

# Evaluation of oxygen isotopes and trace elements in planktonic foraminifera from the Mediterranean Sea as recorders of seawater oxygen isotopes and salinity

5 Linda K. Dämmer<sup>1</sup>, Lennart de Nooijer<sup>1</sup>, Erik van Sebille<sup>2</sup>, Jan G. Haak<sup>1,2</sup>, Gert-Jan Reichart<sup>1,3</sup>

<sup>1</sup>Department of Ocean Systems, NIOZ Royal Netherlands Institute for Sea Research, and Utrecht University, Texel, The Netherlands

<sup>2</sup>Department of Physics, Institute for Marine and Atmospheric research Utrecht (IMAU), Utrecht University, Utrecht, The Netherlands

10 <sup>3</sup>Department of Earth Sciences, Faculty of Geosciences, Utrecht University, Utrecht, The Netherlands

Correspondence to: Linda K. Dämmer (Linda.Daemmer@nioz.nl)

**Abstract.** The Mediterranean Sea is characterized by a relatively strong west to east salinity gradient, which makes it an area suitable to test the effect of salinity on foraminiferal shell geochemistry. We collected living specimens of the planktonic foraminifer *Globigerinoides ruber albus* (~~white~~) to analyse the relation between element/Ca ratios, stable oxygen isotopes of their shells and surface seawater salinity, isotopic composition and temperature. The oxygen isotopes of sea surface water correlate with salinity in the Mediterranean also during winter, when sampled for this study. Sea water oxygen and hydrogen isotopes are positively correlated in both the eastern and western Mediterranean Sea, though especially in the eastern part the relationship differs from values reported previously for that area. The slope between salinity and seawater oxygen isotopes is lower than previously published. Still, despite the rather modest slope, seawater and foraminiferal carbonate oxygen isotopes are correlated in our dataset although with large residuals and high residual variability. This scatter can be due to either biological variability in vital effects or environmental variability. Numerical models backtracking particles show ocean current driven mixing of particles of different origin might dampen sensitivity and could result in an offset caused by horizontal transport. Results show that Na/Ca is positively correlated to salinity and independent of temperature. Foraminiferal Mg/Ca increases with temperature, as expected, and in line with earlier calibrations, also in the high salinity environment. By using living foraminifera during winter, the previously established Mg/Ca-temperature calibration is extended to temperatures below 18 °C, which is a fundamental prerequisite of using single foraminifera for reconstructing past seasonality.

## 1 Introduction

Ocean circulation plays an important role in Earth's climate, by redistributing heat and also by impacting global biogeochemical cycles. Seawater temperature and salinity are key parameters for reconstructing ocean circulation, since together they determine seawater density and thereby large scale circulation patterns, including a substantial part of meridional overturning circulation. Reconstruction of past ocean environments largely relies on so-called proxy calibrations in which a

variable which can be measured in the geological record is related to a target environmental parameter. The incorporation of trace metals in foraminiferal shell carbonate, for example, is a popular tool to reconstruct past ocean parameters. More specifically, the incorporation of Mg (often expressed as the calcite's Mg/Ca) increases exponentially with seawater temperature, as first observed in culture studies (Nürnberg et al., 1996) and later confirmed by field calibrations (Anand et al., 2003).

In addition to temperature, salinity and inorganic carbon chemistry also affect Mg/Ca in some species of foraminifera (Allison et al., 2011; Dueñas-Bohórquez et al., 2011; Geerken et al., 2018; Gray et al., 2018; Hönisch et al., 2013; Kisakürek et al., 2008; Lea et al., 1999). For the best possible accuracy such effects need to be corrected for when using foraminiferal Mg/Ca for the reconstruction of temperature, which calls for independent proxies for these other environmental parameters.

Currently, salinity is often reconstructed through indirect relationships with other variables, such as the ratio of stable oxygen isotopes of sea water, which are recorded in planktonic foraminifera (Rohling, 2007), although direct approaches have also been suggested recently (Wit et al., 2013; Bertlich et al., 2018). Since seawater oxygen isotope ratio and salinity are both affected by evaporation and precipitation, the two often are linearly related (Rohling, 2007; Bahr et al., 2013), with their calibration depending on local conditions. If foraminifera precipitate their calcite in equilibrium with respect to sea water oxygen isotopes, their  $\delta^{18}\text{O}$  should reflect that of the seawater, and hence salinity. However, as seawater temperature affects stable oxygen isotope fractionation during calcification (McCrea, 1950; Urey et al., 1951) independent temperature reconstructions are needed to estimate seawater  $\delta^{18}\text{O}$  from  $\delta^{18}\text{O}_{\text{calcite}}$  (Rohling, 2007). Independent temperature reconstructions can be based for example on organic proxies such as  $\text{U}^{\text{K}}_{37}$  (Prahl and Wakeham, 1987), TEX86 (Schouten et al., 2006) or the Mg/Ca of the foraminifera themselves (Mashiotta et al., 1999; Elderfield and Ganssen, 2000). Accuracy and precision of such reconstructions is debated because propagation of errors from combined inaccuracies of the analyses and the uncertainties in calibrations due to combining several proxies, is difficult to assess and seems too large for meaningful reconstructions of changes in salinity over time (Rohling, 2007). Because of the lack of a suitable alternative approach, the use of Mg/Ca to determine the temperature effect of foraminiferal  $\delta^{18}\text{O}$  ~~is continued~~ ~~continues~~ to be applied in settings which are prone to large changes in salinity such as the Mediterranean Sea. This calls for an independent in-situ calibration in which all the involved parameters are measured and not determined by proxy-relationships.

Culture experiments using the benthic, symbiont-barrren *Ammonia tepida* (Wit et al., 2013) and the planktonic *Globigerinoides ruber* (pink) (Allen et al., 2016), showed that Na incorporation in foraminiferal shell carbonate is positively correlated with sea water salinity. A field calibration confirmed this positive correlation for the planktonic foraminiferal species *G. ruber albus* (white) (Mezger et al., 2016), as well as for *Trilobatus sacculifer* (previously called *Globigerinoides sacculifer*) in the Red Sea and the Atlantic Ocean (Mezger et al., 2016; Bertlich et al., 2018). Comparison of Na/Ca-salinity calibrations shows, however, that absolute Na/Ca values and also sensitivities to salinity vary between species (Mezger et al., 2016).

When using field calibrations to constrain accuracy and precision of potential reconstruction approaches, it is important to also consider the potential impact of lateral transport of foraminifera due to (ocean) currents. Foraminifera collected at a specific sampling location might actually have added the majority of their shell carbonate at a different location and hence under

different environmental conditions as they have been transported to the sampling location. This may add to the uncertainty in the variable to cross-correlate against or even introduce a bias in the resulting calibration. Recently this has been [shown suggested](#) for dinoflagellate cysts (Nootboom et al., 2019) and planktonic foraminifera, collected from [the water column](#) (Ganssen and Kroon, 1991), [from](#) sediment (van Sebille et al., 2015) and also from sediment traps (Steinhardt et al., 2014), but can also be applied to specimens collected living from the sea surface.

Here we used a plankton pump and sea water samples collected from the Mediterranean Sea in January and February of 2016 to test viability of deconvolving salinity from combined temperature and sea water oxygen isotope reconstructions. We also investigate the potential of the newly developed salinity proxy Na/Ca in the Mediterranean Sea. Using samples collected in winter we also extend the calibration of Mg/Ca to sea water temperature for *G. ruber* [albus](#) towards its lower temperature tolerance limits (14°C; Bijma et al., 1990), which is essential for the application of this species for past seasonality reconstructions.

## 2 Materials and Methods

During two cruises (NESSC Cruises 64PE406 and 64PE407, RV Pelagia) between January 12th and February 25th in 2016, a total of 98 plankton samples were collected from the surface waters of the Mediterranean Sea along an east-west transect using a plankton pump system (Ottens, 1992). Surface water was continuously pumped on board from 5m water depth and lead through a plankton net with 100 µm mesh size. Replacing the cod-end every 6h (filtering 57m<sup>3</sup> of sea water on average, constantly monitored using a water gauge), accumulated samples were washed out of the net into a 90 µm sieve, rinsed thoroughly with deionized water to remove smaller particles as well as salts, and subsequently stored onboard at -80°C. At NIOZ all plankton samples were then freeze-dried, and dry oxidation by low temperature ashing (100°C) was used to combust the organic material while minimizing potential impacts on carbonate trace metal concentrations and δ<sup>18</sup>O (Fallet et al., 2009). After ashing, samples were rinsed again thoroughly with de-ionized water and ethanol to remove potential ash residues. A variety of samples containing specimens of *G. ruber albus* (Morard et al., 2019) ~~(white)~~ was selected to cover a large range in salinities and temperatures. Specimens used for analyses were *G. ruber sensu stricto (Morphotype A)* selected from the size fraction 150 - 250 µm, [even though it has been reported that at this size fraction \*G. ruber albus\* and \*Globigerinoides elongatus\* cannot always be confidently distinguished due to similar morphology](#) (Aurahs et al., 2011). Surface seawater samples for stable oxygen isotopes were collected every 60 minutes from the same pump, resulting in a set of 309 samples. A volume of 2 ml was stored without headspace at 4°C during the cruise to be analyzed at the home laboratory.

The elemental ratio of the final foraminiferal chamber (named the F-chamber) of individual shells were measured by laser ablation quadrupole inductively coupled plasma mass spectrometry (LA-Q-ICP-MS) using a circular spot with a diameter of 60-80 µm, depending on the size of the last chamber. The laser system (NWR193UC, New Wave Research) at Royal NIOZ was used in combination with a two-volume sample cell (TV2), which allows detecting variability in elemental ratios within the foraminiferal chamber wall due to a short wash-out time of 1.8s (van Dijk et al., 2017). Ablating only F-chambers minimizes sampling of older carbonate that might have formed under different environmental conditions due to lateral and

vertical transport. All specimens were ablated with an energy density of  $1 \pm 0.1$  J/cm<sup>2</sup> and a repetition rate of 6Hz in a helium environment. A 0.7L/m helium flow transported the resulting aerosol to an in-house-built smoothing device before entering the quadrupole ICP-MS (iCAP-Q, Thermo Fisher Scientific). Masses 7Li, 11B, 23Na, 24Mg, 25Mg, 27Al, 43Ca, 44Ca, 57Fe, 88Sr, 137Ba and 238U were monitored, 44Ca served as an internal standard for quantification of the associated elements. The synthetic carbonate standard MACS-3 was used for calibration, in addition carbonate standards JCP-1, JCT-1, NFHS1 (NIOZ foraminifera house standard; Mezger et al., 2016) as well as glass standards SRM NIST610 and NIST612 were used for monitoring data quality. Accuracy of the analyses was 97%, while precision was 3.0% for Mg and 2.4% for Na measurements. Stable oxygen and carbon isotopes of foraminiferal calcite were measured [on groups of whole specimens different from those used for LA-Q-ICP-MS](#), using an automated carbonate device (Thermo Kiel IV) which was connected to Thermo Finnigan MAT 253 Dual Inlet Isotope Ratio Mass Spectrometer (IRMS). The NBS 19 limestone was used as a calibration standard, the NFHS1 standard was used for drift detection and correction. The standard deviation and offset of the NBS19 and the NFHS-1 were always within 0.1‰ for  $\delta^{18}\text{O}$ .

Due to the large amount of material required (20 to 40  $\mu\text{g}$ ) and the small amount of specimens present in the samples, specimens from different samples sometimes needed to be combined. This resulted, for example, in combining 12 and 8  $\mu\text{g}$  of foraminifera from two adjacent transects and hence, the average temperature, salinity and  $\delta^{18}\text{O}$  seawater for these transects was calculated based on the relative contribution of the foraminiferal weight of the individual transects (i.e. 60 and 40 % respectively). Sea water oxygen and hydrogen stable isotopes were analysed with the Liquid Water Isotope Analyser (LWIA; Los Gatos Research Model 912-0008). This system measures the water samples using Off-Axis Integrated-Cavity Output Spectroscopy (OA-ICOS). The LWIA was connected with a GC PAL from CTC Analytics to inject 1  $\mu\text{l}$  per measurement. To achieve this, the GCPAL was equipped with a 1.2  $\mu\text{l}$  Hamilton syringe. In-House standards (S35, S45, NSW, LGR5 and double distilled water) were calibrated against VSMOW2-, VSLAP2- and GISP2- standard water obtained from IAEA in Vienna, using the same setup. The use of standard water VSMOW2, which has  $\delta^{18}\text{O}$  values identical to the older SMOW standard, allows for simple comparison with older data that was calibrated using SMOW, without additional corrections. Every sample and standard was measured 14 times sequentially, the first four runs were only used to flush the system while the last 10 measurements were used for the analysis. Additionally, between every sample or standard, the sample introduction line was rinsed with double distilled water. Data were processed with LGR LWIA Post Processor Software v. 3.0.0.88. Average standard deviation per sample was 0.14‰ for oxygen isotope measurements and 0.71‰ for hydrogen isotope measurements.

The likely provenance of the foraminifera sampled was computed by backtracking virtual particles in a high-resolution ocean model. For this, we used the Copernicus Marine Environmental Monitoring Service (CMEMS) Global Reanalysis model. The ocean surface currents, temperature and salinity are available at daily resolution and 1/12 degree horizontal resolution. In these fields, we backtracked particles using the OceanParcels v2.1.1 software (Lange and Seville, 2017; Delandmeter and van Seville, 2019). We released 10,000 particles equally spaced between the start and end locations of 25 of the transects (i.e. all for which there were sufficient foraminiferal specimens for isotope analysis), on the day these transects were sampled, and tracked the particles back for 30 days with a 4th order Runge-Kutta algorithm with a 1 hour time step, storing local temperature,

salinity and location for each particle every day. To avoid beaching of particles, we used an unbeaching Kernel similar to that in Delandmeter and van Sebille (2019). The full code of the simulations is available at  
135 [https://github.com/OceanParcels/MedForams\\_Daemmer/](https://github.com/OceanParcels/MedForams_Daemmer/).

### 3 Results

#### 3.1 The Mediterranean Sea [geochemistry](#)

The sampled East-West transect spans a salinity gradient from 39.2 to 36.2 and an accompanying temperature gradient from 19°C (east) to 14°C. The 6 hour-intervals represented on average a distance of 57 kilometres (min 0 km, max 117 km). On  
140 average, this resulted in an internal variability of 0.14 salinity units and 0.33°C for each of the 98 transects.

Measured sea water  $\delta D$  values show a range from 2.83 to 9.46‰ VSMOW in the western Mediterranean Sea and from 5.98 to 11.15‰ VSMOW in the east. Values from the individual transects were used in combination with the  $\delta^{18}O_{\text{water}}$  to check for internal consistency (Fig. 1). The  $\delta^{18}O$  values of the seawater varies between 0.13 and 2.29‰ VSMOW in the west, and between 0.73 and 2.43‰ VSMOW in the east (Fig. 1). In our dataset,  $\delta^{18}O$  and  $\delta D$  of the water are positively correlated in  
145 both the western and eastern part of the Mediterranean Sea (Fig. 1). The sensitivities of the  $\delta D$  to  $\delta^{18}O$  correlations are, indistinguishable. The sea water oxygen isotopes are also linearly correlated with sea water salinity (Fig. 2) and do not show an offset between the eastern and western basins (p-value < 0.001;  $R^2 = 0.17$ ).

#### 3.2 Foraminiferal geochemistry

The foraminiferal oxygen isotope ratios ( $\delta^{18}O_{\text{foraminifer}}$ ) range from -0.41 to 0.68‰ and are significantly correlated to seawater oxygen isotope ratio (Fig. 3 a), albeit with much scatter ( $R^2 = 0.42$ , p-value < 0.001). The  $\delta^{18}O_{\text{foraminifer}}$  are also positively correlated with sea surface salinity (Fig. 3 b) showing a similarly large amount of scatter ( $R^2 = 0.44$ , p-value < 0.001).

#### 3.3 Na/Ca vs Salinity

Na/Ca values measured on individual F-chambers of *G. ruber* ~~*albus(white)*~~ from the Mediterranean Sea range from 6.8 to 12.7 mmol/mol and are positively correlated with sea surface salinity (Fig. 4 a). The variability between individuals (1-2 mmol/mol)  
155 observed within transects is orders of magnitude higher than the analytical uncertainty (RSD of 5%) and is also higher than the uncertainty in the slope of the Na/Ca-salinity calibration (Fig. 4 a).

#### 3.4 Mg/Ca vs Temperature

Mg/Ca-values measured on individual F-chambers of *G. ruber* ~~*albus(white)*~~ from the Mediterranean Sea range from 1.34 to 7.63 mmol/mol and are positively correlated with in-situ measured sea surface temperatures, although the temperature range  
160 sampled during winter time was rather narrow (Fig. 4 b).

Formatted: Font: Italic

### 3.5 Particle backtracking

Particle backtracking shows that foraminifera collected at each transect might actually have travelled long distances within the 30 days prior to sampling at the sample locations. The length of the modelled trajectories varies greatly from location to location, ranging between 200-500km. This resulted in a variabilities (SD) within one transect ranging from 0.11 to 1.0°C and 0.03 to 0.4 salinity units.

## 4 Discussion

### 4.1 Salinity, $\delta^{18}\text{O}$ and $\delta\text{D}$ of the sea water

A single uniform and stable trend in sea water stable isotopes with salinity is a prerequisite for reconstructing past salinities. This is important not only when using the stable oxygen isotopes measured on foraminiferal shell carbonates, but also for the interpretation of the hydrogen isotopic composition of alkenones, which are also used as proxies for paleo-salinity (Schouten et al., 2006; Vasiliev et al., 2013; Weiss et al., 2017).

The data presented here substantially increases the amount of data on the relation between salinity and water isotopes of the Mediterranean (Fig. 2). Although the new data clearly overlap with existing data, we also observe slight [but statistically significant](#) differences in the average salinity to  $\delta^{18}\text{O}$  relationship for the different data sets. The overall lower  $\delta^{18}\text{O}$  values of sea water measured here compared to the combined set of surface sea water isotopes from Stahl and Rinow (1973), Pierre et al. (1986), Gat et al. (1996), Pierre (1999) and Cox (2010) of approximately 0.3‰ (Fig. 2) may be explained by inter-decadal, seasonal and geographical variability between sample sets, or a combination of these factors. Importantly such offsets also give a first order indication of the limit to the accuracy and precision of reconstructions of past salinity using a combined temperature-stable isotope approach from the primary relationship used.

Although Gat et al. (1996) reported a markedly different  $\delta\text{D}/\delta^{18}\text{O}$  relationship for the Eastern Mediterranean Sea compared to that of the Western Mediterranean Sea, our results show no sign of such a longitudinal discontinuity for the same area (Fig. 2). This implies that the water isotopic composition of the entire Mediterranean Sea can primarily be described by a single mixing line between two end-members, with high versus lower salinity, respectively. The remarkable trend between  $\delta\text{D}/\delta^{18}\text{O}$  observed previously by Gat et al. (1996) was explained as a deuterium excess effect due to a combination of the composition of the lowermost air vapor and mixing with the enriched surface waters, most notable in winter months. The discrepancy in  $\delta\text{D}/\delta^{18}\text{O}$  relationship observed between our data and those of Gat et al. (1996) may be due to inter-decadal variability in the hydrological cycle or by differences in seasonal coverage. Potentially the observations of Gat et al. (1996) were hence either related to unusual conditions, [spatially restricted features not covered by our sampling locations](#) or the hydrological cycle in the eastern Mediterranean has recently changed considerably. Either way the observed offset between the western and the eastern basin is apparently not stable and should therefore probably not be considered when using Mediterranean stable isotope signatures for reconstructing paleo-salinities.

#### 4.2 Na/Ca vs Salinity

The Na/Ca ratios measured on the carbonate shells of *G. ruber albus* from the Mediterranean Sea are significantly and linearly correlated to salinity (Fig. 4 a). This relationship is similar to the one reported previously for plankton pump-collected *G. ruber albus* from the Red Sea (Mezger et al., 2016). Mezger et al. (2016) suggested that there might a combined effect of different environmental factors such as carbonate chemistry, salinity and temperature on the Na/Ca values in the field-collected specimens. In the Red Sea it is not possible to decouple these factors as they are strongly related. Since in contrast to the Red Sea where there is a strong negative correlation between salinity and temperature, the Mediterranean sea surface salinity and temperature are positively correlated to each other, comparing our data to that of Mezger et al. (2016) allows to decouple temperature from salinity (Fig. 5). This shows that the correlation between foraminiferal Na/Ca values and temperature observed in the Red Sea was not causal and more likely caused by salinity (Mezger et al. 2016). If temperature would have a significant effect on the Na/Ca values in *G. ruber albus*, we would expect different slopes and/or offsets for the Na/Ca to salinity calibrations for the Mediterranean Sea and Red Sea. This implies that temperature has no or only a minor impact on Na/Ca ratios in *G. ruber albus* shells, which is in line with similar findings showing a lack of temperature effect on the Na/Ca of *T. sacculifer* (Bertlich et al., 2018).

The average standard deviation in Na/Ca values for a given salinity corresponds approximately to 2 salinity units, using the calibration given here (Fig. 4 a). This large variability is similar to the inter-chamber and inter-specimen variability in other El/Ca ratios, such as for example in Mg/Ca reported by Sadekov et al. (2008) and appears to be inherent to single-chamber El/Ca (de Nooijer et al., 2014b). It has been suggested that such variability between individuals and also between different chambers of the same individual, may be caused by differences in living depth (and hence environmental conditions (Mezger et al., 2018b)), lateral transport (van Sebille et al., 2015) or variability in element incorporation during biomineralization due to vital effects (Erez, 2003; de Nooijer et al., 2014a; Spero et al., 2015) or individual timing of chamber formation (Dämmer et al., 2019). Since specimens used here were collected from surface waters and add new chambers very frequently, vertical or lateral migration into waters with significantly different conditions as suggested by Mezger et al. (2018) and Van Sebille et al. (2015) appears to be an unlikely cause for heterogeneity between specimens in this case. The relatively large scatter in Na/Ca values observed for single chambers (Fig. 4 a) implies that accurate and precise reconstructions of salinity can only be based on combining a substantial number of specimens (Wit et al., 2013).

If salinity is reconstructed from the Na/Ca measurements using the calibration published by Mezger et al. (2016) and compared versus salinity measured in situ in the Mediterranean Sea, the reconstructed salinity follows the in situ measurements closely almost 1:1. The largest deviation from this 1:1 relationship occurs in the lower salinity range, at a salinity of 36.52 the reconstructed salinity estimates underestimate salinity by 0.71 salinity units. The average difference between in situ salinity measurements and salinity reconstructed based on one single-chamber measurement is an underestimation of salinity by 0.46

225 salinity units. This is still higher than the theoretical uncertainty associated when combining foraminiferal  $\delta^{18}\text{O}$  and temperatures derived from Mg/Ca measured at exactly the same specimens (Rohling, 2007). An uncertainty (1SD) of 1 °C in the Mg/Ca-temperature calibration (which may be particularly optimistic at high seawater temperatures), would result in an uncertainty of ~0.37 units for the reconstructed difference between two salinities. This approach will lead to an improved salinity reconstruction when the (change in) past temperatures are determined more precisely, for example by reducing the error through increased sample size. The same applies for salinity reconstructions based on Na/Ca, for which not many calibrations are available and hence, leaves room for improvement.

230 While these reconstructions as well as the lack of a strong temperature effect are very encouraging results for the use of Na/Ca as a salinity proxy, the incorporation of Na into foraminiferal calcite does not appear to be homogenous across the entire shell. It has been shown that the majority of Na in *G. ruber albus* is located in the spines (Mezger et al., 2018a, 2018b), which are not well preserved in the fossil record.

#### 4.3 *G. ruber albus* Mg/Ca values

240 The increase in Mg/Ca in *G. ruber albus* with temperature (Fig. 4 b) fits recent calibration efforts for Mg-incorporation and temperature (e.g. Gray et al., 2018). Since salinity and inorganic carbon chemistry also both affect Mg incorporation in this species (Kisakürek et al., 2008; Gray et al., 2018), and the Mediterranean exhibits large gradients in these parameters, it is necessary to correct measured Mg/Ca values for these parameters. After normalizing Mg/Ca values to a sea water salinity of 35, using the calibration of Gray et al. (2018), the dependency of the Mg/Ca on temperature- is similar to previously reported calibrations (e.g. Gray et al., 2018), although the Mg/Ca values at the lower most temperatures tend appear to be somewhat higher than expected (Fig. 68). This could potentially be caused by a combination of an underestimation of the salinity effect in these highly saline waters, since salinities observed here are well outside the calibration range used by Gray et al. (2018), and low temperatures, impacting the foraminiferal Mg/Ca comparatively little.

245 Adding our results to published Mg/Ca-temperature-calibrations for *G. ruber albus* (Anand et al., 2003; Babila et al., 2014; Fallet et al., 2010; Friedrich et al., 2012; Gray et al., 2018; Haarmann et al., 2011; Huang et al., 2008; Kisakürek et al., 2008; Mathien-Blard and Bassinot, 2009; McConnell and Thunell, 2005; Mohtadi et al., 2009) now extends the combined calibration to lower temperatures (i.e. < 18°C), maintaining the same comparatively low temperature sensitivity in the colder part of the calibration (Fig. 6). This not only increases confidence in the application of Mg/Ca in this species as a paleotemperature reconstruction tool for colder temperatures, but also support application of individual foraminiferal Mg/Ca values for reconstructing seasonality (Wit et al., 2010). Although low densities were reported previously for *G. ruber albus* in the Mediterranean Sea during winter time, including being absent in large areas (Pujol and Grazzini, 1995; Bárcena et al., 2004) 255 our finding implies that lowest values in Mg/Ca can be related to winter temperatures. *G. ruber albus* is not only present

Formatted: Font: Italic

Formatted: Font: Italic



throughout the year [as also shown by Rigual-Hernández et al. \(2012\)](#) and Avnaim-Katav et al. (2020), but it also registers the in-situ temperature, also during seasons which are close to its lower temperature limit. Admittedly the large scatter also observed at one single sampling time (i.e. season) makes the deconvolution of seasonality from analyzing single specimen Mg/Ca values challenging.

260

#### 4.4 $\delta^{18}\text{O}_{\text{foraminifer}}$ + particle backtracking

##### 4.4.1 Role of lateral transport on $\delta^{18}\text{O}_{\text{foraminifer}}$ Particle backtracking

Horizontal transport of planktonic foraminifera may increase exposure to variable environmental conditions, including different temperatures, salinities and seawater stable isotope compositions (van Sebille et al., 2015). Comparing the sampled transects with the calculated back tracking trajectories shows that especially close to the straits (Alboran Sea and Strait of Sicily) the area where the foraminifera might be derived from, potentially extends over considerable distances and therefore, variability in environmental parameters. With the surface variability in temperature and salinity during the sampling period, the calculated variability in these parameters varied between  $\pm(\text{1SD})$  0.11 and 1.03°C per transect and 0.04 and 0.39 salinity units per transect (Fig. 7 b, c). This means that the majority of foraminifera experienced a variability of approximately 0.5°C and 0.15 salinity units.

265

270

When considering calibrations, this is not affecting the measured proxy variables as the as the difference may be unbiased, but adds to the uncertainty of the environmental parameter to be reconstructed. Since foraminifera grow by periodically adding chambers and since the size of the added chambers increases exponentially in many species, the carbonate added closer to the sampling location makes up a larger proportion of the total shell mass than carbonate added at earlier life stages. ~~This implies that although the first chambers mostly formed further away from the sampling location, this~~ Therefore, chambers formed early during a foraminifer's life, ~~has have a relatively less minor~~ impact on average shell composition and hence the calibration. ~~Therefore, the plotted and the~~ back tracking trajectories (Fig. 7 a-c) thus indicate the largest possible range of conditions experienced by a single foraminifer. This is relevant when considering whole-shell chemistry (i.e. oxygen isotopes; Fig. 4 a) and to a lesser extent also when considering the elemental composition of the final chamber (Fig. 4 b). The last chamber is affected by a much smaller range in environmental conditions, i.e. only the timespan during which the final chamber was built, not more than a few days prior to sampling.

275

280

##### 4.4.2 ~~Impact on $\delta^{18}\text{O}_{\text{foraminifer}}$~~

Since  $\delta^{18}\text{O}$  of the calcite could not be measured on F-chambers only, like for element/Ca ratios, and several specimens were needed for a single analysis, results reflect average composition of foraminiferal populations at the sampling areas. The averaging effectively cancels out differences due to inter- and intra-individual variability, but not offsets due to lateral

285

transport. When transport directions are largely uniform, this result in biases and should not add to the scatter in the calcite's isotope composition. Hence this transport affects the calibration, but does not affect precision.

#### 4.4.23 Implications for proxies

290 Combining existing calibrations for foraminiferal Mg/Ca and temperature (Gray et al., 2018) and calibrations relating  $\delta^{18}\text{O}_{\text{foraminifera}}$  with temperature (Mulitza et al., 2003), the  $\delta^{18}\text{O}_{\text{seawater}}$  can be calculated. With our dataset we here assess the quality of such reconstructions by comparison to measured  $\delta^{18}\text{O}_{\text{seawater}}$  (Fig. 8). The Mg/Ca values used here were not corrected for salinity effects, since salinity is the target parameter that has to be reconstructed and is thus treated as unknown. Even though there is a carbonate ion effect on the Mg/Ca in *G. ruber albus* (Evans et al., 2016; Gray et al., 2018; Kisakürek et al.,  
295 2008), the measured values were not corrected for this, since this factor is also unknown in paleo-reconstructions.

Calculated and measured  $\delta^{18}\text{O}_{\text{seawater}}$  do not ~~only not~~ follow a 1:1 correspondence, ~~but are not correlated at all ( $p$ -value  $> 0.05$ )~~ which ~~is likely due to~~ could be caused by uncertainties in the different proxy calibrations, analytical uncertainties, heterogeneous element and isotope composition within and between specimens, ~~and~~ variability in the location and timing of their calcification and the effect of salinity and pH on Mg/Ca. The lack of a strong correlation between calculated and measured  $\delta^{18}\text{O}_{\text{seawater}}$  in our dataset implies that calculating salinity from reconstructed  $\delta^{18}\text{O}_{\text{seawater}}$  values will not yield meaningful salinity reconstructions, since reconstructed values for  $\delta^{18}\text{O}_{\text{seawater}}$  are not well correlated to in situ measured  $\delta^{18}\text{O}_{\text{seawater}}$ . Calculating salinities from  $\delta^{18}\text{O}_{\text{seawater}}$  clearly adds much uncertainty due to spatial and temporal variability in the correlation of these two parameters (Conroy et al., 2017; LeGrande and Schmidt, 2006; McConnell et al., 2009).

It is important to note that the scatter in the foraminiferal chemistry can only to a small degree be explained by lateral transport (Fig. 7). This effect may be larger in areas where the environmental conditions vary more strongly over the distance travelled by the foraminifer, and/or in basins where there is simply more lateral transport over the foraminifer's lifetime. In our exercise, the calculated trajectories add only a minor component to the uncertainty in T (often within 0.75 °C; Fig. 7) and salinity (often within 0.25 salinity units).

In our dataset, the uncertainty in salinity estimates based on  $\delta^{18}\text{O}_{\text{seawater}}$  is much smaller when using in situ measured temperatures (Fig. 8). The sum of squares of the residuals (difference between reconstructed and measured values) is 9.04 when using temperatures derived from Mg/Ca and  $\delta^{18}\text{O}_{\text{foraminifera}}$ , but only 3.56 when using temperatures measured in situ, indicating a better reconstruction.

This shows that the uncertainty or offset in temperatures derived from Mg/Ca, even though the Mg/Ca-temperature relationship is studied relatively extensively for *G. ruber albus*, is most likely the most limiting step. Even though in our dataset temperatures reconstructed from Mg/Ca deviated less than 2°C from the measured temperature, these small offsets have a large effect on the reconstructed  $\delta^{18}\text{O}_{\text{seawater}}$ . It is therefore crucial to choose temperature proxies carefully, use a large enough number of specimens for analysis, be aware about potential effects of lateral particle transport as well as other environmental parameters, and to be conscious about how errors propagate in paleoclimate reconstructions.

320 Combining all foraminiferal shell chemistry results show that salinities based on  $\delta^{18}\text{O}$  and Mg/Ca may under some specific conditions allow calculating past salinity, but the uncertainties in  $\delta^{18}\text{O}_{\text{seawater}}$  are large even in a setting with a large salinity gradient such as the Mediterranean Sea. This is in line with predictions of uncertainty based on theoretical considerations (Rohling, 2007). The most limiting step in these calculations is the reconstruction of past temperatures, which should be better than 2 degrees. The development, validation and improvement of other, more direct salinity proxies such as foraminiferal Na/Ca therefore remains crucial for more reliable paleo-salinity reconstructions.

## 325 5. Conclusion

Using plankton pump samples from the Mediterranean Sea, we showed that 1) the relationship of Mg/Ca in *G. ruber albus* and sea water temperature at lower temperatures follows an exponential relationship, therefore the proxy can now also be applied to lower temperature ranges (<18°C) than before, covering almost the entire temperature tolerance range of that species, though sensitivity of the calibration is comparatively low at low temperatures, 2) the combination of foraminiferal  $\delta^{18}\text{O}$  and Mg/Ca together with assumptions about  $\delta^{18}\text{O}_{\text{seawater}}$  values and  $\delta^{18}\text{O}_{\text{seawater}}$  – salinity relationships does not lead to useful reconstructions of seawater salinity 3) foraminiferal Na/Ca correlates well with sea surface salinity and is independent from temperature, making it a potentially valuable tool for salinity reconstructions.

335 **Data availability:** Upon publication, the data on which this manuscript is based will be available at the 4TU.Centre for Research Data ([data.4tu.nl/repository](https://data.4tu.nl/repository)).

**Author contribution:** [LKD, LdN and GJR designed the study and performed the sample collection. LKD and JGH prepared and processed the samples and the corresponding data. EvS performed the particle backtracking. All authors were involved in data interpretation. LKD drafted the manuscript with contributions from all authors.](#)

**Competing interests:** [The authors declare that they have no conflict of interest.](#)

345 **Acknowledgements:** [We thank the editor as well as Michal Kucera and a second, anonymous referee for their comments and suggestions that helped to improve this manuscript. We thank the captain, crew and scientific staff of RV Pelagia NESSC Cruises 64PE406 and 64PE407, especially Anne Roepert \(Utrecht University\) who took the majority of the sea water samples used in this study. We also thank Sharyn Ossebaar, Piet van Gaever and Wim Boer \(NIOZ\) for their vital technical assistance, as well as Prof Geert-Jan A. Brummer \(NIOZ/Vrije Universiteit Amsterdam\) for his support during sample preparation and many discussions. This work was carried out under the program of the Netherlands Earth System Science Centre \(NESSC\), financially supported by the Ministry of Education, Culture and Science \(OCW\) \(grant no. 024.002.001\).](#)

## References

- Allen, K. A., Hönisch, B., Eggins, S. M., Haynes, L. L., Rosenthal, Y. and Yu, J.: Trace element proxies for surface ocean conditions: A synthesis of culture calibrations with planktic foraminifera, *Geochim. Cosmochim. Acta*, 193, 197–221, doi:10.1016/j.gca.2016.08.015, 2016.
- 355 Allison, N., Austin, H., Austin, W. and Paterson, D. M.: Effects of seawater pH and calcification rate on test Mg/Ca and Sr/Ca in cultured individuals of the benthic, calcitic foraminifera *Elphidium williamsoni*, *Chem. Geol.*, 289(1–2), 171–178, doi:10.1016/j.chemgeo.2011.08.001, 2011.
- Anand, P., Elderfield, H. and Conte, M. H.: Calibration of Mg/Ca thermometry in planktonic foraminifera from a sediment trap time series, *Paleoceanography*, 18(2), n/a-n/a, doi:10.1029/2002PA000846, 2003.
- 360 Aurahs, R., Treis, Y., Darling, K. and Kucera, M.: A revised taxonomic and phylogenetic concept for the planktonic foraminifer species *Globigerinoides ruber* based on molecular and morphometric evidence, *Mar. Micropaleontol.*, 79(1–2), 1–14, doi:10.1016/j.marmicro.2010.12.001, 2011.
- Avnaim-Katav, S., Herut, B., Rahav, E., Katz, T., Weinstein, Y., Alkalay, R., Berman-Frank, I., Zlatkin, O. and Almogi-Labin, A.: Sediment trap and deep sea core top sediments as tracers of recent changes in planktonic foraminifera assemblages in the southeastern ultra-oligotrophic Levantine Basin, *Deep. Res. Part II Top. Stud. Oceanogr.*, 171(September 2019), 104669, doi:10.1016/j.dsr2.2019.104669, 2020.
- 365 Babila, T. L., Rosenthal, Y. and Conte, M. H.: Evaluation of the biogeochemical controls on B/Ca of *Globigerinoides ruber* white from the Oceanic Flux Program, Bermuda, *Earth Planet. Sci. Lett.*, 404, 67–76, doi:10.1016/j.epsl.2014.05.053, 2014.
- 370 Bertlich, J., Nürnberg, D., Hathorne, E. C., de Nooijer, L. J., Mezger, E. M., Kienast, M., Nordhausen, S., Reichart, G.-J., Schönfeld, J. and Bijma, J.: Salinity control on Na incorporation into calcite tests of the planktonic foraminifera *Trilobatus sacculifer* – evidence from culture experiments and surface sediments, *Biogeosciences*, 5991–6018, 2018.
- Conroy, J. L., Thompson, D. M., Cobb, K. M., Noone, D., Rea, S. and Legrande, A. N.: Spatiotemporal variability in the  $\delta^{18}\text{O}$ -salinity relationship of seawater across the tropical Pacific Ocean, *Paleoceanography*, 32(5), 484–497, doi:10.1002/2016PA003073, 2017.
- 375 Cox, K.: *Stable Isotopes as Tracers for Freshwater Fluxes into the North Atlantic*, University of Southampton., 2010.
- Dämmer, L. K., Nooijer, L. J. De and Reichart, G. J.: Light Impacts Mg Incorporation in the Benthic Foraminifer *Amphistegina lessonii*, *Front. Mar. Sci.*, 6(July), 1–8, doi:10.3389/fmars.2019.00473, 2019.
- Dueñas-Bohórquez, A., Raitzsch, M., de Nooijer, L. J. and Reichart, G.-J.: Independent impacts of calcium and carbonate ion concentration on Mg and Sr incorporation in cultured benthic foraminifera, *Mar. Micropaleontol.*, 81(3–4), 122–130, doi:10.1016/j.marmicro.2011.08.002, 2011.
- 380 Erez, J.: The source of ions for biomineralization in foraminifera and their implications for paleoceanographic proxies (Review), *Rev. Mineral. geochemistry*, 54(1), 115, doi:10.2113/0540115, 2003.

- Evans, D., Wade, B. S., Henehan, M., Erez, J. and Müller, W.: Revisiting carbonate chemistry controls on planktic foraminifera  
385 Mg / Ca: Implications for sea surface temperature and hydrology shifts over the Paleocene-Eocene Thermal Maximum and  
Eocene-Oligocene transition, *Clim. Past*, 12(4), 819–835, doi:10.5194/cp-12-819-2016, 2016.
- Fallet, U., Brummer, G. J., Zinke, J., Vogels, S. and Ridderinkhof, H.: Contrasting seasonal fluxes of planktonic foraminifera  
and impacts on paleothermometry in the Mozambique Channel upstream of the Agulhas Current, *Paleoceanography*, 25(4),  
doi:10.1029/2010PA001942, 2010.
- 390 Friedrich, O., Schiebel, R., Wilson, P. A., Weldeab, S., Beer, C. J., Cooper, M. J. and Fiebig, J.: Influence of test size, water  
depth, and ecology on Mg/Ca, Sr/Ca,  $\delta^{18}\text{O}$  and  $\delta^{13}\text{C}$  in nine modern species of planktic foraminifers, *Earth Planet. Sci. Lett.*,  
319–320, 133–145, doi:10.1016/j.epsl.2011.12.002, 2012.
- Ganssen, G. and Kroon, D.: Evidence for Red Sea surface circulation from oxygen isotopes of modern surface waters and  
planktonic foraminiferal tests, *Paleoceanography*, 6(1), 73–82, 1991.
- 395 Gat, J. R., Shemesh, A., Tziperman, E., Hecht, A., Georgopoulos, D. and Basturk, O.: The stable isotope composition of waters  
of the eastern Mediterranean Sea, *Earth Planet. Sci. Lett.*, 101, 6441–6451, doi:10.1016/0012-821X(84)90103-1, 1996.
- Geerken, E., de Nooijer, L. J., van Dijk, I. and Reichart, G.-J.: Impact of salinity on element incorporation in two benthic  
foraminiferal species with contrasting magnesium contents, *Biogeosciences*, 15(7), 2205–2218, doi:10.5194/bg-15-2205-  
2018, 2018.
- 400 Gray, W. R., Weldeab, S., Lea, D. W., Rosenthal, Y., Gruber, N., Donner, B. and Fischer, G.: The effects of temperature,  
salinity, and the carbonate system on Mg/Ca in *Globigerinoides ruber* (white): A global sediment trap calibration, *Earth Planet.*  
*Sci. Lett.*, 482, 607–620, doi:10.1016/j.epsl.2017.11.026, 2018.
- Haarmann, T., Hathorne, E. C., Mohtadi, M., Groeneveld, J., Klling, M. and Bickert, T.: Mg/Ca ratios of single planktonic  
foraminifer shells and the potential to reconstruct the thermal seasonality of the water column, *Paleoceanography*, 26(3), 1–  
405 14, doi:10.1029/2010PA002091, 2011.
- Hönisch, B., Allen, K. A., Lea, D. W., Spero, H. J., Eggins, S. M., Arbuszewski, J., DeMenocal, P., Rosenthal, Y., Russell, A.  
D. and Elderfield, H.: The influence of salinity on Mg/Ca in planktic foraminifers - Evidence from cultures, core-top sediments  
and complementary  $\delta^{18}\text{O}$ , *Geochim. Cosmochim. Acta*, 121, 196–213, doi:10.1016/j.gca.2013.07.028, 2013.
- Huang, K. F., You, C. F., Lin, H. L. and Shieh, Y. T.: In situ calibration of Mg/Ca ratio in planktonic foraminiferal shell using  
410 time series sediment trap: A case study of intense dissolution artifact in the South China Sea, *Geochemistry, Geophys.*  
*Geosystems*, 9(4), doi:10.1029/2007GC001660, 2008.
- Kisakürek, B., Eisenhauer, A., Böhm, F., Garbe-Schönberg, D. and Erez, J.: Controls on shell Mg/Ca and Sr/Ca in cultured  
planktonic foraminiferan, *Globigerinoides ruber* (white), *Earth Planet. Sci. Lett.*, 273(3–4), 260–269,  
doi:10.1016/j.epsl.2008.06.026, 2008.
- 415 Lea, D. W., Mashiotta, T. A. and Spero, H. J.: Controls on magnesium and strontium uptake in planktonic foraminifera  
determined by live culturing, *Geochim. Cosmochim. Acta*, 63(16), 2369–2379, doi:10.1016/S0016-7037(99)00197-0, 1999.
- LeGrande, A. N. and Schmidt, G. A.: Global gridded data set of the oxygen isotopic composition in sea water, *Geophys. Res.*

- Let., 33(12), 1–5, doi:10.1029/2006GL026011, 2006.
- Mathien-Blard, E. and Bassinot, F.: Salinity bias on the foraminifera Mg/Ca thermometry: Correction procedure and implications for past ocean hydrographic reconstructions, *Geochemistry, Geophys. Geosystems*, 10(12), doi:10.1029/2008GC002353, 2009.
- McConnell, M. C. and Thunell, R. C.: Calibration of the planktonic foraminiferal Mg/Ca paleothermometer: Sediment trap results from the Guaymas Basin, Gulf of California, *Paleoceanography*, 20(2), 1–18, doi:10.1029/2004PA001077, 2005.
- McConnell, M. C., Thunell, R. C., Lorenzoni, L., Astor, Y., Wright, J. D. and Fairbanks, R.: Seasonal variability in the salinity and oxygen isotopic composition of seawater from the Cariaco Basin, Venezuela: Implications for paleosalinity reconstructions, *Geochemistry, Geophys. Geosystems*, 10(6), doi:10.1029/2008GC002035, 2009.
- Mezger, E. M., de Nooijer, L. J., Boer, W., Brummer, G. J. A. and Reichart, G. J.: Salinity controls on Na incorporation in Red Sea planktonic foraminifera, *Paleoceanography*, 31(12), 1562–1582, doi:10.1002/2016PA003052, 2016.
- Mezger, E. M., De Nooijer, L. J., Bertlich, J., Bijma, J., Nürnberg, D. and Reichart, G.-J. J.: Planktonic foraminiferal spine versus shell carbonate Na incorporation in relation to salinity, *Biogeosciences*, 16(6), 1147–1165, doi:10.5194/bg-16-1147-2019, 2018a.
- Mezger, E. M., de Nooijer, L. J., Siccha, M., Brummer, G.-J. A., Kucera, M. and Reichart, G. J.: Taphonomic and Ontogenetic Effects on Na/Ca and Mg/Ca in Spinose Planktonic Foraminifera From the Red Sea, *Geochemistry, Geophys., Geosystems*, doi:10.1029/2018GC007852, 2018b.
- Mohtadi, M., Steinke, S., Groeneveld, J., Fink, H. G., Rixen, T., Hebbeln, D., Donner, B. and Herunadi, B.: Low-latitude control on seasonal and interannual changes in planktonic foraminiferal flux and shell geochemistry off south Java: A sediment trap study, *Paleoceanography*, 24(1), doi:10.1029/2008PA001636, 2009.
- Morard, R., Füllberg, A., Brummer, G.-J. A., Greco, M., Jonkers, L., Wizemann, A., Weiner, A. K. M., Darling, K., Siccha, M., Ledevin, R., Kitazato, H., de Gardiel-Thoron, T., de Vargas, C., Kucera, M., De Gardiel-Thoron, T., de Vargas, C., Kucera, M., de Gardiel-Thoron, T., de Vargas, C. and Kucera, M.: Genetic and morphological divergence in the warm-water planktonic foraminifera genus *Globigerinoides*, *PLoS One*, 14(12), 1–30, doi:10.1371/journal.pone.0225246, 2019.
- Mulitza, S., Boltovskoy, D., Donner, B., Meggers, H., Paul, A. and Wefer, G.: Temperature:  $\delta^{18}\text{O}$  relationships of planktonic foraminifera collected from surface waters, *Palaeogeogr. Palaeoclimatol. Palaeoecol.*, 202(1–2), 143–152, doi:10.1016/S0031-0182(03)00633-3, 2003.
- de Nooijer, L. J., Spero, H. J., Erez, J., Bijma, J. and Reichart, G. J.: Biomineralization in perforate Foraminifera, *Earth-Science Rev.*, 135, 48–58, doi:10.1016/j.earscirev.2014.03.013, 2014a.
- de Nooijer, L. J., Hathorne, E. C., Reichart, G. J., Langer, G. and Bijma, J.: Variability in calcitic Mg/Ca and Sr/Ca ratios in clones of the benthic foraminifer *Ammonia tepida*, *Mar. Micropaleontol.*, 107, 32–43, doi:10.1016/j.marmicro.2014.02.002, 2014b.
- Pierre, C.: The oxygen and carbon isotope distribution in the Mediterranean water masses, *Mar. Geol.*, 153(1–4), 41–55, doi:10.1016/S0025-3227(98)00090-5, 1999.

Pierre, C., Vergnaud Grazzini, C., Thuoron, D. and Saliège, J.-F.: Compositions Isotopiques de L'Oxygène Et Du Carbone Des Masses D'Eau Méditerranée, Mem. Soc. Geol. It., 36, 165–174, 1986.

455 Rigual-Hernández, A. S., Sierro, F. J., Bárcena, M. A., Flores, J. A. and Heussner, S.: Seasonal and interannual changes of planktic foraminiferal fluxes in the Gulf of Lions (NW Mediterranean) and their implications for paleoceanographic studies: Two 12-year sediment trap records, Deep. Res. Part I Oceanogr. Res. Pap., 66, 26–40, doi:10.1016/j.dsr.2012.03.011, 2012.

Rohling, E. J.: Progress in paleosalinity: Overview and presentation of a new approach, Paleoceanography, 22(3), 1–9, doi:10.1029/2007PA001437, 2007.

460 Sadekov, A., Eggins, S. M., De Deckker, P. and Kroon, D.: Uncertainties in seawater thermometry deriving from intratest and intertest Mg/Ca variability in Globigerinoides ruber, Paleoceanography, 23(1), 1–12, doi:10.1029/2007PA001452, 2008.

Schouten, S., Ossebaar, J., Schreiber, K., Kienhuis, M. V. M., Langer, G., Benthien, A. and Bijma, J.: The effect of temperature, salinity and growth rate on the stable hydrogen isotopic composition of long chain alkenones produced by *Emiliana huxleyi* and *Gephyrocapsa oceanica*, Biogeosciences, 3(1), 113–119, doi:10.5194/bg-3-113-2006, 2006.

465 van Sebille, E., Scussolini, P., Durgadoo, J. V., Peeters, F. J. C., Biastoch, A., Weijer, W., Turney, C., Paris, C. B. and Zahn, R.: Ocean currents generate large footprints in marine palaeoclimate proxies, Nat. Commun., 6(March), doi:10.1038/ncomms7521, 2015.

Spero, H. J., Eggins, S. M., Russell, A. D., Vetter, L., Kilburn, M. R. and Hönisch, B.: Timing and mechanism for intratest Mg/Ca variability in a living planktic foraminifer, Earth Planet. Sci. Lett., 409, 32–42, doi:10.1016/j.epsl.2014.10.030, 2015.

470 Stahl, W. and Rinow, U.: Sauerstoffisotopenanalysen an Mittelmeerwässern: Ein Beitrag zur Problematik von Paläotemperaturbestimmungen, "Meteor" Forschungsergebnisse. R. C Geol. und Geophys., 14, 55–59, 1973.

Vasiliev, I., Reichert, G. J. and Krijgsman, W.: Impact of the Messinian Salinity Crisis on Black Sea hydrology-Insights from hydrogen isotopes analysis on biomarkers, Earth Planet. Sci. Lett., 362, 272–282, doi:10.1016/j.epsl.2012.11.038, 2013.

475 Weiss, M. G., Pfannerstill, Y. E., Schouten, S., Sinninghe Damsté, S. J. and Van Der Meer, T. J. M.: Effects of alkalinity and salinity at low and high light intensity on hydrogen isotope fractionation of long-chain alkenones produced by *Emiliana huxleyi*, Biogeosciences, 14(24), 5693–5704, doi:10.5194/bg-14-5693-2017, 2017.

Wit, J. C., de Nooijer, L. J., Wolthers, M. and Reichert, G. J.: A novel salinity proxy based on Na incorporation into foraminiferal calcite, Biogeosciences, 10(10), 6375–6387, doi:10.5194/bg-10-6375-2013, 2013.

480

Formatted: German (Germany)

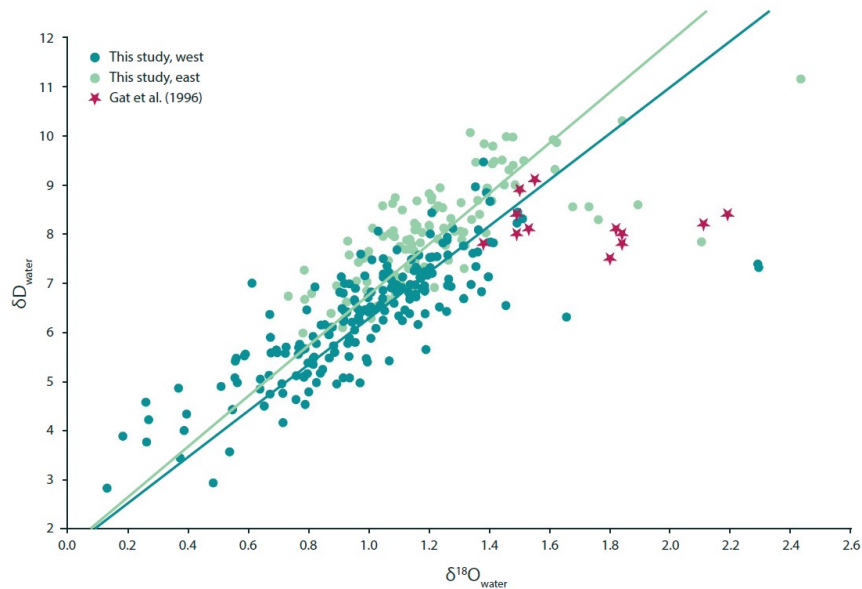


Figure 1: The  $\delta D$  of the Mediterranean surface sea water is positively correlated with the local  $\delta^{18}O$ . The orthogonal regression of the western Mediterranean can be described as  $\delta D_{\text{water}} = 4.8272 * \delta^{18}O_{\text{water}} + 1.67$  (dark green). The eastern Mediterranean is very similar to the western basin, the relationship between sea water  $\delta^{18}O$  and  $\delta D$  is  $\delta D_{\text{water}} = 5.19 * \delta^{18}O_{\text{water}} + 1.68$  (light green) here. Statistically they cannot be told apart. This was determined using a bootstrapping approach that generated 100 slopes and intercepts for both the eastern and the western dataset and subsequent t-testing using the mean and standard deviation of both groups of slopes and intercepts, which resulted in p-values  $> 0.05$ . In both areas the relationship is very different from the observations made by Gat et al. (1996), whose dataset suggested no statistically significant relationship between  $\delta D$  and  $\delta^{18}O$  of the sea water (p-value  $> 0.05$ ).



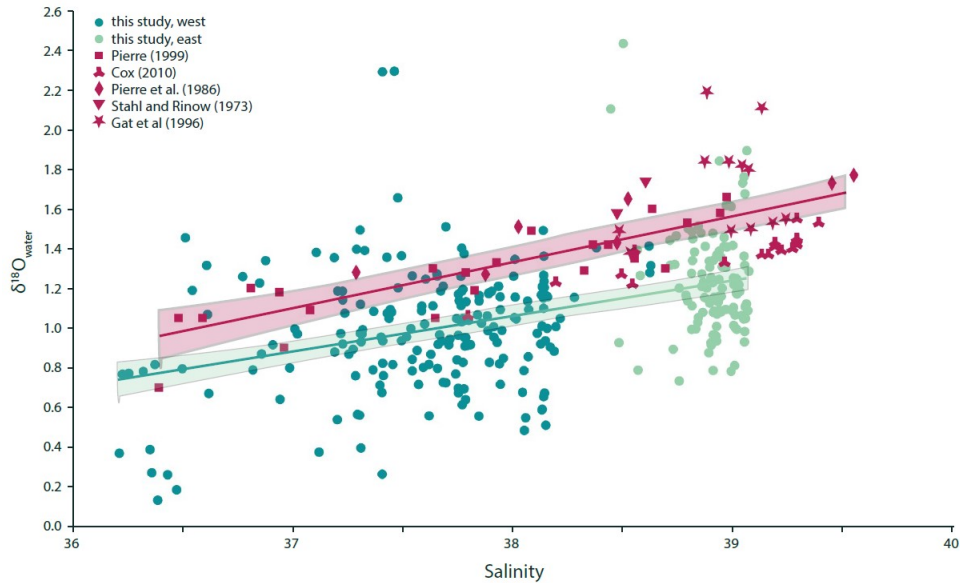


Figure 2: Surface sea water  $\delta^{18}\text{O}$  is positively correlated with sea surface salinity in the Mediterranean Sea, the relationship observed can be described as linear regression  $\delta^{18}\text{O}_{\text{water}}=0.17*S-5.39$  (p-value < 0.001, adjusted  $R^2=0.17$ ). Previously published data can be combined into one dataset with a similar relationship with a slightly steeper slope, that is offset towards relatively higher  $\delta^{18}\text{O}$  ( $\delta^{18}\text{O}_{\text{water}}=0.22*S-7.19$ ; p-value < 0.001, adjusted  $R^2=0.48$ ). The two regression lines are significantly different from each other (ANOVA p-value < 0.01).

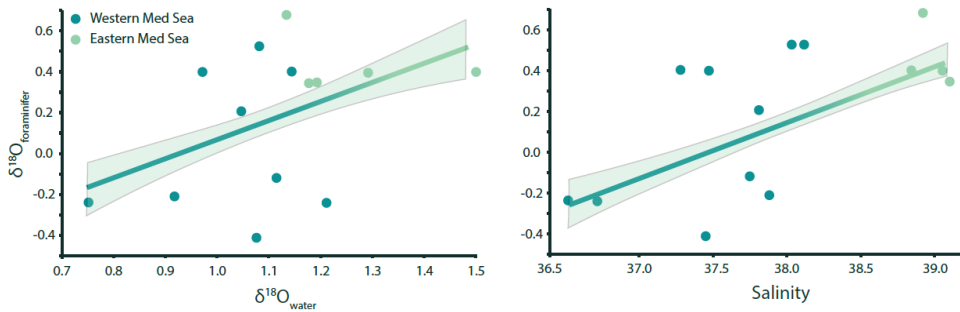
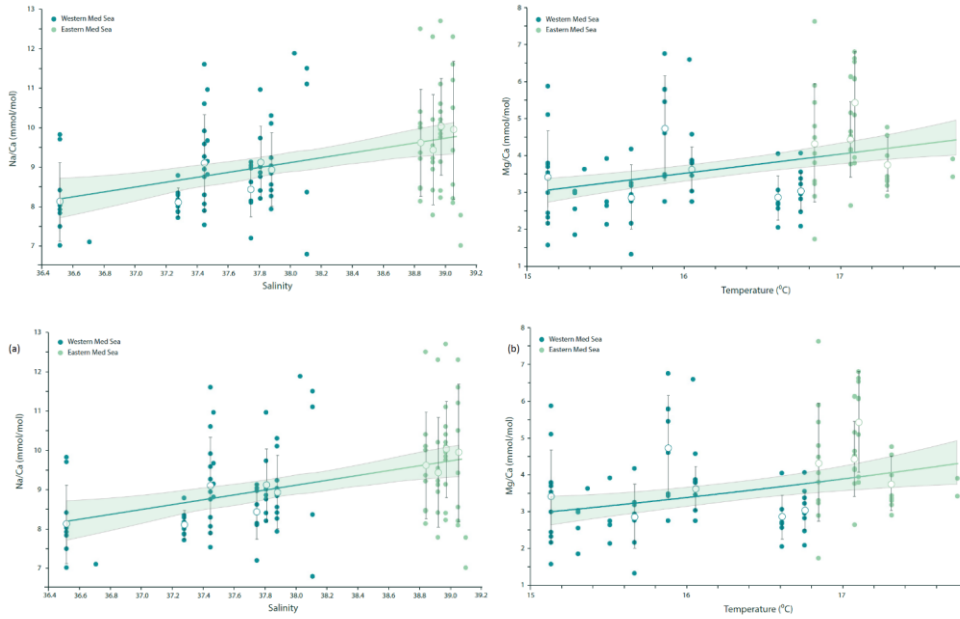


Figure 3: *G. ruber albus*  $\delta^{18}\text{O}$  measurements are positively correlated (p-value < 0.001) to both sea water  $\delta^{18}\text{O}$  (a) and salinity (b). The relationships can be described using the following equations:  $\delta^{18}\text{O}_{\text{foraminifera}}=0.28*S-10.59$  (adjusted  $R^2 = 0.42$ ) and  $\delta^{18}\text{O}_{\text{foraminifera}}$

Formatted: Font: Italic

$= -0.95 * \delta^{18}\text{O}_{\text{water}} - 0.89$  (adjusted  $R^2 = 0.24$ ). The relationship between  $\delta^{18}\text{O}_{\text{water}}$  and salinity in this subset of samples is linear and comparable to that of the entire dataset ( $\delta^{18}\text{O}_{\text{water}} = 0.13 * S - 3.91$ ;  $p$ -value  $< 0.05$ ,  $R^2 = 0.37$ ).

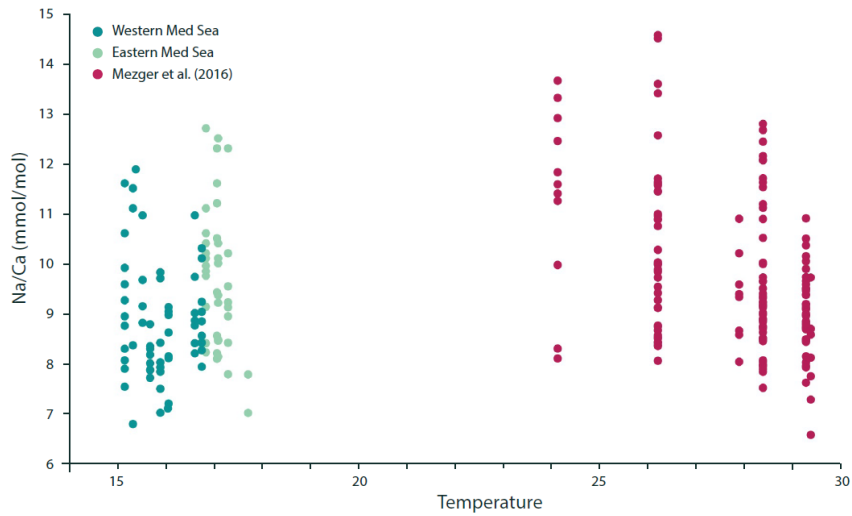


505 Figure 4: (a) Na/Ca measured in *G. ruber albus* F-chambers collected as living specimens from the eastern and western Mediterranean Sea correlates well with local salinity ( $p$ -value  $< 0.001$ ,  $\text{Na/Ca} = 0.66 - 60 * S - 16.07 [3.84]$ ), even though a large natural spread of elemental composition around the mean values per station exists ( $R^2 = 0.13$ ). For salinities with more than 5 individual Na/Ca measurements, hollow circles with whiskers indicate average values and standard deviations. (b) Mg/Ca in F-chambers of *G. ruber albus* specimens collected from the water column of the Mediterranean Sea is positively correlated with sea surface temperature and can be described with the exponential relationship  $\text{Mg/Ca} = 0.37 * \exp(0.14 * T)$ , linear relationship  $\text{Mg/Ca} = 0.47 * T - 3.98$  ( $p$ -value  $< 0.05$ ). For temperatures with more than 5 individual Mg/Ca measurements, hollow circles with whiskers indicate average values and standard deviations. Regression lines were calculated using all individual data points.

510

Formatted: Font: Italic

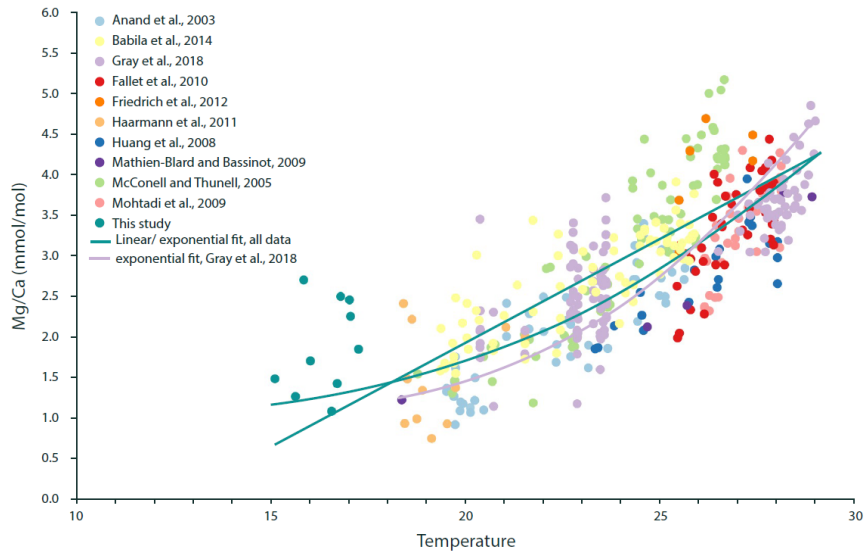
Formatted: Font: Italic



[515] Figure 5: The ratio of Na/Ca in *G. ruber albus* appears to be independent from sea water temperature. While Mezger et al. (2016) showed a negative relationship between temperature and foraminiferal Na/Ca in specimens collected from the Red Sea, the addition of new data from the Mediterranean Sea shows clearly that the previously hypothesized negative impact of temperature on Na/Ca is likely an artefact of the negative relationship of temperature and salinity in the Red Sea and that temperature has no significant impact on Na/Ca.

Formatted: Font: Italic

520



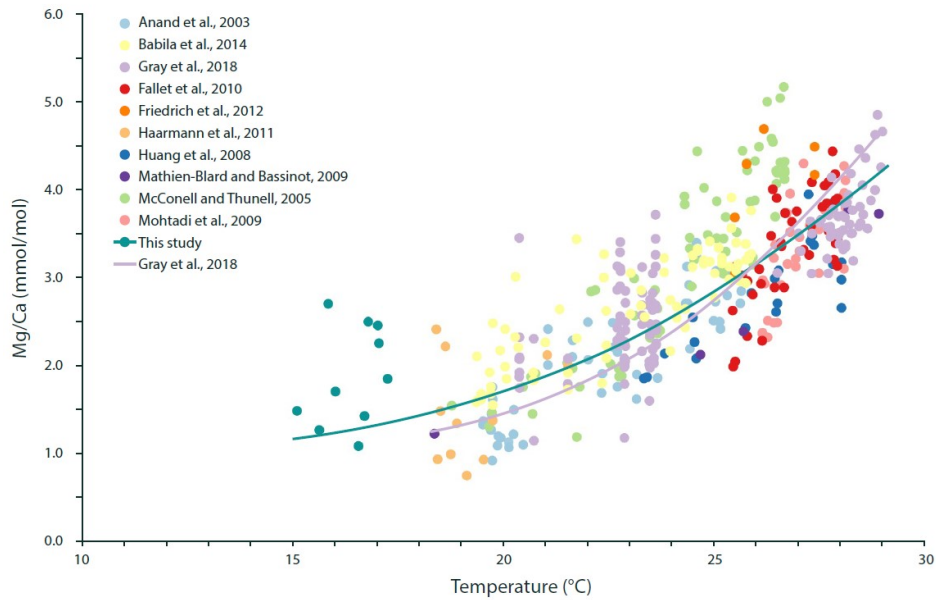


Figure 6: The relationship between Mg/Ca in *G. ruber albus* and temperature during calcification can be described using the following exponential equation:  $Mg/Ca=0.278*\exp(0.093*T)$  for a temperature range from 15.1 to 29.1°C.

Formatted: Font: Italic

525

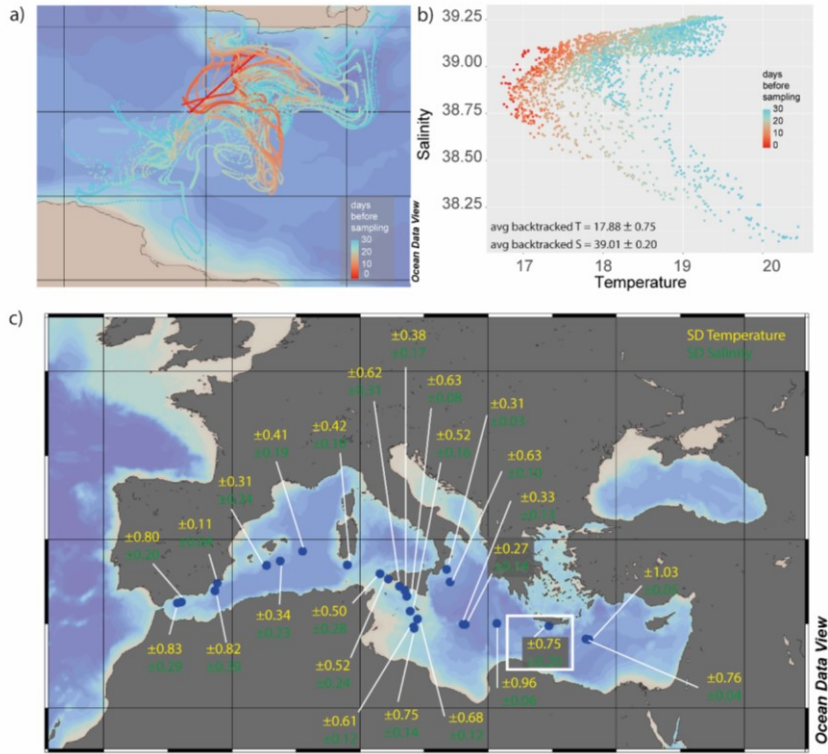
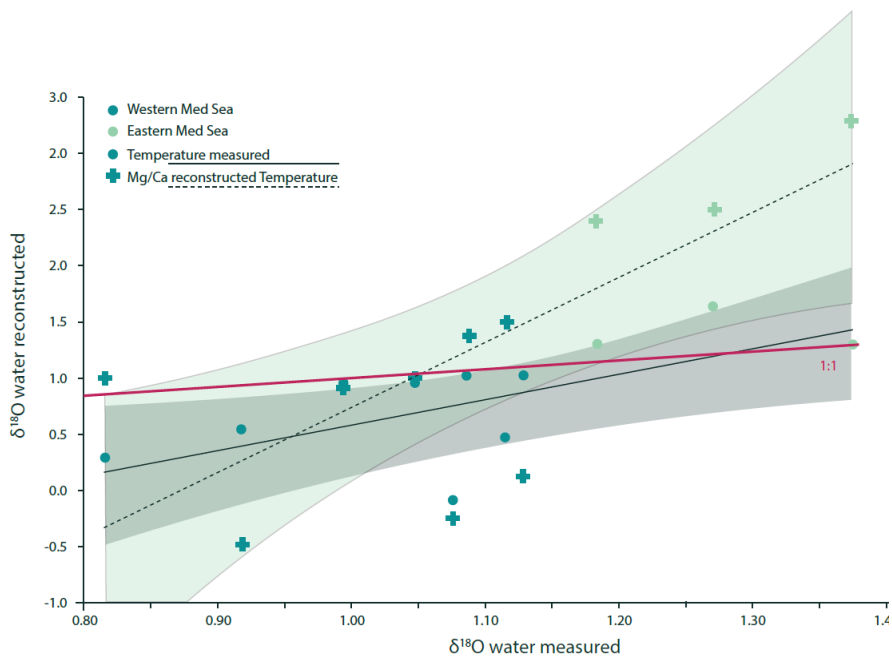


Figure 7: (a) Example of back-tracked pathways for a single transect (the one marked by a white rectangle in panel c). The colour indicates the time before sampling up to 30 days. (b) Analysing the different environmental conditions at the different locations of these potential paths show that foraminifera sampled very likely experienced a large range in temperatures as well as salinities. (c) The variability in potentially experienced environmental conditions varies considerably from location to location, as indicated by notation of 1 standard deviation for both parameters for each sampling location. Maps in (a) and (c) were generated using Ocean Data View version 4.

530



535 Figure 8: The relationship between  $\delta^{18}\text{O}_{\text{seawater}}$  measured in the Mediterranean Sea and  $\delta^{18}\text{O}_{\text{seawater}}$  calculated from foraminiferal  
 geochemistry (*G. ruber albus-white*). The relationship shown with dashed lines and cross shaped markers represents values calculated  
 using foraminiferal  $\delta^{18}\text{O}$  as well as Mg/Ca as an additional temperature proxy to decouple the effect of temperature and salinity on  
 $\delta^{18}\text{O}$ . ~~This relationship is non-significant (p-value > 0.05).~~ The relationship shown with the continuous lines and circular markers  
 540 shows the same samples, but instead of using temperature values derived from foraminiferal Mg/Ca ratios, in situ measurements  
 for temperature were used, the relationship can be described as  $\delta^{18}\text{O}_{\text{water, reconstructed}} = 2.62(\pm 0.69) * \delta^{18}\text{O}_{\text{water, measured}} - 63.99(\pm 26.11)$  with  
 an adjusted  $R^2$  of 0.37. The temperature gradient was 2.2°C.

545

#	Comments Reviewer 1	Our replies	Changes made in manuscript
1.01	It would be effective to insert in the Discussion section a short paragraph with “recommendation for the applications of the proxies” that the authors (based on this study) consider relevant for the paleo-reconstructions (i.e., using more specimens for the analyses, uncertainty in salinity estimates?, how collect the samples, etc. . .).	We have extended section “4.4.3 Implications for proxies” and added the suggested topics to our discussion.	Added: “It is therefore crucial to choose temperature proxies carefully, use a large enough number of specimens for analysis, be aware about potential effects of lateral particle transport as well as other environmental parameters, and to be conscious about how errors propagate in paleoclimate reconstructions” to section 4.4.3
1.02	In fig 2 ( $\delta^{18}\text{O}$ seawater versus salinity) all data are from Mediterranean sea except data from Cox (2010) that are from North Atlantic. I do not understand why the authors use the North Atlantic data, otherwise the authors can discuss this in paragraph 4.1 (when they report geographical variability, lines 173,174).	We agree that mixing data from different basins should be avoided for this analysis. While the publication title for the reference “Cox (2010)” is indeed “ <i>Stable Isotopes as Tracers for Freshwater Fluxes into the North Atlantic</i> ”, Katharine A. Cox’ publication does not only contain North Atlantic sea water isotope data, but also a number of Mediterranean Sea water isotope measurements which are presented in her Appendix D.2 “ <i>Oxygen and Hydrogen Isotope Data analyzed at UC Davis: Table D.4: Station locations, depths, salinity data and isotopic parameters of the 2001 M51–3 water samples from Mediterranean, the 2004 JR106b water samples from Kangerdlussuaq Fjord, Denmark Strait and the 2005 D298 and 2008 D332 water samples from Cape Farewell</i> ”, p. 138ff. From this data set we selected surface water $\delta^{18}\text{O}$ and $\delta\text{D}$ measurements from the Mediterranean Sea samples M51-3 to include in our analysis and Fig. 2 to compare with our own Mediterranean Sea data	No changes made.
1.03	3.1 I suggest as title: seawater geochemistry or	We have adjusted the title of section 3.1 to “Mediterranean	Changed title of section 3.1 to “Mediterranean Sea geochemistry”



	Mediterranean Sea geochemistry	Sea geochemistry” following this suggestion.	
	4.2 The analyses were performed on Na/Ca ratios measured on the carbonate shells of <i>G. ruber</i> . It is <i>G. ruber</i> (white) as reported in the paragraph 3.3?	Correct, all analysis was performed on the same species, as also described in section “2 Materials and Methods”. We have updated the species name throughout the manuscript to <i>Globigerinoides ruber albus</i> to reflect the recent suggestions presented by Morard et al. (2019) and to avoid any ambiguity.	Updated the species name throughout the manuscript to <i>Globigerinoides ruber albus</i> .
1.04	The same observation for paragraph 4.3	See previous reply	-
1.05	Line 191- Fig. 4a (add a point)	We have added the missing point	Point added
1.06	Line 226 <i>G.ruber</i> in italic font	We have changed the font to italic.	Font changed
1.07	Fig. 3, 4, 5: <i>G. ruber</i> in italic font	We have changed the font to italic.	Font changed
	Comments Reviewer 2	Our replies	Changes made in manuscript
2.01	The authors present a comprehensive regional dataset of trace element and stable oxygen isotope data measured on foraminifera collected from plankton samples with rich contextual physical and chemical data. The analyses were carried out to test to what degree the strong salinity (or seawater oxygen isotope) gradient in the Mediterranean could have been reconstructed from shell chemistry. The results are sobering, which I believe is not to be taken negatively but as an extremely important result, confirming the growing body of evidence that there is something we fundamentally do not understand about the way the proxy signal in the sediment is generated. In this way, the manuscript makes an important fresh contribution to the field and the data and analyses in my opinion	<p>We will rephrase parts of the discussion to put less emphasis on the impact of lateral transport, following the suggestions of the reviewer.</p> <p>We will include explanations concerning the carbonate ion effect in the next version of the manuscript.</p> <p>We have included a comment about the possibility that specimens of <i>G. elongatus</i> might have incorrectly be identified as <i>G. ruber albus</i> in section 2 Materials and Methods, as described in our reply to a later comment. We have found no bimodality in the single shell LAQICPMS measurements of our specimens and therefore assume that the effect of species misidentification is minimal.</p>	<p>We combined 4.4.1 and 4.4.2 and rephrased the heading to reflect the decrease in emphasis on the particle backtracking results. We also shortened the text and instead now focus on the most important consequence of this exercise.</p> <p>Added “The Mg/Ca values used here were not corrected for salinity or effects, since salinity is the target parameter that has to be reconstructed and is thus treated as unknown. Even though there is a carbonate ion effect on the Mg/Ca in <i>G. ruber albus</i> (Evans et al., 2016; Gray et al., 2018; Kisakürek et al., 2008), the measured values were not corrected for this, since this factor is also unknown in paleo-reconstructions.” To section 4.4.3</p> <p>“Specimens used for analyses were selected from the size fraction 150 - 250 µm, even though it has been reported that</p>

	<p>warrant publication in <i>Climate of the Past</i>.</p> <p>That said, I would advise the authors to place less emphasis on the aspect of lateral advections, for reasons explained below, and to provide a more explicit quantitative evaluation of the magnitude and direction of the various candidate processes invoked to explain the large scatter. Beyond the individual comments listed below, I would like the authors to explain if/how they dealt with the carbonate ion effect on all of the proxies (oxygen isotopes and Mg/Ca in particular), as this is not really clear from the text and I would like to draw to their attention the possibility that the analyses of the <i>G. ruber</i> from the plankton in the chosen size fraction could have been affected by differential contribution of specimens representing pre-adult <i>G. elongatus</i>, which may follow a different calibration line. Perhaps the results already contain some hints (bimodality or not of the single-shell measurements, for example)?</p>		<p>at this size fraction <i>G. ruber albus</i> and <i>Globigerinoides elongatus</i> cannot always be confidently distinguished due to similar morphology (Aurahs et al., 2011)“</p>
2.02	<p>Finally, I would like to urge the authors to make sure that the data that will be made available on the Utrecht data server are as comprehensive as possible and that they are stored in a way that they will be found in any future attempts to synthesize seawater or foraminifera chemistry data.</p>	<p>All the environmental and geochemical data from this project will be available for download at the 4TU.Centre for Research Data where we are certain it can easily be found and accessed by anyone</p>	<p>No changes made.</p>
2.03	<p>Taken together, these points and the individual points below all aim to make the most out of the nice dataset that the authors have, which I believe they will be able to do</p>	<p>We have changed the manuscript title according to this suggestion into “Evaluation of oxygen isotopes and trace elements in planktonic foraminifera from</p>	<p>Changed title of the manuscript to “Evaluation of oxygen isotopes and trace elements in planktonic foraminifera from the Mediterranean Sea as recorders</p>

	without having to substantially restructure the paper or change its conclusions. Comments to individuals points: Title: Instead of “isotopes and elements”, I would recommend to be either more specific (oxygen isotopes and trace elements) or less specific (shell geochemistry), or else the title appears to promise more than what is delivered.	the Mediterranean Sea as recorders of seawater oxygen isotopes and salinity”	of seawater oxygen isotopes and salinity”
2.04	Line 30: large scale	We have corrected this typo.	Changed from “largescale” to “large scale”
2.05	Line 54: continues to be	We have corrected this typo.	Changed from “is continued to be” to “continues to be”
2.06	Line 66: please specify what exactly “has been shown for foraminifera”. In my opinion the effect of expatriation on shell chemistry in foraminifera has been previously shown by the work of Ganssen and Kroon in the Red Sea, but not really outside of that extreme environment. The studies cited in this place were mainly concerned with attempts to use particle tracking in models and describe potential effects, rather than documenting these effects empirically, or the empirical detection was indirect, inferred from sediment trap material where the dwelling depth is unknown.	We agree with this comment and thus have changed the manuscript text from “has been shown” to “has been suggested” to account for the fact that the quoted studies are not empirical. We also added the suggested reference Ganssen and Kroon (1991) to the Introduction.	Changed to “Recently this has been shown suggested for dinoflagellate cysts (Nootboom et al., 2019) and planktonic foraminifera, collected from the water column (Ganssen and Kroon, 1991), from sediment (van Sebille et al., 2015) and also from sediment traps (Steinhardt et al., 2014), but can also be applied to specimens collected living from the sea surface.”
2.07	Line 87: there are no formally and objectively defined and biologically or ecologically meaningful morphotypes within the species <i>G. ruber</i> . The concept of “morphotypes”, re-introduced into the literature by Wang, has been superseded by the discovery based on genetic data (Aurahs et al., 2011), that the species concept as introduced by Parker (1965) is incorrect and that the species	We have adjusted this paragraph by using the correct taxon label <i>G. ruber albus</i> , including the reference Morard et al. (2019) and exclude the mention of morphotypes (see next comment for updated version of this line). We also updated the species name to <i>G. ruber albus</i> throughout the rest of the manuscript.	Changed to “A variety of samples containing specimens of <i>G. ruber albus</i> (Morard et al., 2019) was selected to cover a large range in salinities and temperatures. Specimens used for analyses were selected from the size fraction 150 - 250 µm, even though it has been reported that at this size fraction <i>G. ruber albus</i> and <i>Globigerinoides elongatus</i> cannot always be confidently distinguished due to similar

	<p>G. elongatus, synonymised by her with G. ruber, should have been retained. The same genetic data have also revealed that the pink and white varieties of G. ruber are genetically distinct and these have been now formally distinguished at the level of subspecies. The correct label of the analysed taxon is thus Globigerinoides ruber albus (Morard et al., 2020), with morphology corresponding to what Kontakiotis et al. (2017) label as Morphotype A.</p>		<p>morphology (Aurahs et al., 2011).“</p>
2.08	<p>Line 88: I fully understand the decision to concentrate on the relatively small size fraction for analyses, as this likely yielded most material. However, I would like to point out that Aurahs et al. (2011), also working with plankton material, also from the Mediterranean, showed that the features distinguishing G. ruber albus from G. elongatus are not yet present among all specimens in the plankton, allowing separation of plankton-derived specimens to the ruber and elongatus only to about 75 % accuracy. Since G. elongatus is abundant (if not dominant) in the Mediterranean, the authors must consider the possibility that some of the analysed specimens may have belonged to that species.</p>	<p>The decision to take specimens from size fraction 150-250µm was indeed made after an initial examination of size fraction &gt;250µm yielded insufficient material for analysis. We would like to thank the referee for bringing the Aurahs et al. (2011) study to our attention. After reading it we decided to include a comment about this in section 2 Materials and Methods, it now reads:  <i>“A variety of samples containing specimens of G. ruber albus (Morard et al. 2019) was selected to cover a large range in salinities and temperatures. Specimens used for analyses were selected from the size fraction 150 - 250 µm, even though it has been reported that at this size fraction G. ruber albus and Globigerinoides elongatus cannot always be confidently distinguished due to similar morphology (Aurahs et al., 2011).”</i></p>	<p>Changed to: <i>“A variety of samples containing specimens of G. ruber albus (Morard et al., 2019) was selected to cover a large range in salinities and temperatures. Specimens used for analyses were selected from the size fraction 150 - 250 µm, even though it has been reported that at this size fraction G. ruber albus and Globigerinoides elongatus cannot always be confidently distinguished due to similar morphology (Aurahs et al., 2011)“</i></p>
2.09	<p>Line 104: The methods section here is not entirely clear in how the oxygen isotopes were measured. Whereas it is clear that Mg/Ca was determined on final chambers of individual</p>	<p>We agree that the previous version of this section was not sufficiently clear on this point. We have adjusted this paragraph by including the</p>	<p>Changed to <i>“Stable oxygen and carbon isotopes of foraminiferal calcite were measured on groups of whole specimens different from those used for LA-Q-ICP-MS, using an automated carbonate</i></p>

	shells, the authors should specify if the isotopes were also measured on final chambers or whole shells, on single shells or multiple shells (and then how many) and whether the same shells as for Mg/Ca were used or different shells. This has all implications for the understanding of the origin of the apparent noise in the measurements.	previously missing information: <i>“Stable oxygen and carbon isotopes of foraminiferal calcite were measured on groups of whole specimens different from those used for LA-Q-ICP-MS, using an automated carbonate device (Thermo Kiel IV) which was connected to Thermo Finnigan MAT 253 Dual Inlet Isotope Ratio Mass Spectrometer (IRMS).”</i>	device (Thermo Kiel IV) which was connected to Thermo Finnigan MAT 253 Dual Inlet Isotope Ratio Mass Spectrometer (IRMS)”
2.10	Figure 1: I agree that the two regressions (correctly using a total least squares approach) are similar, but could the authors please provide a formal statistical test for the similarity of the slopes, to support their statement that the sensitivities are indistinguishable, and for the equality of the intercepts, to dispel the impression that the regression lines are offset, indicating different endmember composition? Also, I am not convinced that it is correct to consider the results of Gat et al. (1996) as being different, as all of their values fall within the range of the presented data.	We agree that a formal statistical test for the similarity of the regressions shown in figure 1 would greatly support our argument and will therefore include the results of this test in the manuscript. We would like to thank the reviewer for this suggestion. We understand that both datasets (Gat et al. (1996) and the data presented in our manuscript) could be considered not different from each other since ours includes “Gat-type” data points, too, we have now specified better what we meant by “different” in the original version of the manuscript. This part of the figure description of figure 1 now reads: <i>“In both areas the relationship is different from the observations made by Gat et al. (1996), whose dataset suggested no statistically significant relationship between <math>\delta D</math> and <math>\delta^{18}O</math> of the sea water (<math>p</math>-value &gt; 0.05).”</i>	Changed caption of Fig 1 to “The $\delta D$ of the Mediterranean surface sea water is positively correlated with the local $\delta^{18}O$ . The orthogonal regression of the western Mediterranean can be described as $\delta D_{\text{water}} = 4.72 * \delta^{18}O_{\text{water}} + 1.67$ (dark green). The eastern Mediterranean is very similar to the western basin, the relationship between sea water $\delta^{18}O$ and $\delta D$ is $\delta D_{\text{water}} = 5.19 * \delta^{18}O_{\text{water}} + 1.68$ (light green) here. Statistically they cannot be told apart. This was determined using a bootstrapping approach that generated 100 slopes and intercepts for both the eastern and the western dataset and subsequent t-testing using the mean and standard deviation of both groups of slopes and intercepts, which resulted in p-values > 0.05. In both areas the relationship is very different from the observations made by Gat et al. (1996), whose dataset suggested no statistically significant relationship between $\delta D$ and $\delta^{18}O$ of the sea water ( $p$ -value > 0.05). “
2.11	Figure 2: Could the authors please state which regression has been used here and also provide a formal test for the lack of difference in the east and west and for the presence of a difference in the slope and intercept between their	We used ordinary least squares regressions for figure 2, assuming a linear response model. We have added the missing R <sub>2</sub> -values to the figure description. The adjusted R <sub>2</sub> is 0.48 for the regression based on previously published data,	Changed caption of Fig 2 to “Surface sea water $\delta^{18}O$ is positively correlated with sea surface salinity in the Mediterranean Sea, the relationship observed can be described as linear regression $\delta^{18}O_{\text{water}} = 0.17 * S - 5.39$ ( $p$ -value <

	<p>data and literature data? Also please provide R2 for all regressions in the figure caption and/or text.</p>	<p>and 0.17 for the regression based on our data. When analyzing the sampling locations (east and west) separately, the regression for the eastern samples is statistically insignificant (p-value &gt; 0.05), likely due to the large amount of scatter and small range in salinity values compared to the overall dataset. The western part of the dataset can be described as <math>\delta^{18}\text{O}_{\text{water}}=0.15*S-4.75</math> (p-value &lt; 0.001, adjusted <math>R^2=0.06</math>). The results of a one-way ANOVA show that these two subsets of the data set are significantly different (p-value &lt; 0.05), we still decided to combine them in this case, since the range of salinities is very limited for the eastern part. We have now included these information in the caption of figure 2. We will perform a one-way ANOVA to show the difference in slope and intercept between our data and literature data, and include the result in the caption of figure 2.</p>	<p>0.001, adjusted <math>R^2=0.17</math>). Previously published data can be combined into one dataset with a similar relationship with a slightly steeper slope, that is offset towards relatively higher <math>\delta^{18}\text{O}</math> (<math>\delta^{18}\text{O}_{\text{water}}=0.22*S-7.19</math>; p-value &lt; 0.001, adjusted <math>R^2=0.48</math>). The two regression lines are significantly different from each other (ANOVA p-value &lt; 0.01).”</p>
2.12	<p>Line 145: Considering that seawater oxygen isotopes and salinity only correlated with R2 od 0.2, the authors need an explanation for what the isotopes in foraminifera correlated more strongly with both variables. Could it be that each of the variables explains a different part of the total variance? Then, a multiple regression of foraminifera isotopes against seawater isotopes an salinity should explain significantly more variance. If it does not, it means that the two explanatory variables explain the same amount of variance. This could be because of a</p>	<p>We have now calculated the adjusted <math>R^2</math> values for the regressions shown in Figure 3. They are 0.24 for <math>\delta^{18}\text{O}_{\text{foraminifera}}</math> vs <math>\delta^{18}\text{O}_{\text{water}}</math> and 0.42 for <math>\delta^{18}\text{O}_{\text{foraminifera}}</math> vs salinity, these are now stated in the figure caption. We will include a paragraph discussing these in the manuscript, as well as an adjusted <math>R^2</math> value for the relationship between sea water salinity and <math>\delta^{18}\text{O}</math> from the subset of water samples used for the calculations shown in figure 3, following the reviewer’s suggestions.</p>	<p>Changed caption of Fig 3 to “<i>G. ruber albus</i> <math>\delta^{18}\text{O}</math> measurements are positively correlated (p-value &lt; 0.001) to both sea water <math>\delta^{18}\text{O}</math> (a) and salinity (b). The relationships can be described using the following equations: <math>\delta^{18}\text{O}_{\text{foraminifera}}=0.28*S-10.59</math> (adjusted <math>R^2 = 0.42</math>) and <math>\delta^{18}\text{O}_{\text{foraminifera}} =0.95* \delta^{18}\text{O}_{\text{water}}-0.89</math> (adjusted <math>R^2 = 0.24</math>). The relationship between <math>\delta^{18}\text{O}_{\text{water}}</math> and salinity in this subset of samples is linear and comparable to that of the entire dataset (<math>\delta^{18}\text{O}_{\text{water}} = 0.13*S - 3.91</math>; p-value &lt; 0.05, <math>R^2 = 0.37</math>). ”</p>

	<p>fortuitous choice of sampling and the authors should thus also calculate the R2 for salinity and seawater isotopes only for the samples shown in Figure 3.</p>		
2.13	<p>Figure 4: Could the authors again specify what regression has been used and how exactly the regression lines were calculated (regression of individual values or of the means)? Please state R2 for all regressions. Also, the Mg/Ca to T relationship is known to be exponential, so why not fitting an exponential curve? The linearity of the relationship could simply reflect the fact that the regression is fitted over a relatively narrow temperature range.</p>	<p>For figure 4 we used ordinary least squares regressions and included all individual data points instead of using just the mean values. The adjusted R<sub>2</sub> values of 0.13 for the salinity to Na/Ca calibration, and 0.07 for the relationship between Mg/Ca and temperature, due to the large scatter. We originally chose to use a linear regression in this case even though we are aware that an exponential curve is probably more correct, as also used for figure 6. We will fit an exponential model to this data and will update the figure caption as well as the text accordingly.</p>	<p>Fig 4 has been replaced, the new version shows an exponential instead of a linear fit.</p> <p>Caption of Fig 4 changed to “(a) Na/Ca measured in <i>G. ruber albus</i> F-chambers collected as living specimens from the eastern and western Mediterranean Sea correlates well with local salinity (p-value &lt; 0.001, Na/Ca=0.60 * S-13.84), even though a large natural spread of elemental composition around the mean values per station exists (R<sup>2</sup>=0.13). For salinities with more than 5 individual Na/Ca measurements, hollow circles with whiskers indicate average values and standard deviations. (b) Mg/Ca in F-chambers of <i>G. ruber albus</i> specimens collected from the water column of the Mediterranean Sea is positively correlated with sea surface temperature and can be described with the exponential relationship Mg/Ca = 0.37*exp(0.14*T). For temperatures with more than 5 individual Mg/Ca measurements, hollow circles with whiskers indicate average values and standard deviations. Regression lines were calculated using all individual data points.”</p>
2.14	<p>Line 155: Considering that Mg/Ca is also changing as a function of salinity, why not plotting Mg/Ca against salinity and analyzing the strength of that relationship as well?</p>	<p>Since temperature and salinity are co-varying strongly in the Mediterranean Sea, any relationship between Mg/Ca and salinity obtained from such analysis would be heavily influenced by temperature and thus appear stronger than it actually is.</p>	<p>No changes made.</p>
2.15	<p>Line 160: it is true that the foraminifera may have</p>	<p>We agree with all points brought up here, the</p>	<p>No changes made.</p>

	<p>travelled a long distance over the 30 days of the simulation, but I question the significance of the so derived variability for the interpretation of the shell geochemistry. Culturing observations indicate that <i>G. ruber</i> in the size range as analysed here produces a new chamber about every two days. Thus, the particle tracking result has no bearing on the laser-ablation data. For the isotope data, if we assume a total lifespan of 4 weeks and a life expectancy of the specimens in the analysed size range of two weeks, then the collected specimens would have only had two weeks to grow, not 30 days. On top of that, because of the exponential growth of the shell, almost all of the analysed calcite and thus almost all of the isotopic signal is present in the last few chambers of the shell, so in reality, the backtracking relevant to the analysed signal should not have been carried back for more than a week. This is not to say that the result stated here is wrong – it is just that the result is not relevant for the interpretation of the measured geochemical signals. I note that your discussion in 4.4.1 resonates well with what I write, but then I do not really understand what was the merit or the justification of showing the particle backtracking results in figure 7 over 30 days?</p>	<p>geochemical imprint of environmental parameters experienced longer ago have little impact on the bulk shell geochemistry compared to more recently experienced conditions, due to the strong increase in chamber size during foraminiferal growth. We still decided to show the trajectory for the full 30 days, which might exceed these specific specimens' lifetimes, as a worst case scenario that could be transferred to studies using larger specimens from sediment, for example. We also think that while the impact is small, it still contributes to the overall puzzle and needs to be addressed. The more factors and impacts can be quantified (and even better if they turn out to be small!), the more we become aware of the limitations of paleoclimate proxies and can make informed decisions on whether they can be applied confidently or not.</p>	
2.16	<p>Line 170 and onwards: please see the comments above as to the necessity to provide statistical tests to support the presence or absence of differences in regression shapes. Also, please consider</p>	<p>We agree that statistical tests are needed to conclusively show the difference between the data presented by us and the literature data. We will report the results of these tests in the caption of figure 2</p>	<p>Changed caption of Fig 2 to "Surface sea water <math>\delta^{18}O</math> is positively correlated with sea surface salinity in the Mediterranean Sea, the relationship observed can be described as linear regression</p>



	<p>the location of the sampling by Gat and yours: what if the apparent offset from your regression that he reports simply reflects the fact that he sampled at locations where the relationship is unusually confounded by secondary variables and that your data would detect the same if you only had measurements at those locations? I am also concerned by the origin of the lower oxygen isotope values measured for the given salinity in your data: was the sampling method comparable between your data and those of the previous studies (collecting from the same depth)?</p>	<p>and include a statement about it in the Discussion section 4.1</p> <p>The majority of sampling locations used by Gat et al (1996) are nearby sampling locations used during our cruises. We have though considered that spatial differences in the sampling campaigns could contribute to the observed differences and have mentioned this in the discussion now. The updated section reads:</p> <p><i>“Potentially the observations of Gat et al. (1996) were hence either related to unusual conditions, spatially restricted features not covered by our sampling locations or the hydrological cycle in the eastern Mediterranean has recently changed considerably”</i></p> <p>We have carefully selected data presented in previous publications to only reflect surface waters to ensure comparability since we had also sampled at 5m water depth. Therefore different water depths do not play a role in the differences observed in the data.</p>	<p><math>\delta^{18}\text{O}_{\text{water}}=0.17*S-5.39</math> (p-value &lt; 0.001, adjusted <math>R^2=0.17</math>).</p> <p>Previously published data can be combined into one dataset with a similar relationship with a slightly steeper slope, that is offset towards relatively higher <math>\delta^{18}\text{O}</math> (<math>\delta^{18}\text{O}_{\text{water}}=0.22*S-7.19</math>; p-value &lt; 0.001, adjusted <math>R^2=0.48</math>). The two regression lines are significantly different from each other (ANOVA p-value &lt; 0.01).”</p> <p>Changed text to “Potentially the observations of Gat et al. (1996) were hence either related to unusual conditions, spatially restricted features not covered by our sampling locations or the hydrological cycle in the eastern Mediterranean has recently changed considerably”</p>
2.17	<p>Line 202 and onwards: Considering all your specimens were collected from the surface and that you measured only the composition of the final chamber, would it not be logically at this place to reject some of the hypotheses that you list here? Otherwise, you would have to imply that the specimens migrate vertically tens of meters over a few days, or stay alive without adding new chambers for weeks to allow lateral transport to have an effect. So perhaps we are left with the variable biomineralisation as</p>	<p>We agree and have changed the end of this section to address these comments. It now reads:</p> <p><i>“Since specimens used here were collected from surface waters and add new chambers very frequently, vertical migration into water depths with significantly different conditions as suggested by Mezger et al. (2018) and Van Sebille et al. (2015) appears to be an unlikely cause for heterogeneity between specimens in this case.”</i></p>	<p>Added “Since specimens used here were collected from surface waters and add new chambers very frequently, vertical or literal migration into waters with significantly different conditions as suggested by Mezger et al. (2018) and Van Sebille et al. (2015) appears to be an unlikely cause for heterogeneity between specimens in this case.” to section 4.2</p>

	the only remaining candidate mechanism?		
2.18	<p>Line 216: I fear the Mg/Ca data are revealing more than what the authors imply. Firstly, since the authors have both temperature and salinity, they should derive the correction independently of Gray et al. (2018) or at least check if the relationship they obtain holds. Second, I wonder why the authors do not discuss the fact that once the salinity effect is removed, their Mg/Ca data are no longer correlated with temperature or if correlated then with a much steeper slope (at least this is what I see looking at Figure 8). Third, I do not agree with the statement that the corrected values are slightly higher than expected based on the global regression – I observe that they are all higher than predicted by the exponential regression (the linear regression in Figure 8 is in my opinion superfluous). Why is that? Could there be a salinity-temperature interaction affecting the salinity-Mg/Ca relationship? This is an important result that deserves some more thought.</p>	<p>Since temperature and salinity are very strongly, positively correlated in the Mediterranean Sea, it is unfortunately not possible to use our data set to disentangle the effect of salinity on foraminiferal Mg/Ca from other factors such as temperature, and we therefore chose to correct for this using the equation published by Gray et al. (2018), which also increases comparability with their data set. It is indeed possible that the effect of salinity is currently underestimated, which would explain why our Mg/Ca values appear to be fairly high. We have expanded our discussion to include this consideration in our manuscript. We also have removed the word “slightly” from our description of the low temperature values. This section now reads: <i>“After normalizing Mg/Ca values to a sea water salinity of 35, using the calibration of Gray et al. (2018), the dependency of the Mg/Ca on temperature is similar to previously reported calibrations (e.g. Gray et al., 2018), although the Mg/Ca values at the lower most temperatures appear to be higher than expected (Fig. 6). This could potentially be caused by a combination of an underestimation of the salinity effect in these highly saline waters, since salinities observed here are well outside the calibration range used by Gray et al. (2018), and low temperatures, impacting the foraminiferal Mg/Ca comparatively little.”</i></p>	<p>Changed text to “After normalizing Mg/Ca values to a sea water salinity of 35, using the calibration of Gray et al. (2018), the dependency of the Mg/Ca on temperature is similar to previously reported calibrations (e.g. Gray et al., 2018), although the Mg/Ca values at the lower most temperatures appear to be higher than expected (Fig. 6). This could potentially be caused by a combination of an underestimation of the salinity effect in these highly saline waters, since salinities observed here are well outside the calibration range used by Gray et al. (2018), and low temperatures, impacting the foraminiferal Mg/Ca comparatively little.”</p> <p>Fig 6 has been replaced, the new version does not include the linear regression anymore.</p>

		From context we assume the reviewer meant to refer to the linear regression in figure 6 instead of 8. We had erroneously referred to figure 6 as figure 8 in line 219 of the original manuscript, but have corrected this now. We agree that the linear regression shown in Figure 6 is not necessary, will remove it and adjust the figure caption.	
2.19	Line 226 (and some figure captions): please make sure species names are always written in italics	We have carefully checked the manuscript to ensure all species names are now in italics.	Changed font to italics.
2.20	Line 228: an argument on the presence (production) of <i>G. ruber</i> in different seasons in the Mediterranean would benefit from references to sediment trap data. There is a nice long time series from the west (Rigual-Hernandez et al., 2012) and a new dataset from the east (Avnaim-Katav et al., 2020, Deep-Sea Research) that could be used to support these statements.	We have now included the suggested references Rigual-Hernandez et al. (2012) and Avnaim-Katav et al. (2020) in this section of the manuscript and would like to thank the reviewer for bringing these studies to our attention.	Changed text to “Although low densities were reported previously for <i>G. ruber albus</i> in the Mediterranean Sea during winter time, including being absent in large areas (Pujol and Grazzini, 1995; Bárcena et al., 2004) our finding implies that lowest values in Mg/Ca can be related to winter temperatures. <i>G. ruber albus</i> is not only present throughout the year as also shown by Rigual-Hernández et al. (2012) and Avnaim-Katav et al. (2020), but it also registers the in-situ temperature, also during seasons which are close to its lower temperature limit“
2.21	Sections 4.4.1 and 4.4.2: I believe the authors could do better in providing quantitative constraints on the strength of the processes invoked to explain the large deviations in trace metals and oxygen isotopes from the theoretical calibration curves. For example, in section 4.4.2 they seem to imply that the oxygen isotope signal should be much less affected by the individual variability, but not by lateral transport. Notwithstanding of what the value of the 30-day calculation is, one should then ask: how much lateral transport would	The majority of the scatter observed in foraminiferal $\delta_{18}O$ , Mg/Ca and Na/Ca likely does not stem from lateral transport, but appears to be an issue inherent to foraminiferal biomineralization. We do agree with the reviewer though that this aspect deserves more consideration and explanation and will include an extra paragraph in the discussion to cover these questions.	We added the following paragraph within 4.4.3: ” It is important to note that the scatter in the foraminiferal chemistry can only to a small degree be explained by lateral transport (Fig. 7). This effect may be larger in areas where the environmental conditions vary more strongly over the distance travelled by the foraminifer, and/or in basins where there is simply more lateral transport over the foraminifer’s lifetime. In our exercise, the calculated trajectories add only a minor component to the uncertainty in T (often within 0.75 °C; Fig. 7) and

	<p>be needed at each of the locations to explain the isotopic scatter? Where would the calcification have to occur? Is the offset due to lateral transport large enough or not to be considered the main mechanism behind the scatter. Similarly, if all other other processes do not act on oxygen isotopes then the scatter in isotopes (residuals) should be less than in the Mg/Ca. Is it? I feel the authors should take the discussion further and provide at least first-order assessment of the strength and direction of the invoked processes and evaluate the plausibility of those processes in explaining the scatter.</p>		<p>salinity (often within 0.25 salinity units).”</p>
<p>2.22</p>	<p>Line 265: on the same note: why is the lack of correlation “likely” due to all those uncertainties? How big are these uncertainties exactly? The reader needs to see the values to be able to evaluate statements like on line 271, which are intuitively correct, but not really supported by any calculations. Please provide R2 and p for both regressions shown in Figure 8. Also, the method by which the oxygen isotopes in seawater have been estimated is not sufficiently documented. For example, it is not clear if and how the salinity effect on Mg/Ca has been considered.</p>	<p>We have adjusted the section and replaced “likely” by “could be caused”.</p> <p>We have also added a quantification of the quality of the two different reconstructions from Figure 8 to the main text of the discussion, by presenting the residual sum of squares (comparing reconstructed values to measured values, thus using residuals of the 1:1 relationship, not the regression lines), to support our statements made in line 271 of the original version of the manuscript.</p> <p>We have added the missing p-value and R<sup>2</sup> value for the relationship between measured <math>\delta^{18}\text{O}_{\text{water}}</math> and reconstructed <math>\delta^{18}\text{O}_{\text{water}}</math> using in situ measured temperatures. They are &lt; 0.05 and 0.37 respectively.</p> <p>We agree that we did not describe the calculations well enough, we have added explanations about this to the discussion section of the</p>	<p>Changed text from “likely” to “could be caused”.</p> <p>Added “The sum of squares of the residuals (difference between reconstructed and measured values) is 9.04 when using temperatures derived from Mg/Ca and <math>\delta^{18}\text{O}_{\text{foraminifera}}</math>, but only 3.56 when using temperatures measured in situ, indicating a better reconstruction.” to section 4.4.3.</p> <p>Changed caption of Fig 8 to “The relationship between <math>\delta^{18}\text{O}_{\text{seawater}}</math> measured in the Mediterranean Sea and <math>\delta^{18}\text{O}_{\text{seawater}}</math> calculated from foraminiferal geochemistry (<i>G. ruber albus</i>). The relationship shown with dashed lines and cross shaped markers represents values calculated using foraminiferal <math>\delta^{18}\text{O}</math> as well as Mg/Ca as an additional temperature proxy to decouple the effect of temperature and salinity on <math>\delta^{18}\text{O}</math>. The relationship shown with the continuous lines and circular markers shows the</p>

		<p>manuscript. We did not correct for salinity in this case to avoid circular reasoning, since it is the aim of this section of the manuscript to reconstruct salinity, we therefore treated it as unknown.</p>	<p>same samples, but instead of using temperature values derived from foraminiferal Mg/Ca ratios, in situ measurements for temperature were used, the relationship can be described as <math>\delta^{18}\text{O}_{\text{water\_reconstructed}} = 2.62(\pm 0.69) * \delta^{18}\text{O}_{\text{water\_measured}} - 63.99(\pm 26.11)</math> with an adjusted <math>R^2</math> of 0.37. The temperature gradient was 2.2°C.”</p> <p>Added “The Mg/Ca values used here were not corrected for salinity effects, since salinity is the target parameter that has to be reconstructed and is thus treated as unknown.”</p>
2.23	<p>Line 281: why do the authors not take this opportunity to compare the performance of Na/Ca and the combined isotope and Mg/Ca on the resulting salinity estimates? There is no need to end with a general statement, when the authors have all the data to carry out the comparison.</p>	<p>We will include a section comparing the two methods, as well as describing potential issues with the use of Na/Ca as a proxy for paleosalinity.</p>	<p>Added “If salinity is reconstructed from the Na/Ca measurements using the calibration published by Mezger et al. (2016) and compared versus salinity measured in situ in the Mediterranean Sea, the reconstructed salinity follows the in situ measurements closely almost 1:1. The largest deviation from this 1:1 relationship occurs in the lower salinity range, at a salinity of 36.52 the reconstructed salinity estimates underestimate salinity by 0.71 salinity units. The average difference between in situ salinity measurements and salinity reconstructed based on one single-chamber measurement is an underestimation of salinity by 0.46 salinity units. This is still higher than the theoretical uncertainty associated when combining foraminiferal <math>\delta^{18}\text{O}</math> and temperatures derived from Mg/Ca measured at exactly the same specimens (Rohling, 2007). An uncertainty (1SD) of 1 °C in the Mg/Ca-temperature calibration (which may be particularly optimistic at high seawater temperatures), would result in an uncertainty of ~0.37 units for the reconstructed difference between two salinities.</p>

			<p>This approach will lead to an improved salinity reconstruction when the (change in) past temperatures are determined more precisely, for example by reducing the error through increased sample size. The same applies for salinity reconstructions based on Na/Ca, for which not many calibrations are available and hence, leaves room for improvement.</p> <p>While these reconstructions as well as the lack of a strong temperature effect are very encouraging results for the use of Na/Ca as a salinity proxy, the incorporation of Na into foraminiferal calcite does not appear to be homogenous across the entire shell. It has been shown that the majority of Na in <i>G. ruber albus</i> is located in the spines (Mezger et al., 2018a, 2018b), which are not well preserved in the fossil record.“ to section 4.2</p>
--	--	--	--