



- 1 Reconstructing seasonality through stable isotope and trace element analysis of the Proserpine
- 2 stalagmite, Han-sur-Lesse Cave, Belgium: indications for climate-driven changes during the last 400
- 3 years
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# 14 Abstract

15 Annually laminated speleothems allow the reconstruction of paleoclimate down to a seasonal scale. In 16 this study, an annually laminated stalagmite from the Han-sur-Lesse Cave (Belgium) is used to study 17 the expression of the seasonal cycle in northwestern Europe during the Little Ice Age. More specifically, two historical 12-year-long growth periods (ca. 1593-1605 CE and 1635-1646 CE) and one modern 18 19 growth period (1960-2010 CE) are analysed on a sub-annual scale for their stable isotope ratios ( $\delta^{13}$ C 20 and  $\delta^{18}$ O) and trace element (Mg, Sr, Ba, Zn, Y, Pb, U) content. Seasonal variability in the proxies is 21 confirmed with frequency analysis. Zn, Y and Pb show distinct annual peaks in all three investigated 22 periods related to annual flushing of the soil during winter. A strong seasonal in phase relationship 23 between Mg, Sr and Ba in the modern growth period reflects a substantial influence of prior calcite 24 precipitation (PCP). In particular, PCP occurs during summers when recharge of the epikarst is low. This 25 is also evidenced by earlier observations of increased  $\delta^{13}$ C values during summer. In the 17<sup>th</sup> century intervals, there is a distinct antiphase relationship between Mg, Sr and Ba, suggesting that varying 26 degrees of incongruent dissolution of dolomite control the observed seasonal variations. The 27 28 processes controlling seasonal variations in Mg, Sr and Ba in the speleothem appear to change 29 between the 17<sup>th</sup> century and 1960-2010 CE. The Zn, Y, Pb and U concentration profiles, stable isotope 30 ratios and morphology of the speleothem laminae all point towards increased seasonal amplitude in cave hydrology and higher drip water discharge during the 17<sup>th</sup> century. These observations reflect an 31 32 increase in water excess above the cave and recharge of the epikarst, due to a combination of lower summer temperatures and increased winter precipitation during the 17<sup>th</sup> century. This study indicates 33 34 that the transfer function controlling Mg, Sr and Ba seasonal variability varies over time. Which process 35 is dominant, either PCP or dolomite dissolution, is clearly climate-driven and can thus be used as a 36 paleoclimate proxy itself.

Keywords: Speleothem, seasonality, Little Ice Age, trace element concentrations, stable isotope ratios,
 proxy transfer functions

### 39 1. Introduction

Speleothems have been successfully used to reconstruct paleoclimate on various time scales (Fairchild and Baker, 2012), from tropical latitudes (e.g. Wang et al., 2001) to temperate areas (e.g. (Genty et al., 2003). Their ability to hold distinct annual layering enables paleoclimate reconstructions down to seasonal scale. The occurrence of visible annual laminae in speleothems has been reported from sites all over the world (Baker et al., 2008). A common expression of this visible layering is an alternation of dark compact laminae (DCL) and white porous laminae (WPL), as defined by Genty and Quinif (1996).

46 According to Baker et al. (2008), the origin of visible seasonal layering is related to seasonal variations





47 in drip rate, in drip water supersaturation and/or in cave climatology. However, in most cases, visible 48 seasonal layering is formed by changes in drip water discharge (Baker et al., 2008). Such changes in 49 drip rate often coincide with the presence of a varying degree of prior calcite precipitation (PCP). PCP 50 is the process of calcite precipitation upstream of the site of speleothem deposition (Fairchild et al., 51 2000). An increase in PCP occurs when the ability of cave waters to degas increases. Therefore, a higher degree of PCP is attributed to drier periods (Fairchild et al., 2000; Fairchild and Treble, 2009). Variations 52 in the amount of PCP have been observed on a seasonal scale (e.g. Johnson et al., 2006). The presence 53 54 of seasonally laminated speleothems in Belgian cave systems is known for several decades (e.g. Genty 55 and Quinif, 1996). The best known example is the Proserpine stalagmite collected from Han-sur-Lesse 56 Cave and first described by Verheyden et al. (2006). The speleothem has a well-expressed visual and 57 geochemical seasonal layering over the last 500 years according to layer counting and U/Th dating (Van Rampelbergh et al., 2014). This geochemical layering is reflected by sub-annual variations of stable 58 isotope ratios ( $\delta^{13}$ C and  $\delta^{18}$ O). A thorough understanding of modern seasonal control on variations in 59  $\delta^{13}$ C and  $\delta^{18}$ O in speleothem calcium carbonate results from rigorous monitoring of the conditions at 60 61 the sample site in Han-sur-Lesse cave as carried out by Van Rampelbergh et al. (2014) for the period 62 2012-2014.

In addition to the commonly used speleothem  $\delta^{18}O$  and  $\delta^{13}C$  proxies, the use of trace elemental 63 64 concentrations (e.g. Mg, Sr, Ba, Zn and U) as paleoclimate and paleoenvironmental proxies is becoming standard practice in speleothem reconstructions (Fairchild et al., 2000; Regattieri et al., 2016). The use 65 of trace elements brings additional information that can be used to unravel seasonal variability in 66 67 speleothem chemistry. Examples of this include the use of trace element concentrations as proxies for precipitation (Baldini et al., 2002; Warken et al., 2018), soil processes (Regattieri et al., 2016) or 68 69 changes in sediment supply (Regattieri et al., 2016) and can be used to identify volcanic ash fall events 70 from speleothem records (Jamieson et al., 2015).

71 The first objective of this study is to better characterize the geochemical layering by adding trace 72 element proxies to improve the understanding of processes driving the geochemical layering and to 73 further resolve the relation with seasonal climatic variability. In addition, this work also compares the seasonal cycle within earlier identified cold periods (Verheyden et al., 2006; Van Rampelbergh et al., 74 2015; Supp. Mat. Fig. 1) to present-day seasonal signals. To achieve this, two 12-year long stalagmite 75 76 growth periods (1593-1605 CE ± 30, hereafter P16 and 1635-1646 CE ± 30, hereafter P17) and a recent 77 analogue deposited between 1960-2010 CE (hereafter referred to as P19) are analysed on a sub-annual 78 scale for their stable isotopic ( $\delta^{13}$ C and  $\delta^{18}$ O) and trace elemental variations. This information is then 79 interpreted in terms of climatic changes during the last 400 years.





# 80 2. Geological setting

### 81 2.1 Han-sur-Lesse Cave

82 With a total length of approximately 10 km, the Han-sur-Lesse Cave system, located within a limestone 83 belt of Middle Devonian age, is the largest known subterranean karst network in Belgium (Fig. 1A). The cave system was formed by a meander cut-off of the Lesse River within the Massif de Boine, which is 84 85 part of an anticline structure consisting of Middle to Late Givetian reefal limestones (i.e. the Mont-86 d'Haurs and Fromelennes Formations (Fm.); Delvaux de Fenffe, 1985). The thickness of the epikarst 87 zone above the cave is estimated to be around 40 m (Quinif, 1988). Studies have shown the local 88 presence of dolomite in these Givetian limestones. Within the Mont-d'Haurs Fm., the biostromal limestones are alternated with fine-grained micritic limestones and dolomitic shales (Preat and 89 90 Bultynck, 2006). Additionally, a recent study by Pas et al. (2016) on Middle Devonian outcrops has 91 shown that dolomitized beds also occur within the limestones of the Fromelennes Fm.

92 The Han-sur-Lesse Cave is located ~200 km inland at an elevation of 200 m above sea level. The region 93 is marked by a warm temperate, fully humid climate with cool summers, following the Köppen-Geiger 94 classification (Kottek et al., 2006). In the period 1999-2013, annual temperatures averaged 10.2 °C and 95 average annual rainfall amount was 820 mm yr<sup>-1</sup> in Rochefort, 10 km from Han-sur-Lesse (Royal 96 Meteorological Institute). The study site is affected by a North Atlantic moisture source all year round 97 (Gimeno et al., 2010) and the amount of precipitation does not follow a seasonal distribution. Calculations applying the Thornthwaite formula (Thornthwaite and Mater, 1957) show that there is a 98 99 strong seasonal trend in the water excess, i.e. the amount of rainfall minus the amount lost by evapotranspiration, with water excess only occurring from October to April (Genty and Deflandre, 100 101 1998; Genty and Quinif, 1996).

102 The studied speleothem was retrieved from the Salle-du-Dôme, a 150 m wide and 60 m high chamber 103 that formed by roof-collapse of the limestone (Fig 1B). The Salle-du-Dôme is well ventilated, as it is located close to the cave exit and connected through two passages to nearby chambers. Monitoring 104 105 of cave atmosphere within the Salle-du-Dôme for the period 2012-2014 showed that in 2013 the 106 temperature inside the chamber varied seasonally between 10.5 and 14.5 °C (Van Rampelbergh et al., 2014). Similar seasonal trends in temperature are observed for the drip water, but the average is 0.5 107 108 °C colder. The pCO<sub>2</sub> of the cave air averages around 500 ppmv for the whole year. Yet, in summer (July-109 August), a rapid and temporary increase to 1000 ppmv is observed. Also during summer, rainwater 110  $\delta^{18}$ O and  $\delta$ D above the cave increase by 3 ‰ and 30 ‰ (VSMOW, Vienna Standard Mean Ocean Water) respectively, likely due to the atmospheric temperature effect as described by Rozanski et al. (1992). 111 In contrast, drip water  $\delta^{18}$ O and  $\delta$ D remain fairly stable throughout the year, with averages of -7.65 ‰ 112





- and -50.1 % VSMOW and standard deviations of 0.07 % and 0.6 % VSMOW, respectively. During late
- summer (September), an increase of 1.5 % is observed in the  $\delta^{13}$ C record of dissolved inorganic carbon
- 115 (DIC) within the drip water

# 116 2.2 Proserpine speleothem

The Proserpine speleothem is a 2 m high, tabular shaped stalagmite. The speleothem has a surface 117 118 area of 1.77 m<sup>2</sup> and is fed by a drip flow with drip rates ranging between 100 and 300 mL min<sup>-1</sup>. The 119 speleothem grew over a period of approximately 2 kyr and has thus an exceptionally high average 120 growth rate of 1 mm yr<sup>-1</sup>. The large speleothem was drilled and a 2 m long core was retrieved. The 121 upper 50 cm of this core, dating back to approximately 1500 CE (Supp. Mat. Fig. 2), shows a well-122 expressed layering of alternating DCL and WPL (Verheyden et al., 2006). Previous studies concluded 123 that multi-decadal simultaneous changes in different proxies (such as crystal fabric, growth rate, layer 124 thickness, and oxygen and carbon stable isotope ratios) indicate that these are controlled by common 125 climatic, environmental or anthropogenic factors, despite the observation that some parts of the Proserpine speleothem appear to have been deposited out of isotopic equilibrium with the drip water 126 127 (Verheyden et al., 2006; Van Rampelbergh et al., 2015). Based on a detailed cave monitoring study at the Proserpine site in the years 2012 to 2014, Van Rampelbergh et al. (2014) showed that  $\delta^{18}$ O and 128 129  $\delta^{13}$ C of seasonally deposited calcite reflect isotopic equilibrium conditions and that variations of stable 130 isotope ratios are induced by seasonal changes. These seasonal changes in stable isotope ratios correspond with the observed visible layering. The speleothem  $\delta^{18}$ O value is believed to reflect changes 131 in seasonal cave climatology. While drip water  $\delta^{18}$ O remains constant, calcite  $\delta^{18}$ O decreases with ~0.6 132 133 ‰ in summer months, caused by temperature-dependent fractionation during calcite precipitation. This fractionation was calculated to be -0.2 % °C<sup>1</sup>. In contrast,  $\delta^{13}$ C reflects seasonal changes occurring 134 at the epikarst level. A ~1.5 % increase of  $\delta^{13}$ C in drip water DIC during late summer is directly reflected 135 136 in the freshly deposited speleothem calcite. The enrichment in drip water  $\delta^{13}$ C values occurs shortly 137 after the observed decrease in drip water discharge, and therefore seasonal variations in the degree of prior calcite precipitation in the epikarst has been hypothesized to be the main driver of seasonal 138 139  $\delta^{13}$ C changes in the drip water (Van Rampelbergh et al., 2014).

# 140 2.3 Dating

The age-depth model of the Proserpine speleothem core has been established and discussed by Van Rampelbergh et al. (2015) and is provided in the supplementary material (Supp. Mat. Fig. 2). This agedepth model was constructed by using a combined approach of U-Th radiometric dating, based on 20 U-Th ages, and layer counting. This has shown that the amount of counted layers is in good agreement with the U-Th ages (see Table 2 in Van Rampelbergh et al., 2015). However, at 9 to 10 cm from the top





146 of the core, a perturbation with heavily disturbed calcite occurs, making it impossible to construct a 147 continuous layer counted chronology. Remains of straw and soot were found within this perturbation, suggesting that at that time, fires were lit on the speleothem's paleosurface (Verheyden et al., 2006). 148 149 Layer counting gave an age of 1857 ± 6 CE for the reestablishment of calcite deposition after the 150 perturbation and U-Th age-depth modeling showed that the start of the perturbation occurred at 1810 151 ± 45 CE (Van Rampelbergh et al., 2015). Radiocarbon dating of the straw fragments embedded in the 152 calcite gave an age between 1760 and 1810 CE, with 95.4 % probability. The age of 1810 ± 45 CE is 153 used to restart the layer counting after the perturbation towards the bottom of the core. This gave an 154 age of 1593 to 1605  $\pm$  30 CE for P16 and 1635 to 1646  $\pm$  30 CE for P17. The more recent section P19 155 studied here is situated above the perturbation and its age could be confidently established through annual layer counting at 1960-2010 CE. 156

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### 158 3. Methods

### 159 3.1 Analytical procedures

The three growth periods studied are shown in Fig. 2 and their age is derived from an age-depth model 160 based on U-Th-dating and layer counting (Verheyden et al., 2006; Van Rampelbergh et al., 2015). For 161 162  $\delta^{13}$ C and  $\delta^{18}$ O analysis, powder samples are acquired using a Merchantek Micromill 163 (Merchantek/Electro Scientific Industries Inc. (ESI), Portland (OR), USA, coupled to Leica GZ6, Leica Microsystems GmbH, Wetzlar, Germany) equipped with tungsten carbide dental drills with a drill bit 164 165 diameter of 300  $\mu$ m. The powders are stored in a 50 °C oven prior the analysis to avoid  $\delta^{13}$ C and  $\delta^{18}$ O 166 isotopic contamination. Measurements for P16 and P17 are carried out on a Nu Perspective isotope ratio mass spectrometer (IRMS) coupled to a Nucarb automated carbonate preparation device (Nu 167 168 Instruments, UK) at the Vrije Universiteit Brussel (Belgium). The  $\delta^{13}$ C and  $\delta^{18}$ O records of P16 and P17 consist of 201 and 116 data points, respectively, resulting in temporal resolutions of ~20 and ~10 data 169 points per year, respectively. The analysis of the P19 interval is an extension of the previously published 170 171 seasonally resolved 1976-1985 transect (Van Rampelbergh et al., 2014) and is carried out on a Delta 172 plus XL IRMS coupled to a Kiel III carbonate preparation unit (Thermo Fisher Scientific, Germany) also at the Vrije Universiteit Brussel. For P19, a total of 350 samples are analysed, providing a temporal 173 174 resolution of ~7 data points per year. Within each batch of ten samples, the in-house reference 175 material MAR2-2, prepared from Marbella limestone and calibrated against NBS-19 (Friedman et al., 176 1982) is measured together with the samples to correct for instrumental drift ( $\delta^{13}$ C: 3.41 ± 0.10 ‰ (2s) 177 VPDB;  $\delta^{18}$ O: 0.13 ± 0.20 ‰ (2s) VPDB). All results are displayed as ‰VPDB (Vienna Pee Dee Belemnite) 178 with the individual reproducibility reported as 2 standard deviation (SD) uncertainties. Averages of the





total 2 SD uncertainties for  $\delta^{13}$ C and  $\delta^{18}$ O are 0.03 ‰ and 0.09 ‰ for the Nu Perspective setup. With the Delta plus XL setup these are slightly higher, being 0.04 ‰ and 0.10 ‰ for  $\delta^{13}$ C and  $\delta^{18}$ O, respectively (Van Rampelbergh et al., 2014).

182 Trace element variations are determined using inductively coupled plasma-mass spectrometry complemented by a laser ablation sample introduction system (LA-ICP-MS) at Ghent University 183 184 (Belgium). The LA-ICP-MS setup consists of a 193 nm ArF\*excimer Analyte G2 laser ablation system (Teledyne Photon Machines, Bozeman, MT, USA) coupled to a single-collector sector field 'Element XR' 185 ICP-MS unit (Thermo Fisher Scientific, Bremen, Germany). The laser is used to sample adjacent 186 187 positions along a line segment parallel to the stalagmite's growth axis. The positions are ablated one-188 by-one for 15 s with a laser spot size of 50  $\mu m$  in diameter, a repetition rate of 30 Hz and a beam energy density of 3.51 J cm<sup>-2</sup>. The line segments for P16, P17 and P19 are drilled at 287, 249 and 445 individual 189 190 positions, respectively. Sampling via individual drilling points is preferred over the conventional 191 approach of continuous line scanning be-cause the single positions can be sampled longer, resulting in an improved limit of detection. To carry out the analyses, the speleothem sections and reference 192 193 materials are mounted in a HELEX 2 double-volume ablation cell. The Helium carrier gas (0.5 L min<sup>-1</sup>) is 194 mixed with Argon make-up gas (0.9 L min<sup>-1</sup>) downstream of the ablation cell, and introduced into the 195 ICP-MS unit, operated in low mass-resolution mode. Transient signals for Magnesium (Mg), Aluminium 196 (Al), Silicon (Si), Phosphorus (P), Sulphur (S), Potassium (K), Iron (Fe), Manganese (Mn), Zinc (Zn), 197 Rubidium (Rb), Strontium (Sr), Yttrium (Y), Barium (Ba), Lead (Pb), Thorium (Th), and Uranium (U) are 198 monitored during analysis of the laser-induced aerosol. Cool plasma conditions (800 W RF power) are 199 used to reduce Argon-based interferences and to increase the sensitivity of the analysis. A gas blank 200 subtraction is performed on the data acquired at each position, based on the signal acquired 10 s prior 201 to the ablation. Precise and accurate trace element concentration data are obtained from offline 202 calibration, using seven international natural and synthetic glass and carbonate reference materials 203 BHVO-2G, BIR-1G, GSD-1G, GSE-1G, and MACS-3 (United States Geological Survey) as well as SRM 610 204 and 612 (National Institute of Standards and Technology). Ca is used as an internal standard, following 205 the assumption that the calcium carbonate in the speleothem is made up of 38 wt. % Ca. Based on the 206 reference materials and settings described, the repeatability for the produced elemental 207 concentration data is typically on the order of 5% relative standard deviation (RSD). Limits of detection 208 (LODs) are given in Table 1.

### 209 3.2 Data Processing

Frequency analysis is applied to study the variations in the different proxy signals, and allowsevaluating which of these proxies records the seasonal cycle. The ability of frequency analysis to assess





212 the potential of a proxy to record the seasonal cycle in speleothems and other incremental climate 213 archives is already recognized by Smith et al. (2009) and de Winter et al. (2017). Furthermore, the 214 method can identify multi-annual trends or variability at the sub-seasonal level. Frequency analysis is 215 performed using Fast Fourier Transformations (FFT) of the isotopic and trace element data in the 216 distance domain. The data are de-trended and padded with zeros. The power spectra are plotted as 217 simple periodograms with frequencies plotted in the distance domain (mm<sup>-1</sup>) to allow intuitive 218 interpretation. The significance level (95%) is evaluated using Monte Carlo noise simulations. The 219 routine used operates in MATLAB® and is based on the scripts provided in Muller and MacDonald 220 (2000), which are explained in more detail in Bice et al. (2012).

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222 An effective method to compare sub-annual variations of different proxies with each other is by 223 resampling multiple annual cycles at a regular interval and stacking the individual cycles (Treble et al., 224 2003; Johnson et al., 2006; Borsato et al., 2007; de Winter et al. 2018). The advantage of this method 225 is that the phase-relations of the different proxies are preserved (Treble et al., 2003). Annual stacks are created based on the moving averages to diminish the influence of low-frequency noise on the 226 227 annual stacks. The number of points used for moving averages is determined as a function of the 228 sampling resolution (i.e., 3-point moving average for stable isotope records and 5-point moving 229 average for trace element records, see Fig. 5). Proxy records with well-constrained seasonal variation 230 are used to define seasonal cycles. In this study, individual years are selected based on  $\delta^{13}$ C (minima) 231 for stable isotope records and Zn (maxima) for the trace element records. Stable isotope ratios and trace element stacks are created separately (Fig. 2). For P16 and P17, all annual cycles are included in 232 233 the stack, except for the first and the last one, since there is no guarantee that these are entirely 234 represented in the record. For P19, only ten years were selected from the full record to avoid any effect multi-decadal variability and to maintain an approach similar to that of P16 and P17. The years are 235 236 indicated by the red line in Fig. 5.

### 237 4. Results

238 The concentration range of each proxy measured in the three different intervals is shown in Fig. 3. For 239  $\delta^{13}$ C and  $\delta^{18}$ O, the average values and ranges (minima to maxima) in P19 are significantly higher than those in P17 and P16. To illustrate the spread in the trace element records, the median is used instead 240 of the average as it is less sensitive to large concentration ranges and outliers. Al, Si, K, Mn, Rb and Th 241 242 are not included in this study since > 25% of the data falls below the LOD. An exception is made in the case of Y; few data points are retained for P17 (81% of the data is < LOD in P17 and 18% and 36% of 243 the data is < LOD in P19 and P16 respectively. However, Y data are discussed because of the clear 244 245 seasonal signal shown in P19 and P16 (Supp. Mat. Fig. 3 and 5).





246 Records of stable isotope ratios ( $\delta^{13}$ C and  $\delta^{18}$ O) and trace element (Mg, Zn, Sr, Y, Ba, Pb and U) 247 concentrations are plotted in the distance domain in Fig. 5. The occurrence of darker laminae (DCL) in the samples is indicated by blue bands, clearly showing that layers are thicker in P16 and P17 (average 248 249 1.135 mm and 1.096 mm, respectively) compared to P19 (average 0.382 mm). For all intervals, the seasonal cycles are well constrained by  $\delta^{13}$ C, with lower  $\delta^{13}$ C values occurring in DCL. The average  $\delta^{13}$ C 250 251 is higher for P19 (-8.36 ‰) compared to P17 and P16 (-9.82 ‰ and -10.04 ‰, respectively). In addition, 252 the amplitude of the individual cycles is larger in P19. Seasonal cycles in  $\delta^{18}$ O are much less 253 pronounced. The most distinctive cycles are observed in P19 and some can be identified in parts of 254 P17 and P16 (e.g. between 4 and 7 mm in P16 or between 3 and 7 mm in P17), while for other parts 255 (e.g. between 7 and 11 mm in P16) they appear to be absent.

256 Seasonal variations are observed for Mg, Sr and Ba in all three intervals investigated (Fig 5). In P17 and 257 P16, the median concentrations of these elements are closely related; 447 and 444  $\mu$ g g<sup>-1</sup> for Mg, 51 and 45  $\mu$ g g<sup>-1</sup> for Sr and 36 and 33  $\mu$ g g<sup>-1</sup> for Ba (Fig. 3). However, in P19 concentrations of Mg and Ba 258 are slightly higher compared to the older intervals, i.e. 706  $\mu$ g g<sup>-1</sup> for Mg and 46  $\mu$ g g<sup>-1</sup> for Ba. This is 259 260 also the case for Pb and U with concentrations significantly lower in P17 (0.14 and 0.05  $\mu g g^{-1}$ , respectively) and P16 (0.14 and 0.07  $\mu$ g g<sup>-1</sup>, respectively) and a seasonal cycle that is less pronounced 261 than in P19 (0.37 and 0.18  $\mu$ g g<sup>-1</sup>). In contrast, P16 has the highest median concentrations of Zn (54  $\mu$ g 262 263  $g^{-1}$ ) and Y (0.04 µg  $g^{-1}$ ) and both elements display a well-defined seasonal covariation. Although the concentration of Zn is lower in P19 and P17 (14 and 25  $\mu$ g g<sup>-1</sup>, respectively), the seasonal cycle is still 264 265 present. Similar observations can be made for Y in P19 (0.02 µg g<sup>-1</sup>). Within P16 and P17, maxima of 266 Zn, Y, Sr and Ba mostly occur within the DCL.

Figure 4 shows an example of the FFT periodograms of  $\delta^{13}$ C, Mg, Zn and P in P16. Additional 267 268 periodograms for the other elements in P16, P17 and P19 are included as supplementary data (Supp. Mat. Fig. 3-5). The frequency analysis confirms the clear seasonal cyclicity of  $\delta^{13}$ C previously observed 269 by Van Rampelbergh et al. (2014) (Fig. 4). The dominant frequency of  $\delta^{13}$ C in P16 is 0.8 mm<sup>-1</sup> (Fig. 4). 270 271 This corresponds to a period of 1.25 mm, which is in good agreement with an observed average layer thickness of 1.13 mm (Supp. Mat. Fig. 6). Because of its distinct seasonal cyclicity, the  $\delta^{13}$ C cycle is used 272 273 as a reference to deduce whether or not other proxies record the seasonal cycle. Mg and Zn appear to 274 track this seasonal cycle well as their periodograms contain peaks at 0.8 and 0.75 mm<sup>-1</sup> respectively, 275 corresponding closely to the frequency of  $\delta^{13}$ C. For Zn, a broader double peak is observed with a main 276 period of 1.18 mm and a smaller period of 1.02 mm. This double peak in the periodogram is caused by 277 small variations in the thickness of the annual cycles around an average thickness of 1.14 m with a 278 lightly skewed distribution towards thinner layers (see Supp. Mat. Fig. 6). The P record doesn't display





- any significant seasonal cycle (95% confidence) (Fig. 4). For P19, visible layers are thinner (average
- 280 0.382 mm) and also the variation in thickness is larger (RSD 28.9%) compared to P16 and P17 (Supp.
- 281 Mat. Fig. 6). This results in broader and less well defined seasonal peaks in the periodograms.

# 282 5. Discussion

### 283 5.1 Seasonal cyclicity in trace element records

284 A schematic overview of the observed changes in the proxies discussed below and the interpretation for the three intervals is provided in Table 2. Assessing the exact phasing of the seasonal cycles of 285 286 different trace elements to  $\delta^{13}$ C and the visible layering remains challenging since 1) often a multitude 287 of factors control trace element variations within speleothems and 2) stable isotope ratios and the trace element concentrations are not measured on the same samples. An example of such a phase 288 289 problem is the occurrence of an additional year in P16 in the trace element curve compared to  $\delta^{13}$ C 290 (Fig. 5, between 1 and 6 mm). Nevertheless, overall,  $\delta^{13}$ C minima occur in the DCL, suggesting a similar timing (and maybe control) on the visible laminae and  $\delta^{13}$ C cycles. Trace element proxies show cyclicity 291 with a similar frequency as the  $\delta^{13}$ C (Fig. 4), in contrast to  $\delta^{18}$ O which seasonal cycles in P16 and P17 292 are less clear. 293

### 294 5.1.1 Zinc, yttrium and lead proxies

295 In earlier monitoring studies carried out in the Père-Noël Cave (also part of Han-sur-Less Cave system, 296 Fig. 1), the presence of a late autumn increase in discharge was identified (Genty and Deflandre, 1998; 297 Verheyden et al., 2008). In-situ conductivity measurements indicated an elevated mineral and/or 298 organic matter increase during this autumnal increase in drip water discharge (Genty and Deflandre, 299 1998; Verheyden et al., 2008). Measurements of the drip water discharge above the Proserpine stalagmite show that in late November, a doubling of the discharge volume occurs. This increased 300 301 discharge is maintained until May, when a gradual decrease is observed (Van Rampelbergh et al., 302 2014). The timing of the elevated discharge agrees with the theoretical water excess occurring above 303 the cave (Genty and Quinif, 1996). The observed seasonal cycle in Zn, Y and Pb in the intervals studied 304 is likely caused by this annual winter flushing. Variations in these trace metal concentrations within 305 annual speleothem layers have previously been attributed to the annual hydrological cycle. For 306 instance, Borsato et al. (2007) linked the peak concentrations of F, P, Cu, Zn, Br, Y and Pb to the annual 307 increase of soil infiltration during autumnal rainfall. Furthermore, it was suggested that the transport 308 of such elements mainly occurs via natural organic matter (NOM) or other form of colloidal material. Enrichments of these soil-derived elements within speleothems are believed to be associated with high 309 310 drip water flows (Fairchild and Treble, 2009). Studies have shown that trace metals, such as Cu, Ni, Zn,





311 Pb, Y and REE, are predominantly transported via complexing by NOM, of which the fraction size in the 312 karstic waters ranges from nominally-dissolved to colloidal-to-particulate (Hartland et al., 2012; Wynn et al., 2014). In the case of Zn and Pb, Fairchild et al. (2010) have shown that in Obir Cave (Austria) the 313 314 visible and ultra-violet lamination forms during autumn and is enriched in Zn, Pb and P. According to 315 Wynn et al. (2014), the correspondence of distinct Zn and Pb peaks with the autumnal laminae is 316 compelling evidence for a high-flux transport of these trace metals with NOM. However, in this study 317 no distinct annual cycle within the P record is observed (Fig. 4). Phosphorus is considered soil derived 318 as it originates from vegetation dieback (e.g. Baldini et al., 2002). Therefore, P has shown similar 319 variations as observed in Zn, Y and Pb in previous studies (Borsato et al., 2007; Fairchild et al., 2010). 320 In the Proserpine speleothem, no relation between P and other soil derived trace elements is detected. 321 An explanation for this can be similar to that proposed by Frisia et al. (2012), being that P is not derived 322 from soil leaching, but from other sources such as phosphate minerals present in the epikarst or 323 microbiological activity.

324 Because of the distinct signature of the seasonal cycle in Zn, the Zn peaks are used as tie-points to 325 create the annual stacks of other trace element records (Fig. 6), with lower concentrations occurring during periods of lower discharge and vice versa. The much higher Zn and Y peaks in P16 compared to 326 P17 and P19 suggest an increased seasonality effect in discharge; therefore the accompanied annual 327 flushing of the soil above cave appears more intense in the early 17th century. Concentrations of Pb 328 are significantly higher in P19 compared to the other periods (median of 0.37  $\mu$ g g<sup>-1</sup> versus 0.14  $\mu$ g g<sup>-1</sup> 329 and 0.14 µg g<sup>-1</sup> in P16 and P17, respectively). An increase of Zn and Y in P19 similar to that in Pb is not 330 331 observed, suggesting that the Pb-enrichment occurs at the soil level from another source. A study of 332 Allan et al. (2015) on Pb isotope ratios in the in the same Proserpine stalagmite shows that the Pb 333 concentrations are soil derived and originate from various sources of anthropogenic atmospheric 334 pollution (coal, industrial activities, steel production and road dust). This explains well the observed 335 higher Pb concentration in P19. Allan et al. (2015) identified increases in Pb concentration during 1945-336 1965 CE and 1975-1990 CE, which are in agreement with the observed higher Pb concentrations in this 337 study between 20-18 mm and 13-5 mm. They also concluded that this 20<sup>th</sup> century anthropogenic 338 pollution only affects Pb and none of the other elements used as paleoseasonality proxy in this study.

### 339 5.1.2 Magnesium, strontium and barium proxies

Figure 6 shows that the annual stacks of Sr and Ba exhibit correlate strongly within all three intervals, evidenced by Pearson correlation coefficients (*r*) of 0.71, 0.97 and 0.82 for P19, P17 and P16, respectively with p-values much smaller than 0.01 (99% confidence level). Magnesium displays an antiphase relationship with Sr and Ba in P16 (r = -0.85, p-value =  $1.8 \times 10^{-7}$ ) whereas in P19 this





344 relationship is in phase (r = 0.64, p-value =  $7.2^{*}10^{-4}$ ). For P17, there is no significant relationship 345 between Mg with Sr and Ba (low correlation, r = -0.13, p-value = 0.53). A strong covariation of Mg with Sr and Ba, as observed in P19, has previously been attributed to reflect the presence of prior calcite 346 347 precipitation (PCP) in the epikarst above the cave, caused by the occurrence of drier periods (Fairchild 348 et al., 2000), even on a seasonal scale (Johnson et al., 2006). The presence of PCP during late summer 349 (with high evapotranspiration above the cave), when strongly reduced drip water discharge exists 350 above the Proserpine stalagmite, has also been evoked to explain the enriched  $\delta^{13}C$  of freshly 351 deposited calcite during the cave's summer mode (Van Rampelbergh et al., 2014). Despite the 352 difficulties of accurately correlating trace elements and stable isotope proxies, there appears to be a 353 good agreement between the P19 Mg and  $\delta^{13}$ C record, with maxima in Mg corresponding with maxima 354 in  $\delta^{13}$ C, confirming the hypothesis of PCP control on these proxies. In contrast, the antiphase variation 355 in Mg with respect to Sr and Ba observed in P16, suggests the involvement of other processes that 356 dominate over PCP. A positive relationship between the Mg partition coefficient and temperature 357 would be expected from thermodynamic considerations, and this has indeed been observed in experimental carbonate precipitation studies (Gascoyne, 1983; Rimstidt et al., 1998; Huang and 358 359 Fairchild, 2001; Day and Henderson, 2013). In similar experiments, strontium partitioning into 360 inorganic carbonate is known to remain constant with increasing temperatures but can be influenced 361 by calcite precipitation rate (Day and Henderson, 2013). Faster precipitation of calcite causes an 362 increased amount of lattice defects, resulting in an increased value for the partition coefficient of Sr 363 (Pingitore and Eastman, 1986) and thus more Sr uptake in the calcite. Higher temperatures, combined with a decrease in drip water discharge, leading to decreased growth rates, could therefore 364 365 theoretically explain the antiphase relationship of Mg and Sr. However, growth rates in P16 are rather high and additionally, it has been suggested that the variations of Sr and Mg in drip water chemistry 366 are often significantly higher than those caused by the processes mentioned above. Roberts et al. 367 368 (1998) concluded that the temperature-dependence of the Mg partition coefficient could theoretically 369 explain the observed seasonal Mg variations, but not the multi-annual trends, for which hydrological 370 changes are likely more important. Such observations have caused the interpretation of the Mg proxy to shift from a temperature relationship to an interpretation in terms of hydrological chances such as 371 372 amount of water recharge in the epikarst (Fairchild and Treble, 2009). In this study, a more likely 373 explanation for the P16 antiphase relation in Mg, Sr and Ba is the incongruent dissolution of dolomite 374 (CaMg(CO<sub>3</sub>)<sub>2</sub>; IDD), taking place during annual periods that are characterized by enhanced water-rock 375 interaction. The presence of dolomite within lateral-equivalent Givetian limestone deposits in Belgium 376 has been recognised by Pas et al. (2016). Dolomitized parts of the limestone host-rock were observed 377 within the nearby Père-Noël Cave (Fairchild et al., 2001). During periods of decreased recharge, i.e. 378 summer for the Han-sur-Lesse Cave, prolonged interaction between water and rock leads to saturation





of the karstic water with respect to CaCO<sub>3</sub>. When saturation is reached, incongruent dissolution of dolomite (IDD) will start and Ca<sup>2+</sup> concentration remains constant due to the precipitation of calcite (Lohmann, 1988). IDD increases the Mg/Ca of the drip water (Fairchild et al., 2000), but lowers the Sr/Ca and the Ba/Ca, because dolomite tends to have lower Sr and Ba contents with respect to calcite (Roberts et al., 1998). The IDD process is believed to overwhelm the PCP signal in P16 and is likely responsible for the observed antiphase relation. During winter recharge, saturation of the water in the epikarst with respect to calcite is not attained and dolomite does not dissolve.

The comparison of the annual stacks for Mg, Sr and Ba of the different intervals corroborates the idea 386 387 that PCP is the main process controlling the seasonal variations of these trace elements in P19 based 388 on the in-phase relation of Mg, Sr and Ba. Within P16, enhanced seasonality in recharge causes IDD to 389 dominate over PCP. This explains the antiphase relation of Mg against Ba and Sr. Somewhere between 390 the P16 and P19 periods, a turnover in the hydrological regime of the epikarst allowed PCP to become 391 dominant over IDD in the seasonal variations in the proxies. Within P17, the relationship between Mg, Sr and Ba is less clear. This could point towards a change in hydrological regime between the periods 392 393 of deposition of P16 and P19.

### 394 5.1.3 Uranium

395 In speleothems, U is thought to be mainly derived from bedrock dissolution (Bourdin et al., 2011; Jamieson et al., 2016) and to be subsequently transported by the ground water towards the 396 speleothem (Fairchild and Baker, 2012). The partition coefficient of U is <1 for calcite (Johnson et al., 397 398 2006; Jamieson et al., 2016). This causes U to be preferentially excluded from the calcite and enriched 399 in the remaining drip water during the process of PCP. However, in P19, where PCP is evoked as the 400 dominant process controlling Mg, Sr and Ba seasonal variations, an antiphase relationship of U with Mg, Sr and Ba is observed (Fig. 6). Johnson et al. (2006) concluded that scavenging of U as uranyl ion 401 402  $(UO_2^{2+})$  from the drip water onto the calcite crystal surfaces during PCP has a more dominant control 403 on seasonal U variability than the partition coefficient. Such mechanism explains why U is antiphase 404 with the Mg, Sr and Ba variations.

### 405 5.2 Seasonal variations in $\delta^{13}$ C and $\delta^{18}$ O

To compare and understand the seasonal variations in  $\delta^{13}$ C and  $\delta^{18}$ O, annual stacks were created (Fig. 7) by virtual resampling based on the occurrence of peaks in  $\delta^{13}$ C values as this proxy reflects the seasonal cycle best (Fig. 4). The minima in  $\delta^{13}$ C always occur in DCL, for P16 and P17. In P19, this relation is less clear, however on close inspection nearly all of the  $\delta^{13}$ C minima occur within the DCL (Fig. 5). Van Rampelbergh et al. (2014) suggested that seasonal changes in  $\delta^{13}$ C of recent calcite are





411 driven by changes in PCP. Higher  $\delta^{13}$ C values occur when more PCP is observed, i.e. during periods of lower recharge. The in phase variations of Mg, Sr and Ba in P19 described above supports the 412 hypothesis of a seasonally changing degree of PCP. Seasonal variations in the amount of PCP and its 413 414 effect on  $\delta^{13}$ C has previously been recognized in monsoon regions (Johnson et al., 2006; Ridley et al., 415 2015). During P16, seasonal changes in incongruent dolomite dissolution dominate the trace element variations of Mg versus Sr and Ba over PCP. However, since the main source of carbon in Han-sur-Lesse 416 417 cave waters is the vegetation cover above the cave (Genty et al., 2001), IDD is not expected to change 418 the  $\delta^{13}$ C signal. For example, a case study carried out by Oster et al. (2014) showed that an increase in 419 IDD did not affect the  $\delta^{13}$ C of the speleothem significantly, despite a difference of ~0.5 % in  $\delta^{13}$ C 420 between the limestone and dolomite component in the host rock. Since  $\delta^{13}$ C is not affected by IDD, the influence of PCP on the  $\delta^{13}$ C remains observable. Indeed, similar as in P19, for both P17 and P16 421  $\delta^{13}$ C minima occur within DCL, suggesting that these DCL layers were deposited by during seasonal 422 423 periods of increased drip water discharge.

424 Observations from cave monitoring have shown that seasonal changes in cave temperature (11°C -425 15°C) are the main driver of  $\delta^{18}$ O variations in freshly deposited calcite (-7.0‰ - -6.2‰; Van Rampelbergh et al., 2014). The  $\delta^{18}$ O periodograms show that the seasonal cycle is less developed 426 427 compared to  $\delta^{13}$ C (Fig. 4 and Supp. Mat. Fig. 3-5). This is also expressed in the annual stacks (Fig. 7). 428 For P19, there is tendency towards a positive correlation of  $\delta^{13}$ C and  $\delta^{18}$ O but in P17 and P16 this is unclear. Although analysis of recent calcite have clearly shown that  $\delta^{18}$ O values are at least partly 429 controlled by the cave temperature, interpretation of the seasonal  $\delta^{18}$ O changes is difficult due to the 430 431 reduced seasonal cyclicity in the  $\delta^{18}$ O records compared to other proxies. However, average  $\delta^{18}$ O 432 values of speleothem calcite are obviously more depleted for P17 and P16 compared to P19 (Fig. 3 and Fig. 5). The hypothesis put forward here is that the lower average  $\delta^{18}$ O values of P16 point towards an 433 434 increase in winter precipitation above the cave, since Van Rampelbergh et al. (2014) has shown that 435 winter precipitation, such as the presence of snow, above Han-sur-Lesse cave causes a severe decrease 436 in  $\delta^{18}$ O of the precipitation. Subsequently, this decrease is then transferred to the drip water and into 437 the speleothem calcite.

### 438 5.3 Variability in the seasonal cycle

The observed changes of the seasonal variations in Mg, Sr and Ba between P19, P17 and P16 can only be explained by a change in the process controlling the seasonal variability in Mg, Sr and Ba. In the recent period, between 1960 and 2010 CE, PCP is identified as the main driver for seasonal changes in Mg, Sr, Ba trace element concentrations. This hypothesis is supported by the  $\delta^{13}$ C variations. In the 17<sup>th</sup> century intervals, Mg, Sr and Ba variations suggest that incongruent dissolution of dolomite rather





444 than PCP, dominates the seasonal signal. Fairchild and Baker (2012) defined the term transfer function 445 to describe the quantitative relation between the speleothem chemistry and changing cave environments or climate. In this case, there is a change in (qualitative) transfer function from 446 447 incongruent dolomite dissolution to prior calcite precipitation. This change in transfer function is likely 448 to be climate-controlled since there are no indications for drastic changes in cave morphology over the 449 last 500 years, as interpreted from the long term stable isotope ratio record (Van Rampelbergh et al., 450 2015 and Supp. Mat. Fig. 1). It is known that the strength of the acting transfer function can be used 451 as a paleoclimate proxy. For example, Jamieson et al. (2016) demonstrated that the seasonal (anti-)correlation between  $\delta^{13}C$  and U/Ca varies through time within a Common Era stalagmite from Belize. 452 453 During drier years, reduced seasonal variability in prior aragonite precipitation causes U/Ca and  $\delta^{13}$ C 454 to correlate more positively compared to wetter years. This illustrates how a transfer function can be 455 regarded as a valuable paleoclimate proxy. In any case, a certain threshold must be reached for a 456 switch between transfer functions to take place. A prerequisite for PCP to occur is the presence of 457 sufficient karstic voids filled with a gas phase characterized by a lower pCO<sub>2</sub> than that with which the infiltrating waters previously equilibrated (Fairchild and Treble, 2009). The presence of such karstic 458 459 voids is dependent on the multi-annual to decadal recharge amount of the karstic aquifer. Indeed, the 460 average values of trace element concentrations imply an increased water availability during P16 and 461 P17 compared to P19. More specifically peaks in soil-derived trace element concentrations (Zn and Y) 462 are higher for P16, pointing towards enhanced flushing and an increased seasonality in water 463 availability. An anthropogenic influence explains the higher concentrations of Pb in P19. In addition, trace element concentrations originating from host rock dissolution (Mg, Sr, Ba and U) are significantly 464 lower for P16, resulting from lower multi-annual water residence time. Lastly, layers in P16 and P17 465 are up to three times thicker compared to P19 (Fig. 2 and Supp. Mat. Fig. 6), which reflects higher 466 growth rates. The positive relationship between water supply and growth rate has been demonstrated 467 468 in the past (Baker et al., 1998; Genty and Quinif, 1996). In large and irregular shaped stalagmites, such 469 as the Proserpine, within-layer thickness can often be quite large (Baker et al., 2008). The long-term layer thickness evolution shows a clear difference between the 17th century and present day. The 470 significantly thinner layers during recent times clearly indicate that less water is available compared to 471 472 the 17<sup>th</sup> century.

A straightforward explanation for the observed wetter cave conditions during, in particular, P16 is an increase in seasonal water excess. An elevated water excess can be caused by an increase in precipitation or a decrease in temperature. A lower temperature, especially during summer, results in a decreased evaporation of surface water. Calculations of present-day potential evapotranspiration (PET) with the Thornthwaite equation (Thornthwaite and Mather, 1957) for the period 1999-2012





478 show a negative water excess lasting from May to September (Fig. 8). Although the Thornthwaite and 479 Mather (1957) method does not include vegetation effects, it is still a reliable tool to provide an idea of the effect of changes in the temperature and/or precipitation on the PET (Black, 2007). The effect 480 481 of a temperature decrease during summer months on the water excess was simulated with an 482 arbitrarily chosen 1°C temperature drop compared to the 1999-2012 average monthly temperature. 483 Such a temperature drop appears to have only a minor influence (Fig. 8). A hypothetical increase of 484 total annual rainfall with 200 mm, equally spread across 12 months, has a much larger effect on the 485 water excess (Fig. 8). However, this would decrease the length of the annual interval during which no 486 recharge occurs (i.e. only during June-July instead of May-September) providing less suitable 487 conditions for dolomite dissolution to occur. Therefore, the most plausible explanation would be to 488 have a stronger seasonal distribution in the amount precipitation (with more winter precipitation), 489 whereas today no seasonality in the amount of rainfall is observed.

### 490 **5.4 Implications for 17<sup>th</sup> century paleoclimate**

The majority of Common Era paleoclimate reconstructions are based on tree-ring data (D'Arrigo et al., 491 492 2006), although other records, for example historical documents (e.g., Dobrovolny et al., 2010), ice 493 cores (e.g., (Zennaro et al., 2014) or speleothems (e.g., Baker et al., 2011; Cui et al., 2012) are used as 494 well. Over the last decades, consensus has been reached that changes in solar irradiance and volcanic 495 activity are the main drivers of short-term natural climate variability during the last millennium (e.g. 496 Crowley, 2000; Bauer et al., 2003). Interpretations of the stable isotope and trace element proxies obtained on the Proserpine speleothem show that a higher recharge state of the karstic aquifer 497 498 characterizes the 17<sup>th</sup> century intervals compared to 1960-2010. Such an increase in recharge requires 499 a decrease in evapotranspiration, which can result either from lower summer temperatures or higher total annual precipitation. Although it is difficult to discriminate between both, the effect of a total 500 501 annual precipitation increase on the recharge is expected to be higher compared to a decrease in 502 summer temperature (Fig. 8). Globally dispersed regional temperature reconstructions indicate that 503 multi-decadal warm or cold intervals, such as the Medieval Warm Period or the Little Ice Age (LIA), are 504 not global events. Yet, a global cooling trend starting at 1580 CE is observed in the majority of the 505 reconstructions (PAGES 2k Consortium, 2013). Several paleoclimate reconstructions agreed upon the occurrence of a cold period around 1600 CE, with negative temperature anomalies persisting in Europe 506 at decadal and multi-decadal scales (Ljungqvist et al., 2012; Luterbacher et al., 2016; Masson-Delmotte 507 508 et al., 2013). Reconstructions of European summer temperature provided by Luterbacher et al. (2016) 509 indicate that the coldest 11 and 51 year period since 755 CE in the area of Han-sur-Lesse cave occurred 510 within the  $17^{\text{th}}$  century. These reconstructions showed a summer temperature decrease of  $1 - 1.5^{\circ}$ C around 1600-1650 CE. Although the 17<sup>th</sup> century has been recognized as the coldest of the past twelve 511





centuries, hydrological climate conditions appear close to the long-term mean (Ljungqvist et al., 2016), with no significant wetting or drying trend. However, to account for the differences between the 1960-2010 interval and the 17<sup>th</sup> century observed in this study, an increase in the amount winter precipitation is needed, suggesting that climatic conditions were wetter during that time. Such a hypothesis is also supported by the depleted  $\delta^{18}$ O values in P16, indicating an increase in winter precipitation.

### 518 6. Conclusions

This study of annual trace element and stable isotope ( $\delta^{13}$ C and  $\delta^{18}$ O) variations over three different 519 520 time intervals of the annually laminated Proserpine stalagmite from the Han-sur-Lesse Cave (Belgium) 521 shows that seasonal changes in Mg, Sr and Ba during the recent period (1960-2010) suggest a strong 522 effect of prior calcite precipitation, caused by lower water availability during summer. In the 17<sup>th</sup> 523 century (1600 CE ± 30 and 1640 CE ± 30), however, Mg is in antiphase with Sr and Ba. This implies that 524 another process overwrites the PCP dominated seasonal cycle in these trace elements. A varying 525 degree of incongruent dolomite dissolution is the most plausible hypothesis, with more dissolution 526 occurring during summer when water residence times in the epikarst are longer. The transfer function governing the trace elements, PCP or a varying degree of dolomite dissolution, depends on water-rock 527 interaction. Stable isotope ratios ( $\delta^{13}$ C and  $\delta^{18}$ O), soil derived trace element concentrations (Zn, Y and 528 529 Pb) and speleothem morphology indicate that the multi-annual recharge of the epikarst was higher in 530 the 17<sup>th</sup> century. The change in the response of Mg, Sr and Ba in the Proserpine speleothem to environmental changes was identified to be climate-driven and likely results from a recharge increase 531 532 caused by a combination of lower summer temperatures and an increase in the amount of winter precipitation in the 17<sup>th</sup> century for the Han-sur-Lesse cave region. The effect of an increase in winter 533 annual precipitation on the recharge is expected to be larger compared to a decrease in summer 534 535 temperature. The data obtained in this study clearly shows a stronger seasonal cycle in cave hydrology during the 17<sup>th</sup> century. 536

This high-resolution, multi proxy study provides a good example of how seasonal proxy transfer functions of trace elements in speleothem calcite can change over time. Such an observation has implications for future speleothem-based paleoclimate reconstructions, since transfer functions for specific cave sites, determined by cave monitoring, are often assumed to remain constant when no drastic changes in the cave environment have occurred. As the change in trace element proxy transfer function observed in this study is climate-driven, this change by itself can serve as a valuable paleoclimate proxy.





### 545 Author contributions

546 Stef Vansteenberge and Sophie Verheyden designed the study. Stef Vansteenberge, Steven Goderis 547 and Stijn Van Malderen carried out LA-ICP-MS measurements. Stef Vansteenberge, Matthias Sinnesael 548 and Niels de Winter carried out stable isotope measurements. Stef Vansteenberge carried out the data 549 processing and plotting with contributions from Steven Goderis, Niels de Winter and Matthias 550 Sinnesael. Frank Vanhaecke and Philippe Claeys provided laboratory facilities and supported the 551 measurements. Stef Vansteenberge, Niels de Winter and Matthias Sinnesael prepared the manuscript 552 with contributions from all co-authors.

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563 Table 1

Isotope	<sup>25</sup> Mg	<sup>27</sup> Al	<sup>29</sup> Si	<sup>31</sup> P	<sup>34</sup> S	<sup>39</sup> K	<sup>55</sup> Mn	<sup>57</sup> Fe 564
LOD (µg g <sup>-1</sup> )	4.0	9.0	100	1.0	7.0	7.0	0.08	4.0
Isotope	<sup>66</sup> Zn	<sup>85</sup> Rb	<sup>88</sup> Sr	<sup>89</sup> Y	<sup>137</sup> Ba	<sup>208</sup> Pb	<sup>232</sup> Th	<sup>238</sup> U <sup>565</sup>
LOD (µg g <sup>-1</sup> )	0.2	0.03	0.08	0.01	0.1	0.008	0.0005	0.0 <u>Q01</u>

567 Table 1: Overview of limits of detection (LOD) of trace elements measured for this study using LA-ICP-

568 MS.

569 Table 2

mm       Thick: 1.096 mm         tions (RSD = 28.9%)       Smaller variations (RSD = 6.3%)         onality: tendency towards in       Weak to no seasonality: unclear relation         lation with $\delta^{13}$ C       Weak to no seasonality: unclear relation         controlled, but other       With $\delta^{13}$ C       Scole:         swell       Clear $\delta^{13}$ C cycle:       Clear $\delta^{13}$ C cycle:         orticled, but other       S <sup>13</sup> C driven by seasonal changes in PCP         by seasonal changes in PCP       Phase relation not clear         by seasonal changes in PCP       Phase relation not clear         nse correlation       Transition period between P16 and P19         PCP       Phase relation not clear         nality in Zn and Y,       Weak seasonality in Zn, Y and Pb         onality in Pb       Decreased flushing         onality in Pb       Decreased flushing         onality antiphase with Mg,       Weak seasonality antiphase with Sr and         and ity, antiphase with Mg,       Ba         ore elements and layering is       Link with trace elements and layering is         tool, scavenging       Link with trace elements and layering is         cole       Link with trace elements and layering is	Proxy	P19	617	P16
Larger variations (RSD = 28.9%)       Smaller variations (RSD = 6.3%)         Strong seasonality: tendency towards in phase correlation with $\delta^{13}$ C       Weak to no seasonality: unclear relation with $\delta^{13}$ C         Partially T-controlled, but other processes as well       Weak to no seasonality: unclear relation with $\delta^{13}$ C         Dow S <sup>13</sup> C cycle:       Lear $\delta^{13}$ C cycle:         Low $\delta^{13}$ C driven by seasonal changes in PCP       S <sup>13</sup> C driven by seasonal changes in PCP         S <sup>13</sup> C driven by seasonal changes in PCP       S <sup>13</sup> C driven by seasonal changes in PCP         Mg, Sr and Ba driven by seasonal changes in PCP       Phase relation not clear         Mg, Sr and Ba driven by seasonal       Transition period between P16 and P19         Mg, Sr and Ba driven by seasonality in Zn, Y and Pb       Phase relation not clear         Mg, Sr and Ba driven by seasonality in PD       Phase relation not clear         Mg, Sr and Ba driven by seasonality in Zn, Y and Pb       Phase relation not clear         Mg, Sr and Ba       Phase relation not clear         Mg, Sr and Ba driven by seasonality in PD       Phase relation not clear         Mg, Sr and Ba       Strong seasonality in PD         Strong seasonality in PD       Decreased flushing         Decreased flushing       Decreased flushing         Decreased flushing       Decreased flushing         Decreased flushing       Strong		Thin: 0.382 mm	Thick: 1.096 mm	Thick: 1.135 mm
Strong seasonality: tendency towards in phase correlation with $\delta^{13}$ CWeak to no seasonality: unclear relation with $\delta^{13}$ CPartially T-controlled, but other processes as wellWeak to no seasonality: unclear relation with $\delta^{13}$ C cycle: Low $\delta^{13}$ C mostly in DCL but not always $\delta^{13}$ C mostly in DCL but not always $\delta^{13}$ C driven by seasonal changes in PCPGood in phase correlation Mg, Sr and Ba driven by seasonal changes in PCPMeak seasonal changes in PCPWeak seasonality in ZhWeak seasonality in Zh, Y and Pb hydrological regimesWeak seasonality in Pb Strong seasonality in Pb Becreased flushingWeak seasonality in Zh, Y and Pb Decreased flushingStrong seasonality in Pb Strong seasonality antiphase with Mg.Weak seasonality in Zh, Y and Pb Becreased flushingStrong seasonality in Pb Strong seasonality antiphase with Mg.Weak seasonality antiphase with Sr and BaStrong seasonality in Pb Strong seasonality antiphase with Mg.Weak seasonality antiphase with Sr and BaLink with trace elements and layering is challengingLink with trace elements and layering is challenging	AVELABE LAYEL LINCKITESS	Larger variations (RSD = 28.9%)	Smaller variations (RSD = 6.3%)	Smaller variations (RSD = 9.5%)
phase correlation with δ <sup>13</sup> C       with δ <sup>13</sup> C         Partially T-controlled, but other       but other         processes as well       Clear δ <sup>13</sup> C cycle:         Low δ <sup>13</sup> C mostly in DCL but not always       DClar δ <sup>13</sup> C cycle:         Low δ <sup>13</sup> C mostly in DCL but not always       DClar δ <sup>13</sup> C cycle:         δ <sup>13</sup> C driven by seasonal changes in PCP       Bi <sup>32</sup> C driven by seasonal changes in PCP         Good in phase correlation       Phase relation not clear         Mg, Sr and Ba driven by seasonal       Phase relation not clear         Mg, Sr and Ba driven by seasonal       Phase relation not clear         Mg, Sr and Ba driven by seasonal       Phase relation not clear         Mg, Sr and Ba driven by seasonal       Phase relation not clear         Mg, Sr and Ba driven by seasonal       Phase relation not clear         Mg, Sr and Ba driven by seasonal       Phase relation not clear         Mg, Sr and Ba       Perceased flushing         Changes in PCP       Weak seasonality in Zn, Y and Pb         Strong seasonality in Pb       Decreased flushing         Decreased flushing, anthropogenic Pb       Perceased flushing         Strong seasonality, antiphase with Mg, Sr and Ba       Sr and Ba         Strong seasonality, antiphase with Mg, Sr and Ba       Ba         No PCP control, scavenging       Chalenging <td></td> <td>Strong seasonality: tendency towards in</td> <td>Weak to no seasonality: unclear relation</td> <td>Weak to no seasonality: unclear relation</td>		Strong seasonality: tendency towards in	Weak to no seasonality: unclear relation	Weak to no seasonality: unclear relation
Partially T-controlled, but other processes as well         Clear $\delta^{13}$ C cycle:           Clear $\delta^{13}$ C mostly in DCL but not always         Clear $\delta^{13}$ C cycle:           Low $\delta^{13}$ C mostly in DCL but not always $\delta^{13}$ C always in DCL $\delta^{13}$ C driven by seasonal changes in PCP         Clear $\delta^{13}$ C driven by seasonal changes in PCP           Good in phase correlation $\delta^{13}$ C driven by seasonal changes in PCP           Good in phase correlation         Phase relation not clear           Mg, Sr and Ba driven by seasonal         Phase relation not clear           Mg, Sr and Ba driven by seasonal         Phase relation not clear           Mg, Sr and Ba driven by seasonal         Phase relation not clear           Mg, Sr and Ba driven by seasonal         Phase relation not clear           Mg, Sr and Ba driven by seasonal         Phase relation and Pla           Neak seasonality in Pb         Photrological regimes           Strong seasonality in Pb         Decreased flushing           Decreased flushing, anthropogenic Pb         Perceased flushing           Ron PCP control, scavenging         Strong seasonality antiphase with Mg.           Sr and Ba         Ba           No PCP control, scavenging         Link with trace elements and layering is challenging           Link with trace elements and layering is challenging         Challenging	\$180	phase correlation with $\delta^{13}C$	with $\delta^{13}$ C	with $\delta^{13}$ C
processes as well         processes as well           Clear $\delta^{13}$ C cycle:         Clear $\delta^{13}$ C cycle:           Low $\delta^{13}$ C mostly in DCL but not always         DCL $\delta^{13}$ C driven by seasonal changes in PCP         Si <sup>32</sup> C driven by seasonal changes in PCP           Good in phase correlation         Phase relation not clear           Mg, Sr and Ba driven by seasonal         Phase relation not clear           Mg, Sr and Ba driven by seasonal         Phase relation not clear           Mg, Sr and Ba driven by seasonal         Phase relation not clear           Mg, Sr and Ba driven by seasonal         Phase relation not clear           Mg, Sr and Ba driven by seasonal         Phase relation of clear           Mg, Sr and Ba driven by seasonal         Phase relation and clear           Mg, Sr and Ba driven by seasonal         Phase relation and clear           Mg, Sr and Ba         Pocreased flushing           Decreased flushing, anthropogenic Pb         Decreased flushing           Strong seasonality, antiphase with Mg, Sr and Ba         Sr and Ba           Strong seasonality, antiphase with Mg, Ba         Mo PCP control, scavenging           Strand Ba         Link with trace elements and layering is challenging	00	Partially T-controlled, but other		
Clear $\delta^{13}$ C cycle:       Clear $\delta^{13}$ C mostly in DCL but not always         Low $\delta^{13}$ C mostly in DCL but not always $\delta^{13}$ C always in DCL $\delta^{13}$ C driven by seasonal changes in PCP $\delta^{13}$ C driven by seasonal changes in PCP         Good in phase correlation $\delta^{13}$ C driven by seasonal changes in PCP         Good in phase correlation       Phase relation not clear         Mg, Sr and Ba driven by seasonal       Phase relation not clear         Mg, Sr and Ba driven by seasonal       Phase relation not clear         Mg, Sr and Ba driven by seasonal       Phase relation not clear         Mg, Sr and Ba driven by seasonal       Phase relation period between P16 and P19         Neek seasonality in PD       Phase relation period between P16 and P19         Veak seasonality in PD       Weak seasonality in Zn, Y and Pb         Strong seasonality in PD       Decreased flushing         Decreased flushing, anthropogenic Pb       Decreased flushing         Strong seasonality, antiphase with Mg.       Strand Ba         Strong seasonality, antiphase with Mg.       Ba         No PCP control, scavenging       Link with trace elements and layering is challenging         Link with trace elements and layering is challenging       Link with trace elements and layering is challenging		processes as well		
Low $\delta^{13}$ C mostly in DCL but not always       Low $\delta^{13}$ C always in DCL         8 <sup>13</sup> C driven by seasonal changes in PCP       S <sup>13</sup> C driven by seasonal changes in PCP         Good in phase correlation       Ranse relation not clear         Mg, Sr and Ba driven by seasonal       Phase relation not clear         Mg, Sr and Ba driven by seasonal       Phase relation not clear         Mg, Sr and Ba driven by seasonal       Phase relation not clear         Mg, Sr and Ba driven by seasonal       Phase relation not clear         Mg, Sr and Ba driven by seasonal       Phase relation and clear         Mg, Sr and Ba driven by seasonal       Phase relation and clear         Mg, Sr and Ba driven by seasonal       Phase relation and clear         Weak seasonality in Pb       Nveak seasonality in Zn, Y and Pb         Strong seasonality in Pb       Decreased flushing         Decreased flushing, anthropogenic Pb       Ba         enrichment       Weak seasonality antiphase with Sr and Ba         Strong seasonality, antiphase with Mg, Ba       Ba         No PCP control, scavenging       Link with trace elements and layering is challenging         Link with trace elements and layering is challenging       Clink with trace elements and layering is challenging		Clear $\delta^{13}$ C cycle:	Clear $\delta^{13}$ C cycle:	Clear $\delta^{13}$ C cycle:
Stad driven by seasonal changes in PCP       Stad driven by seasonal changes in PCP         Good in phase correlation       Fhase relation not clear         Mg, Sr and Ba driven by seasonal       Phase relation not clear         Mg, Sr and Ba driven by seasonal       Phase relation not clear         Mg, Sr and Ba driven by seasonal       Phase relation not clear         Mg, Sr and Ba driven by seasonal       Phase relation period between P16 and P19         Meak seasonality in Zn       Weak seasonality in Zn, Y and Pb         Strong seasonality in Pb       Decreased flushing         Decreased flushing, anthropogenic Pb       Decreased flushing         Strong seasonality, antiphase with Mg, Sr and Ba       Ba         No PCP control, scavenging       scavenging         Link with trace elements and layering is       Link with trace elements and layering is         Link with trace elements and layering is       Link with trace elements and layering is	8 <sup>13</sup> C	Low $\delta^{13}$ C mostly in DCL but not always	Low $\delta^{13}$ C always in DCL	Low $\delta^{13}$ C always in DCL
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Mg, Sr and Ba driven by seasonal       Transition period between P16 and P19         changes in PCP       hydrological regimes         changes in PCP       weak seasonality in Zn, Y and Pb         Weak seasonality in Pb       Weak seasonality in Zn, Y and Pb         Strong seasonality in Pb       Decreased flushing         Decreased flushing, anthropogenic Pb       Decreased flushing         Strong seasonality, antiphase with Mg,       Weak seasonality antiphase with Sr and         Strong seasonality, antiphase with Mg,       Ba         No PCP control, scavenging       Link with trace elements and layering is         Link with trace elements and layering is       Link with trace elements and layering is		Good in phase correlation	Phase relation not clear	Anti-phase correlation between Mg and
changes in PCP       hydrological regimes         Weak seasonality in Zn and Y,       Weak seasonality in Zn, Y and Pb         Strong seasonality in Pb       Weak seasonality in Zn, Y and Pb         Decreased flushing, anthropogenic Pb       Decreased flushing         Estrong seasonality, antiphase with Mg,       Weak seasonality antiphase with Sr and Ba         Strong seasonality, antiphase with Mg,       Ba         No PCP control, scavenging       Link with trace elements and layering is challenging		Mg, Sr and Ba driven by seasonal	Transition period between P16 and P19	Sr, Ba
Weak seasonality in Zn and Y, Strong seasonality in PbWeak seasonality in Zn, Y and PbDecreased flushing, anthropogenic Pb enrichmentDecreased flushing Strong seasonality, antiphase with Mg, 	NIS allu 31 - Da	changes in PCP	hydrological regimes	Seasonally occurring IDD dominates
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Strong seasonality in Pb       Decreased flushing         Decreased flushing, anthropogenic Pb       Decreased flushing         Decreased flushing, anthropogenic Pb       Neak seasonality antiphase with Sr and Sr and Ba         Strong seasonality, antiphase with Mg, Sr and Ba       Weak seasonality antiphase with Sr and Ba         No PCP control, scavenging       scavenging         Link with trace elements and layering is challenging       challenging		Weak seasonality in Zn and Y,	Weak seasonality in Zn, Y and Pb	Very strong seasonality in Zn and Y,
Decreased flushing, anthropogenic Pb         Decreased flushing, anthropogenic Pb           enrichment         Weak seasonality antiphase with Sr and Sr and Ba           Sr and Ba         Ba           No PCP control, scavenging         scavenging           Link with trace elements and layering is challenging         challenging		Strong seasonality in Pb	Decreased flushing	weak seasonality in Pb
enrichment     weak seasonality antiphase with Sr and       Strong seasonality, antiphase with Mg,     Weak seasonality antiphase with Sr and       Sr and Ba     Ba       No PCP control, scavenging     scavenging       Link with trace elements and layering is     Link with trace elements and layering is       challenging     challenging		Decreased flushing, anthropogenic Pb		Enhanced flushing
Strong seasonality, antiphase with Mg,       Weak seasonality antiphase with Sr and         Sr and Ba       Ba         No PCP control, scavenging       scavenging         Link with trace elements and layering is       Link with trace elements and layering is         challenging       challenging		enrichment		
Sr and Ba No PCP control, scavenging scavenging Link with trace elements and layering is challenging challenging challenging		Strong seasonality, antiphase with Mg,	Weak seasonality antiphase with Sr and	No seasonality
No PCP control, scavenging         scavenging           Link with trace elements and layering is         Link with trace elements and layering is           challenging         challenging	D	Sr and Ba	Ba	
Link with trace elements and layering is Link with trace elements and layering is challenging challenging		No PCP control, scavenging	scavenging	
challenging	Demarks	Link with trace elements and layering is	Link with trace elements and layering is	Link with trace elements and layering is
		challenging	challenging	challenging

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# Table 2: Schematic overview providing the observed changes and interpretation for the different proxies of P19, P17 and P16. PCP = prior

calcite precipitation, IDD = incongruent dissolution of dolomite, DCL = dark compact layers, WPL = white porous layers

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# 574 FIGURES



575

Figure 1: (left) Location of the Han-sur-Lesse Cave system (N50.114251, E5.203342) with the entrance
and exit of the Lesse River, the Salle-du-dome and the Père-Noël Cave. North is upwards (right) Map
showing the location of the Proserpine stalagmite within the Salle-du-Dome. The insert shows the
position of the core retrieved from the speleothem. Images adapted from Van Rampelbergh et al.
(2014, 2015).







### 582

**Figure 2:** A. Overview of long (~500 yr) record of stable isotope ratios and annual layer thickness through the Proserpine speleothem measured by Van Rampelbergh et al. (2015). Red boxes indicate the locations of high-resolution transects discussed in this study. B. The three studied growth periods P19 (1960-2010 CE), P17 (1633-1644 ± 30 CE) and P16 (1593-1605 ± 30 CE). The yellow rectangles mark the sections that were drilled/sampled for  $\delta^{13}$ C and  $\delta^{18}$ O analysis, the red lines represent the LA-ICP-MS transects. Numbers in grey indicate the observed layer couplets.







590

591 Figure 3: Ranges of the stable isotope (left) and trace element data (right). For the stable isotope

592 ratios, the data mark the average (black diamonds) and the standard deviation (1σ) of the

593 distribution. For the trace element concentrations, the boxes represent the minimum and maximum

values and the white diamonds mark the median. Numbers on top of the bars represent the

595 percentage of the data that is below the calculated detection limit.







598Figure 4: Periodograms (FFT) of  $\delta^{13}$ C, Mg, Zn and P measured in P16 to illustrate how the quality of a599proxy to record the seasonal cycle can be studied. The red line represents the 95% confidence level.600 $\delta^{13}$ C is taken as a reference. The periodograms include two examples of proxies with a distinct peak in601the seasonal frequency band of 0.8 mm<sup>-1</sup>(Mg and Zn) and one proxy with no peak in the seasonal602frequency band. Periodograms for all periods are provided in the supplementary material.

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Figure 5: Stable isotope ratios and trace element variations plotted against distance for P19, P17 and
 P16. Blue bars mark the DCC laminae. The left side represents the youngest layers. All stable isotope
 ratios are expressed as ‰ VPDB, while trace element concentrations are reported in ppm. Red bars
 indicate years used for annual stack (Fig. 5 and 6).















615 **Figure 7:** Annual stacks of  $\delta^{13}$ C (black) and  $\delta^{18}$ O (red). Dashed lines mark the 2σ uncertainty. The x-axis

616 represents one year. For the years used, see Fig. 5.

617







Figure 8: Chart showing the calculated theoretical amount of water excess calculated with the
Thornthwaite equation (Thornthwaite and Mather, 1957), based on temperature and precipitation
data near Han-sur-Lesse cave from 1999 to 2012 (Royal Meteorological Institute, KMI). X-axis
represents the months from January to December.

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