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Pathways of sea surface paleo-CO₂ off northern Chile

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Surface circulation patterns and the pathways of sea surface carbon dioxide (CO₂) off northern Chile (~27.5° S) between 30 and 10 kyr BP: global and/or local forcing?

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Abstract

We present a reconstruction of past changes in partial pressure of CO₂ ($p\text{CO}_2$) from northern Chile ($\sim 27^\circ\text{S}$), between 10 and 30 kyr BP, based on carbon isotope composition of C_{37:2}-alkenone. The high- $p\text{CO}_2$ during the entire time series indicates that northern Chile upwelling system has been a permanent source of CO₂ to the atmosphere. The multiproxy reconstruction suggests that the CO₂ outgassing and sequestration pathways were modulated by local and global mechanisms. During global glacial conditions, an enhanced coastal upwelling forcing resulted in high-availability of deep water macronutrients and a CO₂-supersaturated water column, which combined with high-inputs of iron from the continent, intensified the carbon sequestration pathway of the biological pump, through diatom biomass export. During the deglacial, a decrease in the upwelling forcing, an increment in water column stability and reduced continental inputs of iron are consistent with a larger role of calcifying organisms in the plankton assemblage in terms of carbon sequestration pathway through the carbonate system.

1 Introduction

Coastal upwelling areas are often described as intense sources of CO₂ and macronutrients (e.g., Borges et al., 2005). The CO₂-supersaturated and high-macronutrient concentration of upwelled waters stimulate intense blooms of phytoplankton that sequester inorganic carbon thus reducing the outgassing of CO₂ (e.g., Simpson and Zirino, 1980).

Large parts of the ocean have been described as High-Nutrient/Low-Chlorophyll (HNLC) areas, where phytoplankton growth is limited by the availability of iron (e.g., Martin, 1990; de Baar and Boyd, 2000; Bruland et al., 2005). Experimental evidence suggests that the availability of dissolved iron permits the rapid interception of upwelled CO₂ by phytoplankton in oceanic (e.g., Bakker et al., 2001) and coastal waters (Torres

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and Ampuero, 2009). High-levels of dissolved iron in coastal upwelled waters are determined by the input of iron into the photic layer from other sources (e.g., dust deposition, river discharges, continental shelf) in addition to the initial dissolved iron contained in the water masses that feed the upwelling (Johnson et al., 1999; Bruland et al., 2005).

5 Upwelling favorable winds are permanent at $\sim 27^\circ$ S off Chile during all year (Shaffer et al., 1999). In this area of narrow continental shelf, the recently upwelled water is characterized by low-temperature, high-nutrient concentrations, low-chlorophyll and supersaturation in CO_2 (e.g., Torres et al., 1999; Torres and Ampuero, 2009). At these windy latitudes (27° – 30° S) (Shaffer et al., 1999), the supersaturation in CO_2 allows for
10 strong- CO_2 fluxes from the ocean to the atmosphere (e.g., Torres et al., 2002). Vertical fluxes of organic matter and water column stratification are key factors that explain the intensity/extension of aerobic and anaerobic production of CO_2 . The intense denitrification reported for upwelling areas off Peru and Chile highlights an additional input of CO_2 associated with anaerobic respiration of organic matter (e.g., Friederich et al.,
15 2008; Paulmier et al., 2008).

Paleoceanographic reconstructions off northern Chile described a strong precessional cyclicity in the pattern of iron flux, suggesting that iron availability is a key factor for past changes in productivity and sequestration of atmospheric CO_2 (Lamy et al., 1998; Dezileau et al., 2004). This was related to increased precipitation during precessional maxima, higher river runoff transporting iron-rich terrigenous material to the ocean inducing high-biological productivity (i.e., diatom blooms) and consequently, a rapid decrease in sea surface water $p\text{CO}_2$ (Dezileau et al., 2004). In contrast, low-precipitation during precessional minima decreased the iron supply to the ocean, thus limiting the primary productivity and enhancing the supersaturation of CO_2 in the surface water column (Dezileau et al., 2004).
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Another study suggested that the rapid melting of the most northerly extension of the Patagonian Ice Sheet (Lamy et al., 2004) enhanced water column stratification during the deglacial (De Pol-Holz et al., 2006), which could be translated into reduced physical ventilation of the intermediate water masses (De Pol-Holz et al., 2007).

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This study is a first approximation of millennial $p\text{CO}_2$ changes in the eastern South Pacific (ESP) margin between 30 and 10 kyr BP, based on carbon isotope composition of $\text{C}_{37:2}$ -alkenone. Our results reveal that important changes in the marine CO_2 outgassing were modulated by local and global mechanisms, including water column stability, iron-limitation, and consequently the productivity patterns, which potentially control the different pathways of removal of the marine CO_2 from surface waters to the deep-ocean.

2 Material and methods

We used deep-sea sediment core GeoB 3375-1 (27°28' S, 71°15' W; 1947 m water depth; core length 489 cm), collected from the Chilean margin. Sediment samples were sub-sampled at 5 cm, freeze-dried and extracted for alkenone analyses according to the methodology described by Prahl et al. (1988) and Harada et al. (2003).

1–2 grams of freeze-dried sediment was ground into a powder with an agate mortar and pestle. The bulk organic fractions were extracted from sediment samples with an accelerated solvent extractor (ASE-200, Dionex Japan, Ltd.) at 100 °C and 1000 psi using a mixture of dichloromethane and methanol (99:1 v/v) as the solvent. The yield of the extraction procedure was >95% for a ketone standard (2-nonadecanone).

The extracts were saponified in 0.5 M KOH in methanol at 80 °C for 2 h. The neutral fraction was recovered with a pipette, dissolved in hexane, and then separated into subfractions by silica gel column chromatography using an automatic solid preparation system (Rapid Trace SPE Workstation, Zymark, UK). Standards for the subfractionation were hydrocarbons (C_{20} – C_{38} hydrocarbons and 5 α -cholestane, GLSciences Inc., Japan), a ketone (2-nonadecanone, Fluka Chemie, Switzerland), and a sterol (cholesterol, GLSciences Inc., Japan). The solvents used were 4 mL of hexane for fraction 1; a mixture of 2 mL of a hexane-toluene mixture (3:1 v/v), 2 mL of hexane-toluene (1:1 v/v), 2 mL of hexane-ethyl acetate (95:5 v/v), and 2 mL of hexane-ethyl acetate (9:1 v/v) for fraction 2; 2 mL of hexane-ethyl acetate (85:15 v/v) followed by 2 mL of hexane-ethyl

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acetate (4:1 v/v) for fraction 3; and 2 mL of hexane-ethyl acetate (4:1 v/v) followed by 2 mL of ethyl acetate for fraction 4. The alkenones eluted in fraction 3. The recovery of a 2-nonadecanone standard with this procedure was >90%. A sample of the alkenone fraction was analyzed by capillary gas chromatography with a Fisons 8000 series gas chromatograph (CE Instruments, USA) equipped with a 50 m long, 0.32 mm internal diameter Chrompack CP-Sil 5CB fused silica column, a cold on column injector, and a flame ionization detector. The temperature regime for gas chromatography was as follows: held at 60 °C for 1 min, increased to 200 °C at 20 °C min⁻¹, increased to 305 °C at 15 °C min⁻¹, held at 305 °C for 32 min, increased to 320 °C at 15 °C min⁻¹, and held at 320 °C for 5 min.

$U_{37}^{K'}$ -index calculations were conducted according to Prahl and Wakeham (1987), and $U_{37}^{K'}$ values were converted to alkenone-based temperature ($U_{37}^{K'}\text{-SST}$) using the equation of Prahl et al. (1988), i.e., $U_{37}^{K'} = 0.034T + 0.039$, where T = temperature (°C).

The C_{37:2}-alkenone isotope analyses were performed using a HP6890 gas chromatography unit and a Finnigan MAT 252 (Thermo Electron Co. Bremen, Germany) mass spectrometer unit. The ¹³C/¹²C isotopic ratio was also measured in a *n*-C₃₆ hydrocarbon (GLSciences Inc., Japan) and National Bureau of Standards standard 19 (International Atomic Energy Agency) as an internal standard. The analytical accuracy of $\delta^{13}\text{C}_{\text{C}_{37:2}\text{-alkenone}}$ was $\pm 0.05\%$.

Alkenone-CO₂ estimates were based on the expression:

$$\varepsilon_{\text{p}32:7} = \varepsilon_f - b / [\text{CO}_{2\text{aq}}] \quad (1)$$

where $\varepsilon_{\text{p}32:7}$ values are based on the isotopic difference between an estimate of the $\delta^{13}\text{C}$ values of the organic matter produced by ancient algae ($\delta^{13}\text{C}_{\text{org}}$) and the co-existing dissolved CO₂ in the photic zone ($\delta^{13}\text{C}_{\text{CO}_{2\text{aq}}}$) (Freeman and Pagani, 2005).

$\delta^{13}\text{C}_{\text{org}}$ is determined by adding 4.2‰ to the measured $\delta^{13}\text{C}_{37:2}$ values, as a correction that reflects lipid-biomass isotopic differences between *Haptophyte* biomass and C_{37:2}-alkenone, with alkenones depleted relative to total organic biomass (e.g., Bidi-



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gare et al., 1997; Popp et al., 1998; Pagani et al., 1999). $\delta^{13}\text{C}_{\text{CO}_2\text{aq}}$ is reconstructed from the carbon isotope composition of planktic foraminifera, which assumes that isotopic and chemical equilibrium prevailed among all the carbon aqueous species and atmospheric CO₂ as well as foraminiferal calcite (e.g., Pagani et al., 1999). We have selected the median of the estimate for ε_f (26‰) with an uncertainty of 1‰, derived from chemostat culture experiments (e.g., Laws et al., 1995; Popp et al., 1998). The b value represents an integration of all the physiological variables that affect the total carbon isotope fractionation during photosynthesis (e.g., growth rate and cell geometry (Jasper et al., 1994; Bidigare et al., 1997). In this study, the b-value was calculated using the geometric mean regression of all available data (Bidigare et al., 1997, 1999; Eek et al., 1999; Popp et al., 1999; Laws et al., 2001) and mean annual phosphate concentration from the NOAA World Ocean Atlas 2001 (Conkright et al., 2002) for the core-site location (27.5° S).

Values of CO_{2aq} were converted to pCO₂ by applying Henry's Law (Weiss, 1974), calculated assuming an annual mean salinity of 34.6 psu (Conkright et al., 2002) for the core-site (27.5° S), and alkenone-based temperature (Lyle et al., 1992).

The age model was based on seven ¹⁴C AMS dates and linear interpolation between the age control points in the upper half of the core (from 40 to 9 kyr BP) (see Mohtadi and Hebbeln, 2004).

3 Results and discussion

3.1 Sea surface temperature reconstruction and possible oceanographic scenarios

At 27.5° S off Chile, the $U_{37}^{K'}$ -SST of core GeoB 3375-1 exhibits a millennial-scale variation of the order of ~4°C during the last 30 kyr BP (Fig. 1a). The core highlights an overall warming trend from 17°C to 20°C between ~30 and 17 kyr BP (Fig. 1a). This warming trend was interrupted by cold-events, at ~27, 23 and 16–17 kyr BP. The

latter was a very abrupt cooling of almost 4°C which was followed by a marked increase in SST during the deglacial. Alkenone-SST and the $\delta^{18}\text{O}$ -*N. pachyderma* (*sin.*) records in core GeoB 3375-1 are similar between 30 and 20 kyr BP in terms of general pattern and cold events, and between 15 and 10 kyr BP with respect to the deglacial warming (Fig. 1a, b). They differ markedly in the interval 20–15 kyr BP: the alkenone-SST warming trend (of ~1.5°C) around 19–17 kyr BP and the large and abrupt 16–17 kyr BP cooling are not seen in the $\delta^{18}\text{O}$ record. This difference between surface waters (recorded by alkenones which are biosynthesized by some algae of the class *Haptophyceae*; e.g., Prah et al., 1988) and deeper waters (recorded by *N. pachyderma* (*sin.*) assumed to calcify at depth below the pycnocline; e.g. Andersson et al., 2009) may point to enhanced stratification of the water column during this time. This was previously proposed by De Pol-Holz et al. (2006) who invoked a collapse of thermocline ventilation due to reduced subduction rates of Eastern South Pacific Intermediate Water (ESPIW) triggered by the melting of the Patagonian Ice Sheet, promoting nitrogen removal that potentially contributed to the changes in atmospheric CO₂ during the deglacial (De Pol-Holz et al., 2007).

The deglacial warming (between ~15 and 10 kyr BP, Fig. 1a and b) indicates substantial changes in subsurface water conditions in the ESP with an increased southward advection of the tropical-sourced subsurface waters, suggesting a reorganization of the system that lags the onset of the initial SST warming recorded in core GeoB3375 (Fig. 1a) and other cores off northern and central Chile (Mohtadi and Hebbeln, 2004). The reorganization included reestablishing upwelling and ventilation, with the consequence of a mixed and CO₂-supersaturated water column (Fig. 1c).

On the other hand, the Last Glacial Maximum (LGM) is not observed in the alkenone-SST and $\delta^{18}\text{O}$ records. Previous studies suggested that during the coldest events (e.g., LGM), the SST gradients between 30° S and the equator seem to have been relatively weaker due to a stronger Peru-Chile Current (PCC) (e.g., Kaiser et al., 2005). In fact, the northward displacement of the oceanographic fronts during the LGM in the ESP (Kaiser et al., 2005) could have increased the upwelling strength, and also

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the formation rates of ESPIW and ventilation of the shallow oceanic layer off northern Chile (De Pol-Holz et al., 2006). On this basis, the enhanced upwelling and ventilation during the LGM would be reflected in a CO₂-supersaturated water column (Fig. 1c, see also discussion next section).

3.2 The pathways of sea surface carbon dioxide ($p\text{CO}_2$)

The measured C_{37:2}-alkenone $\delta^{13}\text{C}$ values ranged from -29.5‰ to -20.5‰ , and the resulting isotopic fractionation (ϵ_p) and sea surface $p\text{CO}_2$ ranged from 5.7‰ to 15.9‰ (Table 1S, data supplement: <http://www.clim-past-discuss.net/6/347/2010/cpd-6-347-2010-supplement.pdf>) and from $200.7\ \mu\text{atm}$ to $415.7\ \mu\text{atm}$, respectively, for the timeframe 30 to 10 kyr BP (Fig. 1c). The sea surface $p\text{CO}_2$ showed a notable decoupling with the atmospheric CO₂ record (e.g., Indermühle et al., 2000) during the last glacial and the early deglacial (Fig. 1c and d). The sea surface $p\text{CO}_2$ displayed a gradual increase between ~ 30 and 28 kyr BP (from 250 to $350\ \mu\text{atm}$), followed by a gradual decrease between ~ 27 and 24.5 kyr BP (from 346 to $235\ \mu\text{atm}$). Between 24 and ~ 17 kyr BP, while atmospheric CO₂ was relatively low and stable (~ 200 ppmv, Fig. 1d), highest sea surface $p\text{CO}_2$ values (350 – $400\ \mu\text{atm}$, Fig. 1c) suggest that the water column remained supersaturated-CO₂ until stratified conditions stopped ventilation of the water column (De Pol-Holz et al., 2006). Consequently, the $p\text{CO}_2$ gradually declined and this lasted until ~ 14 kyr BP (Fig. 1c). In contrast, atmospheric CO₂ increased steadily from ~ 18 kyr BP to the start of the Holocene (Fig. 1d).

In summary, the sea surface $p\text{CO}_2$ exhibited higher levels than atmospheric CO₂, highlighting that the upwelling system off northern Chile (27.5°S) may have been continuously supersaturated with CO₂ (Fig. 1c and d), and thus would have been an uninterrupted source of CO₂ to the atmosphere during the period 30–10 kyr BP.

The sea surface $p\text{CO}_2$ is affected by changes in temperature, salinity, total CO₂ concentration (i.e., dissolved inorganic carbon DIC), and alkalinity. While the water temperature is primarily regulated by physical processes (i.e., solar energy input, sea-air heat exchanges, and mixed-layer thickness), the latter two are primarily controlled

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by biological processes (i.e., photosynthesis, respiration and calcification) and by upwelling of subsurface waters enriched in respired CO₂ and nutrients (e.g., Zeebe and Wolf-Gladrow, 2001; Anderson et al., 2009).

Given that the changes in the sea surface ocean pCO₂ levels could result in important changes in calcareous and/or organic carbon production rates (due to changes in DIC availability; e.g., Zeebe and Wolf-Gladrow, 2001), we analyzed the potential uptake pathways of marine CO₂ between 30 and 10 kyr BP off northern Chile.

We interpret the record of δ¹³C of planktonic foraminifera *N. pachyderma* (*sin.*) (Fig. 1e) in terms of DIC availability (e.g., Köhler et al., 2005) for carbonaceous organisms instead of using it solely as an upwelling strength proxy, as suggested for this region by Dezileau et al. (2004). The δ¹³C of *N. pachyderma* (*sin.*) record displayed a similar pattern to the carbonate (CaCO₃) fluxes (Fig. 1e and f). In general terms, when δ¹³C is enriched in ¹³C, the carbonate rain rate decreases. The relative abundance of *G. bulloides* (Fig. 1g) also exhibited a similar pattern since the LGM and especially since ~15 kyr BP during the reorganization period when the three variables highlight the highest carbonate production, in contrast to the late glacial (Fig. 1e, f and g).

We consider that the glacial increase in organic carbon export as well as biogenic opal fluxes (Fig. 1i and j) occurs under iron-enriched conditions (Fig. 1h). The response of inorganic/organic to the sea surface CO₂ concentrations (Fig. 2) suggests that organic carbon production was favored during this time. On this basis, the iron availability is a critical factor that determines the diatom biological pump as a pathway of carbon (Fig. 1c) to the deeper waters in the form of diatom organic carbon (Fig. 1i and j). In fact, a local experiment developed in the coastal upwelling areas off Chile evidenced that diatom growth and their capacity to lower the high-pCO₂ of recently-upwelled waters, depend on additional inputs of iron (Torres and Ampuero, 2009).

A large reduction in pCO₂ occurred between ~27 and 24 kyr BP, during the gradual iron-enrichment of the water column, which might be related to the pronounced increase of total organic carbon and biogenic opal fluxes (Fig. 1i and j), suggesting that iron enhanced the capacity of the phytoplankton community, probably dominated by

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diatoms, to rapidly lower the $p\text{CO}_2$ surface levels, activating a strong biological pump able to rapidly reduce the intense CO_2 outgassing. Furthermore, the impoverishment in ^{13}C of the $\delta^{13}\text{C-N. pachyderma (sin.)}$ (Fig. 1e), indicates high-DIC availability, which was destined mainly to the photosynthesis process. Moreover, the response of inorganic/organic carbon to CO_2 concentration (Fig. 2) describes that whereas CO_2 dissolution decreases in the sea surface waters, the organic carbon production remained above carbonate production.

Between ~ 24 and 17 kyr BP the water column remained highly supersaturated in CO_2 without significant depletion, due to more $p\text{CO}_2$ released to the atmosphere through the upwelling process. During this period, the high-export of iron-rich terrigenous material (Dezileau et al., 2004) permitted an increase of the total productivity, reflected in the high-organic carbon (Fig. 1i), -biogenic opal (Fig. 1j) and -carbonate (Fig. 1f) fluxes. However, the high-total production in the water column was not enough to counteract the high-concentrations of dissolved CO_2 in the water.

During the deglacial warming (~ 17 – 16 kyr BP, see gray hachured bar in Fig. 1), the well-stratified water column and the collapse of the ESPIW formation (De Pol-Holz et al., 2006) apparently favored the decrease in $p\text{CO}_2$ (Fig. 1c). We expected that surface stratification not only would help maintain primary producers close to the surface (by limiting the depth of turbulent mixing) but also isolate the mixed layer from CO_2 -rich and nutrients-rich water located below the pycnocline. In fact, a slight reduction in organic carbon and opal fluxes is also observed during this period (Fig. 1i and j). On the other hand, all productivity proxies for calcareous organisms displayed a slight increase during this period (Fig. 1f–g), probably favoring a reduction in water column DIC (Fig. 1e).

A simultaneous intensification of the upwelling in the Southern Ocean (SO) would explain the deglacial rise in the atmospheric CO_2 (Fig. 1d) and the high-inputs of nutrients to the SO (Anderson et al., 2009). We would expect that the intense upwelling and high-opal fluxes recorded in the ~ 16 – 10 kyr BP period in the SO (Anderson et al., 2009), would lead to the formation of particularly macro-nutrient rich waters (specifically the

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Subantarctic Mode Water, SAMW). However, the strong-iron limitation associated to deep upwelled waters implies that the waters formed in this period were impoverished in iron and silicates, but enriched in nitrate (Anderson et al., 2009). Therefore, a tendency to develop a secondary silicate limitation (Hutchins and Brunland, 1998; Takeda, 1998; Martin-Jézéquel et al., 2000) while still retaining high-nitrate (Sarmiento et al., 2007). Thus, the upwelling of SAMW (originally formed from upper Circumpolar Deep Water) by wind driven upwelling systems is expected to lead silicate limitation, and consequently non-siliceous functional groups dominated the phytoplankton assemblage during this period, increasing their influence in the biological pump. This scenario is supported by the decoupling recorded between organic carbon (Fig. 1i), total nitrogen (Fig. 1j) and opal fluxes (Fig. 1j), during ~16–10 kyr BP. Certainly, the maximum export of organic carbon during this period (~16–10 kyr BP) corresponds to both a maximum of CaCO₃ (Fig. 1f) and minimum of opal fluxes (Fig. 1j).

The contrast between the trends of biogenic opal and organic carbon fluxes between ~16 and 10 kyr BP off northern Chile could be reconciled by invoking a shift in the phytoplankton communities from diatomaceous to carbonate production. This would have reduced and increased relative contributions of organic carbon and carbonate to total export production, respectively (Fig. 1i and j).

The apparent contrast (i.e., late glacial/deglacial) becomes clear if we consider how the decrease in terrigenous iron inputs (Fig. 1h) affected the nitrate, as well as silicic acid utilization by diatoms, reducing the total nitrogen and opal fluxes (Fig. 1j), and consequently their role in the marine CO₂ drawdown by the biological pump (Fig. 1i).

High-carbonate drawdown (Fig. 1f and g) at warm-SST (Fig. 1a) and high-atmospheric CO₂ (Fig. 1d), suggest that the marine DIC was captured intensively by the carbonate system, enhancing the carbonate precipitation of the calcifying organisms in the plankton assemblage (Fig. 2). On basis of experimental results from Iglesias-Rodriguez et al. (2008), probably, the deglacial rise of the atmospheric CO₂ could also increase calcification rates, and its drawdown to the deep-ocean through the pathway of the carbonate system (Fig. 1f). In fact, the light- $\delta^{13}\text{C}$ of *N. pachyderma*

(*sin.*) signal (Fig. 1e) confirms this hypothesis, since the elevated impoverishment in ^{13}C could be indicative of a major and rapid consumption of DIC by calcifying organisms.

The calcification increment reduces the efficiency of the marine ecosystem to store sea surface CO_2 , due to changes in the buffering capacity of the ocean's carbonate system, as well as to draw down the atmospheric CO_2 (Cermeño et al., 2008). In fact, if (1) calcium carbonate precipitation increases the marine $p\text{CO}_2$ (e.g., Zeebe and Wolf-Gladrow, 2001) and (2) a high- $p\text{CO}_2$ increases the calcification rates (e.g. Iglesias-Rodriguez et al., 2008), a positive feedback will tend to increase the calcification processes. On this basis, only biologically active iron-enrichment conditions would enhance the competitive advantages of diatoms to break the positive feedback and shift the system to capture the marine CO_2 pathway biological pump.

Our findings suggest that the coastal ocean off northern Chile has been a permanent and strong source of CO_2 to the atmosphere between 30 and 10 kyr BP. This CO_2 outgassing has been modulated by local (upwelling strength) and global mechanisms (e.g., ocean circulation, atmospheric CO_2) that have changed the water column properties (e.g., water column stability, macro- and micro-nutrients availability), and consequently driven the role and importance of the biological pump and the marine carbonate system in the sequestration of inorganic carbon to the deep-ocean.

4 Concluding remarks

The sea surface $p\text{CO}_2$ reconstruction indicates that the coastal ocean off northern Chile ($\sim 27.5^\circ\text{S}$) has been a permanent and strong upwelling source of CO_2 to the atmosphere between 30 and 10 kyr BP. The multiproxy reconstruction reveals that the CO_2 outgassing and sequestration was modulated by local and global mechanisms. During the global glacial period, the water column remained highly supersaturated in CO_2 , due to more $p\text{CO}_2$ being released to the atmosphere through upwelling process. Furthermore, the fertilization of iron biologically active and macronutrients intensified

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the carbon sequestration pathway biological pump through diatom biomass export. In contrast, during the deglacial warming the iron-limitation altered significantly the marine CO₂ sequestration pathway biological pump. Warm-SST and high-atmospheric CO₂ conditions suggest that the carbon was captured intensively by the carbonate system, enhancing the carbonate precipitation of the calcifying organisms in the plankton assemblage. Overall findings suggest that biologically active iron-availability conditions controlled the competitive advantages of plankton to shift the system to capture the marine CO₂ pathway biological pump and/or carbonate system off northern Chile in the eastern south Pacific margin between 30 and 10 kyr BP.

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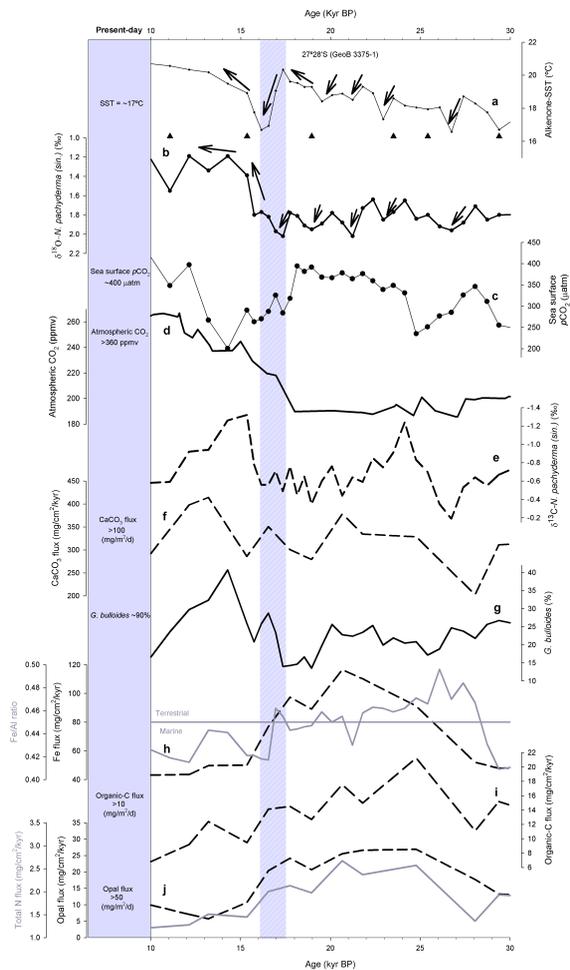


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Fig. 1. Core GeoB3375-1 off northern Chile. A multiproxy reconstruction between 30 and 10 kyr BP: **(a)** Alkenone-based SST (this study). **(b)** $\delta^{18}\text{O}$ of *N. pachyderma (sin)* (Dezileau et al., 2004). **(c)** Sea surface $p\text{CO}_2$ (this study). **(d)** Atmospheric CO₂ concentration in the Antarctic Taylor Dome ice core (composite data from Smith et al., 1999, and Indermühle et al., 2000). **(e)** $\delta^{13}\text{C}$ of *N. pachyderma (sin)* (Dezileau et al., 2004). **(f)** *CaCO₃ fluxes. **(g)** The relative abundance of *G. bulloides* (Mohtadi and Hebbeln, 2004). **(h)** *Iron fluxes (black dotted line) and the Fe/Al ratio (gray line). The Fe/Al ratio indicates marine vs. terrigenous sources of iron (Dezileau et al., 2004). **(i)** *Organic carbon fluxes. **(j)** *Biogenic Opal and *Total Nitrogen fluxes. Gray hatchured bar indicates the onset of increased stratification off 30° S (De Pol-Holz et al., 2006). Gray solid bar on left-hand side represents present-time values (Table 2S, data supplement: <http://www.clim-past-discuss.net/6/347/2010/cpd-6-347-2010-supplement.pdf>). *Thorium-230 normalized fluxes (see details in Dezileau et al., 2004).

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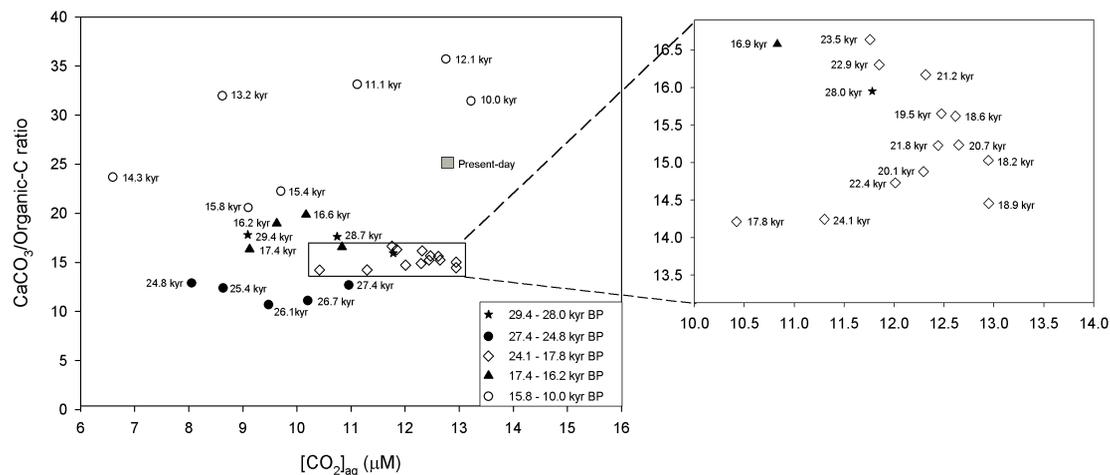


Fig. 2. Response of inorganic (CaCO_3) and organic Carbon (Organic-C) to the CO_2 concentration in sea surface waters between 10 and 30 kyr BP. Time intervals were delimited according to sea surface $p\text{CO}_2$ changes (Fig. 1c). Gray solid square represents the present-day relationship between the $\text{CaCO}_3/\text{organic-C}$ ratio and CO_2 concentration in sea surface waters at the study site (Table 2S: <http://www.clim-past-discuss.net/6/347/2010/cpd-6-347-2010-supplement.pdf>).

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