationship can**  
**be described as  $\delta^{18}\text{O}_{\text{water\_reconstructed}}=2.62(\pm 0.69)*\delta^{18}\text{O}_{\text{water\_measured}}-63.99(\pm 26.11)$  with an adjusted  $R^2$  of 0.37. The temperature gradient**  
**was 2.2°C.**

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